

**EXPLORING MITIGATION OPTION OF GREENHOUSE  
GAS (GHG) EMISSIONS UNDER WATER  
MANAGEMENTS AND SOIL AMELIORATIONS FROM  
MINERAL SOIL AND PEAT SOIL IN INDONESIA**

**February 2016**

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**Graduate School of Horticulture  
CHIBA UNIVERSITY**

(千葉大学学位申請論文)

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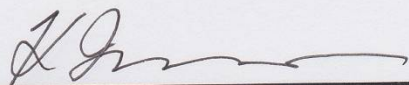
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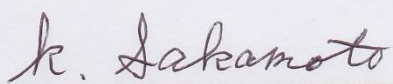
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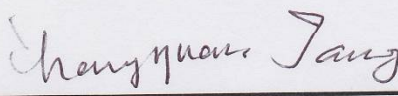
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## General Abstract

Nowadays, global warming is an important issue because it is one of the gravest threats to crop production and environmental sustainability. The agricultural sector is one of the sectors that contribute to global greenhouse gas (GHG) emissions. Rice (*Oryza sativa* L.) is one of the most important staple food crop for over half of the world's population and the demand is expected to increase due to the human population growth. Since irrigated areas are devoted to produce more rice, wastage of the resource, especially water in the rice field should be minimized. It is known that to produce 1 kg of rice in irrigated systems need approximately 4,000-5,000 litres of water. On the other hand, rice cultivation area is expected to reduce due to the conversion of agricultural land into other functions. There is increased competition for land, water, energy, and other inputs into food production. Consequently, the other way to meet increasing agricultural demands is looking toward the new arable land areas including peatland. Expansion of agricultural land is widely recognized as one of the most significant human alterations to the global environment.

In chapter 2, effect of different water treatments on methane (CH<sub>4</sub>) and nitrous oxide (N<sub>2</sub>O) production were studied under different paddy soils. Water treatments affected to CH<sub>4</sub> and N<sub>2</sub>O production potential. Continuous flooding (CF) treatment in incubation experiment produced more CH<sub>4</sub> than flooded-drained-flooded (FDF) in all soil types. However, FDF treatment could not reduce N<sub>2</sub>O production in all soil types. Somehow, reducing water from the soils stimulated N<sub>2</sub>O production. Five soil types showed different production potential of CH<sub>4</sub> and N<sub>2</sub>O which correlated with soil properties. Soil organic carbon (SOC) and ammonium (NH<sub>4</sub><sup>+</sup>) showed positive correlations with CH<sub>4</sub> production while total manganese (Mn), total iron (Fe) and nitrate (NO<sub>3</sub><sup>-</sup>) showed positive correlations with N<sub>2</sub>O production. On the other hand, total Fe and nitrate (NO<sub>3</sub><sup>-</sup>) have negative correlations with CH<sub>4</sub> production.

In chapter 3, effect of different water treatments, such as alternate wetting-drying (AWD), site specific AWD (S-AWD) and CF, on CH<sub>4</sub> and N<sub>2</sub>O emission were also studied in field experiment. Additionally, yield and water productivity from paddy field was also compared. The AWD and S-AWD are promising methods in irrigated rice cultivation with benefits on water saving, maintaining the productivity and reducing GHG emission comparable to CF irrigation. There were positive correlations between potential production of CH<sub>4</sub> – N<sub>2</sub>O and measured CH<sub>4</sub> – N<sub>2</sub>O emission ( $x$ ) from Pati's rice soil. It is important that comparative studies should be conducted in different environments to verify this practice as a way to conserve water under conditions of water scarcity while maintaining or increasing crop yields.

In chapter 4, effect of water table and soil amelioration on GHG emissions from peat soil were conducted in columns. Water depths changed the flux of carbon dioxide (CO<sub>2</sub>) somehow linearly, while it did not linearly change in CH<sub>4</sub> and N<sub>2</sub>O fluxes. There was a positive relationship between water depth and CO<sub>2</sub> emission. Less CO<sub>2</sub> was emitted lower when the water depth near the soil surface. Conversely, a deeper water depth resulted in a slight decrease in the CH<sub>4</sub> emissions. However, the highest N<sub>2</sub>O emissions were found at intermediate water depths. The biochar+compost and steel slag+compost treatments increased the CO<sub>2</sub> and N<sub>2</sub>O emissions from the peat soil columns. Long-term experiments should be developed to monitor changes that occur over time in response to amelioration at various water depths.

From the above research, it could be concluded that reducing water from rice field could save the water, maintain even increase yield and reduce CH<sub>4</sub> emission. However, it need to be cautious while recommending a particular irrigation regime for rice cultivation in order to avoid substantial emission of one or the other greenhouse gas. Although in this study, reducing water from field could reduce GHG emission. Long-term experiments should be developed for monitoring any changes over years in water depths and ameliorations effects, not only with peat soil but also with soil-crop systems in peat soil to determine how GHG

emissions from these type of treatments can be reduced. It might be better to apply ameliorants at higher rates to reach a sustainable reduction in GHG emissions but it should consider the applicability to be used by the farmer.

## General abstract (Japanese)

### インドネシアにおける無機質土壌と泥炭土壌からの 温室効果ガス放出削減に向けた水位管理と資材添加の検討

ヘレナ リナ スシラワティ

温室効果ガス放出による地球温暖化と農業への影響が懸念されており、主要な温室効果ガス放出源である水田からのメタンと一酸化二窒素の生成を抑制しつつ、重要な主食である水稲生産を増加できる技術が求められている。また人口増加に対して、水田への灌漑水供給や水田耕作面積は逼迫しつつある。本研究では、インドネシア、中央ジャワの主要水田土壌中の温室効果ガス生成への水分管理の影響を培養試験で明らかにし、またカリマンタンの泥炭土壌を用いたカラム試験で水位や資材添加がガス生成に及ぼす影響を解明し、さらに実際の水田で水管理のガス放出と水稲生育への影響を総合的に評価した。

第2章では、異なる水処理が5種類の水田土壌中のメタンと一酸化二窒素の生成能に及ぼす影響を培養実験で調べ、常時湛水が間断湛水に比べ多くのメタンを生成させたが、一酸化二窒素の生成は抑制できなかった。水分を低下させると一酸化二窒素の生成が促進された。5種類の土壌タイプの違いがメタンと一酸化二窒素の生成能の違いに影響しており、土壌の有機物含量や無機態窒素含量、鉄やマンガン含量と対応していた。

第3章では、圃場試験で水管理、すなわち常時湛水と間断灌漑(AWD)、地域特異的間断灌漑(Site specific AWD)の違いがメタンと一酸化二窒素の放出並びに水稲収量と水利用効率に及ぼす影響を検証した。間断灌漑と地域特異的間断灌漑は節水、水稲生産性、温室効果ガス削減の点で常時湛水より優れていた。Pati水田におけるメタンと一酸化二窒素の生成能と実測された両ガス放出量との間に正の相関関係が認められ、こうした節水管理が他の土壌タイプにおいても水稲収量の維持向上に貢献できることを示している。

第4章では、泥炭土壌の水位と土壌改良資材が温室効果ガス放出に及ぼす影響をカラム試験で検証した。地下水位と温室効果ガス放出量の間には相関関係があり、地下水位が高く地表付近の場合、CO<sub>2</sub>放出量は少なかった。対照的に地下水位が低いとメタン放出量が微減した。一方、一酸化二窒素放出量は中間的水位で最高に達した。生物炭+コンポスト、製鋼スラグ+コンポストの添加は泥炭土壌からのCO<sub>2</sub>とN<sub>2</sub>O放出を増加させた。作物生産性を考慮し、長期的な水位と資材施用の影響を更に研究されるべきである。

以上の研究より、水田での節水管理はメタン放出量の削減と水稲生産性の維持向上に繋がる可能性があるが、メタン以外の温室効果ガスの増加を防ぐような精密な水管理が求められる。またどの添加資材や添加量が泥炭土壌での温室効果ガス削減に有効かを、農民が利用可能かも含めて、長期試験で検討する必要がある。



## CHAPTER 1

### INTRODUCTION AND OBJECTIVES

#### 1.1 General introduction

Nowadays, greenhouse gas (GHG) emissions and their impact on climate are the important issues for agriculture's sustainability. World agriculture estimated to emit approximately 5.1–6.1 Peta-gram (Pg) CO<sub>2</sub> eq year<sup>-1</sup>, contributing 10–12% to the total global anthropogenic GHG emissions in 2005 (IPCC 2007a). Agriculture is a source for three primary GHG emissions: carbon dioxide (CO<sub>2</sub>), methane (CH<sub>4</sub>) and nitrous oxide (N<sub>2</sub>O) which contribute in global emissions approximately 60%, 15% and 5%, respectively (Watson *et al.* 1996). These gases long-live in the atmosphere and are the major contributors to positive increases in radioactive forces (IPCC 1996). Each of three trace gases have different global warming potentials (GWP) on a mass basis, which are 298 times higher for N<sub>2</sub>O and 25 times higher for CH<sub>4</sub> than CO<sub>2</sub> on a 100-year time scale (IPCC 2007b). Rice fields are important contributors of CH<sub>4</sub> and N<sub>2</sub>O emissions. Recently, emission of CH<sub>4</sub> from paddy fields was estimated for about 5-19% of the total CH<sub>4</sub> emissions, while fertilized agricultural soils was estimated for about 13-24% of annual global N<sub>2</sub>O emission (Mosier *et al.* 1998; Olivier *et al.* 1998; Kroeze *et al.* 1999; IPCC 2007a). Emission of CO<sub>2</sub> from agricultural sector are mainly due to land uses change (Verge' *et al.* 2006).

Rice (*Oryza sativa* L.) is one of the most important staple food crop for over half of the world's population (FAO 2013). The demand for rice production is expected to increase due to the human population growth. The global human population is currently growing, and estimates show that it will double by the middle of the next century. Over 90 percent of the world's rice is produced and consumed in the Asian region by 6 countries (China, India, Indonesia, Bangladesh, Vietnam and Japan) comprising 80% of the world's production and

consumption (Maclean *et al.* 2002). Indonesia is an agricultural country that had a rice-harvest area of approximately 13.8 million ha in 2014 (Statistics Indonesia 2014). In 1999, Indonesia was the third-largest rice producing country in the world and contributed at approximately 8% of total of world rice production (Coats 2003). One of the responses to increase food production from existing farmland is by intensification of cultivation. These production levels will affect the global environment because rice fields are a source of GHG emissions. If the intensification of rice cultivation is undertaken by using current practices and technologies to increase rice yields, then the GHG emissions from paddy fields will increase substantially if the cultivation technologies are employed without regard for the environment. Many cultivation practices have been improved in order to enhance yield potential and to decrease environmental burdens of paddy rice production. Using high-yielding crop varieties, fertilization, irrigation, and pesticides have contributed substantially to the tremendous increases in food production over the past 50 years (Matson *et al.* 1997).

Rice can be grown under irrigated or rainfed conditions. More than 75% of the rice supply comes from irrigated lowlands rice field (Tuong and Bouman 2003). For producing rice, a continuous flooding (CF) is used for the rice irrigation under the traditional system. In this technique, the paddy fields are inundated all the time starting from transplanting until nearly harvesting. Nowadays, the CF irrigation is getting difficult to be applied due to the decreasing water availability. According to Gleick (1993), the availability of water resources per capita in 2025 are expected to decline by 15–54% compared to 1990 in many Asian countries. There is a major challenge for rice cultivation to grow rice with less water and keep maintain or even increase the yield.

Since irrigated areas are devoted to produce more rice, wastage of the resource especially water in the rice field should be minimized (Oliver *et al.* 2008). One method to save water in irrigated rice field is allowed to dry intermittently of the rice fields instead of keeping

them continuously flooded but still adequately supplied with water for the roots of rice plant to maintain rice yields. That method is namely alternate wetting and drying (AWD). The principle of AWD is applied to flood the field in a certain number of days after the disappearance of ponded water. AWD combines the benefit aspect of aerobic and anaerobic rice cultivation. However, the reduced yields and pest control problems (primarily nematodes and weeds) associated with aerobic cultivation is one of the challenges to be addressed in AWD application (Kreye *et al.* 2009).

Recently, the area of rice cultivation is expected to reduce due to the conversion of agricultural land into residential, industrial area and other functions. There is increased competition for land, water, energy, and other inputs into food production. Consequently, the other way to meet increasing agricultural demands is looking toward the areas of arable land. Peatlands cover approximately 3.3% of the land surface area (Hadi *et al.* 2001). Tropical peatlands have been estimated to store up to 15-19% of the global peat carbon pool (Page *et al.* 2011). Approximately 14.9 million ha of peatlands are found in Indonesia and it was estimated 47% of the total tropical peatland areas (Ritung *et al.* 2011; Page *et al.* 2011). Large areas of tropical forest peatland in Indonesia have been converted to agricultural and non-agricultural sectors. Both natural and converted tropical peat soils are sources of CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O due to high levels of Carbon (C)-Nitrogen (N) content and hydrological conditions (Inubushi *et al.* 2003; Arai *et al.* 2014). To be used for agricultural activities, peat soils need to be drained as well as limed and fertilized due to excess water, low macro and micro-nutrient content, high organic acid content and high acidity (Lobb 1997). Expansion of agricultural land is widely recognized as one of the most significant human alterations to the global environment.

## **1.2 Objectives**

### **1.2.1. General objective**

This study was conducted with the general objective to identify the potential option of GHG mitigation in peat soil and mineral soil.

### **1.2.2. Specific objectives**

The specific objectives of this study were:

1. To investigate CH<sub>4</sub> and N<sub>2</sub>O production of five paddy soils in different water treatments under laboratory conditions.
2. To investigate the effects of alternate wetting and drying (AWD), site specific AWD (S-AWD) and continuous flooded (CF) on CH<sub>4</sub> and N<sub>2</sub>O emission, yield and water productivity from paddy field in Indonesia.
3. To determine the effect of different water depths and soil ameliorants on GHG emissions in peat soil columns.

## **1.3 Literature review**

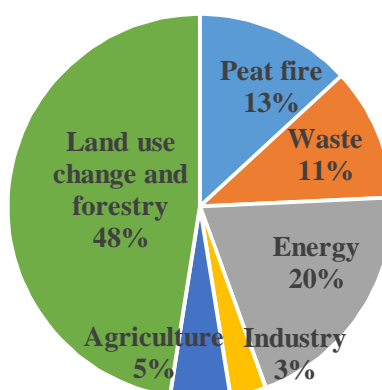
### **1.3.1 Global warming and greenhouse gas**

Nowadays, global warming is an important issue because global warming is one of the gravest threats to crop production and environmental sustainability. Global warming is one of the most prominent challenges in the present era. Global warming is caused by the increased concentration of greenhouse gas (GHG) in the atmosphere and leads to a phenomenon widely known as greenhouse effect. Greenhouse gas are those that absorb infrared radiation in the atmosphere, trapping heat and warming the surface of the Earth. The three GHG associated with agriculture are carbon dioxide (CO<sub>2</sub>), methane (CH<sub>4</sub>), and nitrous oxide (N<sub>2</sub>O). Atmospheric CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O had been recognized as the most important long-live GHG

that had significantly contributed to global warming potential (GWP) due to their great radiative forcing. Calculating their GWP therefore depends on the timeframe considered. For a 100-year timeframe, unit masses of CH<sub>4</sub> and N<sub>2</sub>O are considered to have 25 and 298 times the GWP, respectively, as a unit of CO<sub>2</sub> (IPCC 2007b).

### 1.3.1.1. Indonesian national GHG inventory

The National Greenhouse Gases Inventory of Indonesia (NGHGI) reported the calculations of GHG emissions in the second national communication (SNC) on the three main GHG emissions: CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O. There are six emissions categories defined by the Intergovernmental Panel on Climate Change (IPCC): energy, industrial processes, agriculture, land use and land use change and forestry (LULUCF), waste and peat fire-related emissions. In 2000, total Indonesia GHG emissions from six emissions without LULUCF (LUCF and peat fires) reached 556,499 Giga-gram (Gg) CO<sub>2</sub> eq. However, total Indonesia GHG emissions included LULUCF, increased significantly to about 1,205,753 Gg CO<sub>2</sub> eq (SNC 2010). The contribution of each gases on GHG emissions (in CO<sub>2</sub> eq) i.e., CO<sub>2</sub> emitted 940,879 Gg, representing 78% of the total; CH<sub>4</sub> emitted 236,388 Gg or 20% of the total; and N<sub>2</sub>O emitted 28,341 Gg or 2% of the total. The main contributing sectors were land use change and forestry (48%), followed by energy, peat fire-related emissions, waste, agriculture and industry approximately around 20, 13, 11, 5 and 3%, respectively (Figure 1.1).



**Figure 1.1.** National emissions contributions by sector in 2000 (SNC 2010)

### 1.3.1.2. GHG emissions from agricultural sector

Agriculture is a source for three primary GHG emission: CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O but it can also be a sink for CO<sub>2</sub> through C sequestration into biomass products and soil organic matter. The three main gases are influenced by land management and that are responsible for the potential greenhouse effect. Carbon dioxide, CH<sub>4</sub>, and N<sub>2</sub>O contribute at approximately 60%, 15% and 5%, respectively, in global emissions (Watson *et al.* 1996).

According to presidential regulation of Republic Indonesia no 61 year 2011, there are 2 targets of agricultural sector to reduce GHG emission in Indonesia until 2030, i.e., reduction of emission target was approximately 0.008 Giga tone CO<sub>2</sub> eq (26%) below business-as-usual (BAU) and approximately around 0.011 Giga tone CO<sub>2</sub> eq for conditional 41% reduction with sufficient international support. There are 3 strategies to achieve the targets of reduction emission, i.e., optimize land and water resources, apply land management and agricultural farming strategies that have lowest GHG emissions and absorb CO<sub>2</sub> optimally and stabilize the water level elevation and arrange for uninterrupted circulation of water in irrigation network (presidential regulation no 61 year 2011).

#### a. Carbon dioxide emission

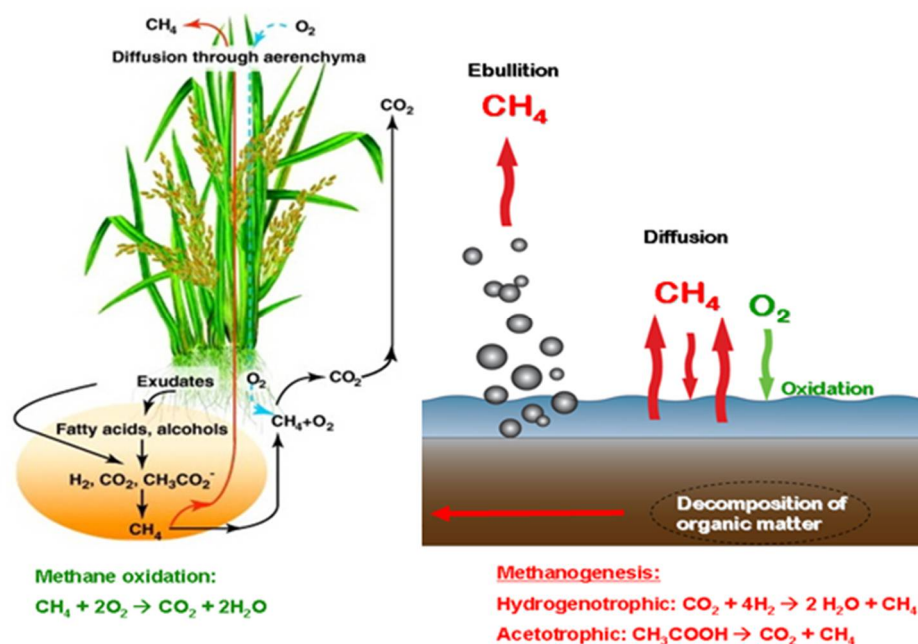
Carbon dioxide emissions from agricultural soils are mainly due to land use change, e.g., when forests are cleared for agricultural development. Soil cultivation and growing annual crops often accelerate the conversion of soil C to CO<sub>2</sub> by soil microbes. After the soils have been cultivated for a few decades, the loss of soil C usually slows down or ceases completely, and the level of soil C becomes stable again, but at a lower percentage (Hutchinson *et al.* 2007). Carbon dioxide, in comparison to CH<sub>4</sub> and N<sub>2</sub>O, is cycled in the largest amounts through agricultural cropping systems. Plants consume large amounts of CO<sub>2</sub> through photosynthesis to make food, feed, fibre, and fuel, but all these plant products eventually convert back to CO<sub>2</sub>

when consumed or when they decomposed. In agricultural sector, the net emission of CO<sub>2</sub> is small in comparison to its total cycling and is mostly due to energy use on-farm or in the manufacture and transport of agricultural products.

## **b. Methane emission**

Methane is a reactive and radioactively trace gas that has strong infrared absorption band characteristics, which is contributing to changes in atmospheric chemistry and may cause global warming (Bouwman 1991). The main contributors of CH<sub>4</sub> production from agricultural sector are rice paddies, ruminants, landfills, natural wetlands and sediments (Yang and Chang 1998). Figure 1.2 shows the mechanism of CH<sub>4</sub> production, consumption and transport from rhizosphere to atmosphere. In paddy fields, CH<sub>4</sub> emission is end product of the organic matter degradation and the result of complex interactions between rice plants and soil microbes under anaerobic conditions (Cicerone and Oremland 1988; Neue and Sass 1994; Conrad 1996; Conrad 2007). Anaerobic condition emit 80% of atmospheric CH<sub>4</sub> by methanogenic bacteria during digestion of organic matter in submerged soils (Ehhalt and Schmidt 1978). The rice plant provides methanogenic substrate through root exudates, decaying root tissues and to a lesser extent, litter fall from above ground parts and creates an active CH<sub>4</sub>-oxidizing site in the rhizosphere (Wassmann and Aulakh 2000). Root exudates of the rice plant consists of carbohydrates, organic acids, amino acids and phenolic compounds (Aulakh *et al.* 2001). Organic matter fermentation result acetate (CH<sub>3</sub>CO<sub>2</sub><sup>-</sup>), CO<sub>2</sub>, hydrogen (H<sub>2</sub>), propionate as well as other fatty acids which are the main substrates to produce CH<sub>4</sub> by methanogenic bacteria (Kruger *et al.* 2002; Conrad *et al.* 2010). Acetate is one of the main precursors of CH<sub>4</sub> production in rice soils through demethylation (Sigren *et al.* 1997) and can be derived either from root exudation (Lin and You 1989) or from fermentation (Neue and Roger 1993). Moreover, it provides methanogenic bacteria with plenty of precursors to produce CH<sub>4</sub> and

accelerates the decline of soil redox potential (Eh), thus forming a favourable environment condition for growth of methanogens, which in turn promotes formation of CH<sub>4</sub> and further affects CH<sub>4</sub> oxidation capacity in the field (Bender and Conrad, 1995; Arif *et al.* 1996). The requirements for soil reduction are the absence of oxygen, the presence of decomposable organic matter, and anaerobic bacterial activity. In the absence of oxygen, facultative and obligate anaerobes use nitrate (NO<sub>3</sub><sup>-</sup>), manganese (Mn(IV)), iron (Fe(III)), sulphate (SO<sub>4</sub><sup>2-</sup>) dissimilation products of organic matter, CO<sub>2</sub>, nitrogen (N<sub>2</sub>), and even hydrogen (H<sup>+</sup>) ions as electron acceptors in their respiration reducing NO<sub>3</sub><sup>2-</sup> to N<sub>2</sub>, Mn(IV) to Mn(II), Fe(III) to Fe(II), SO<sub>4</sub><sup>2-</sup> to hydrogen sulphide (H<sub>2</sub>S), CO<sub>2</sub> to CH<sub>4</sub>, N<sub>2</sub> to NH<sub>4</sub><sup>+</sup>, and H<sup>+</sup> to H<sub>2</sub> (Ponnamperuma 1972).



**Figure 1.2.** Methane production, consumption and transport in rice field (Philippot and Hallin 2011;

<https://www1.ethz.ch/ibp/research/environmentalmicrobiology/research/Wetlands>)

Under anaerobic condition, more than 90% transport CH<sub>4</sub> from rhizosphere into the atmosphere and the oxygen diffusion into roots is mediated by the aerenchyma and



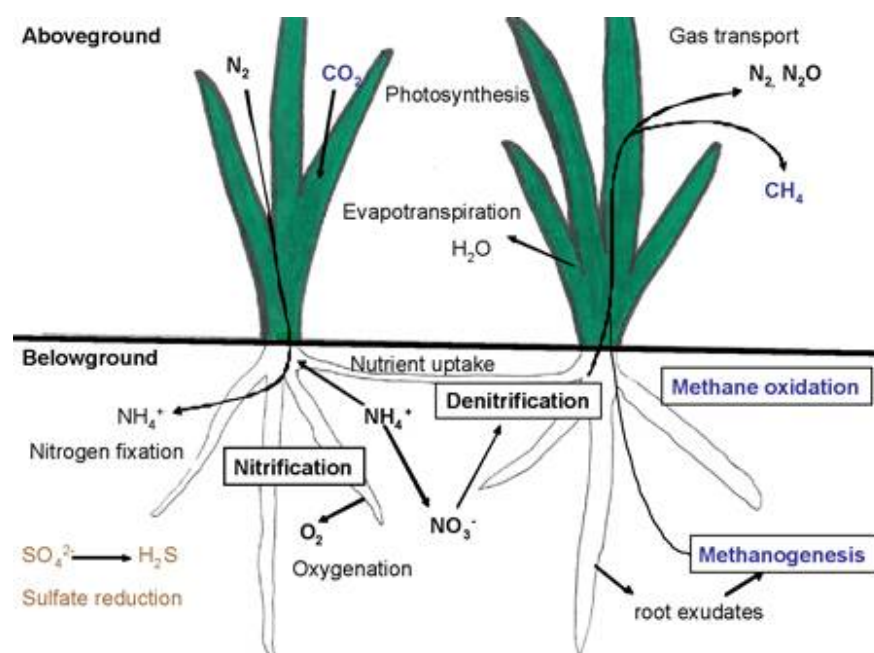
intercellular space system of rice plants in leaf blades, leaf sheaths, culm and roots (Raimbault *et al.* 1977; De Bont *et al.* 1978; Inubushi *et al.* 1989; Schutz *et al.* 1989) and the rest is released by the bubbles. Methane emission from rice field varies substantially among water management strategies. Globally, irrigated rice accounts for 70-80%, rainfed for 15% and deepwater rice for about 10% of the CH<sub>4</sub> produced from rice (Wassmann *et al.* 2000). Upland rice is not considered a significant source of CH<sub>4</sub> due to less water in the field (Neue 1997). Water management during the production of rice is a key factor in minimizing CH<sub>4</sub> during rice production. Draining the water and allowing the soil to become aerobic allows oxidation of CH<sub>4</sub> and reduces CH<sub>4</sub> production (US EPA 2006).

### **c. Nitrous oxide emission**

Nitrous oxide is important trace gas in the global N cycle. Increases in the atmospheric concentration of N<sub>2</sub>O contribute to global warming as well as directly to the destruction of the stratospheric ozone layer (IPCC 1996). Nitrous oxide emissions from paddy fields represent a substantial source of atmospheric N<sub>2</sub>O, although in small quantity compared with those from upland systems (Xing 1998). According to Yan *et al.* (2000), rice plants are an important pathway of N<sub>2</sub>O emission in the presence of flood-water. In the presence of floodwater, N<sub>2</sub>O emission is released around 87% through the rice plants. While in the absence of floodwater, N<sub>2</sub>O is emitted through the soil surface, with only 17.5% on average released through the plants.

The production of N<sub>2</sub>O in wetlands was shown in Figure 1.3. Nitrous oxide is by-products of microbial nitrification or intermediate products of denitrification processes (Mosier and Kroeze 2000). The N<sub>2</sub>O emission from soils by nitrification and denitrification processes depends on environmental and agricultural management factors, such as rain, temperature, fertilization, irrigation, and heavy metal accumulation, as well as on soil properties such as pH,

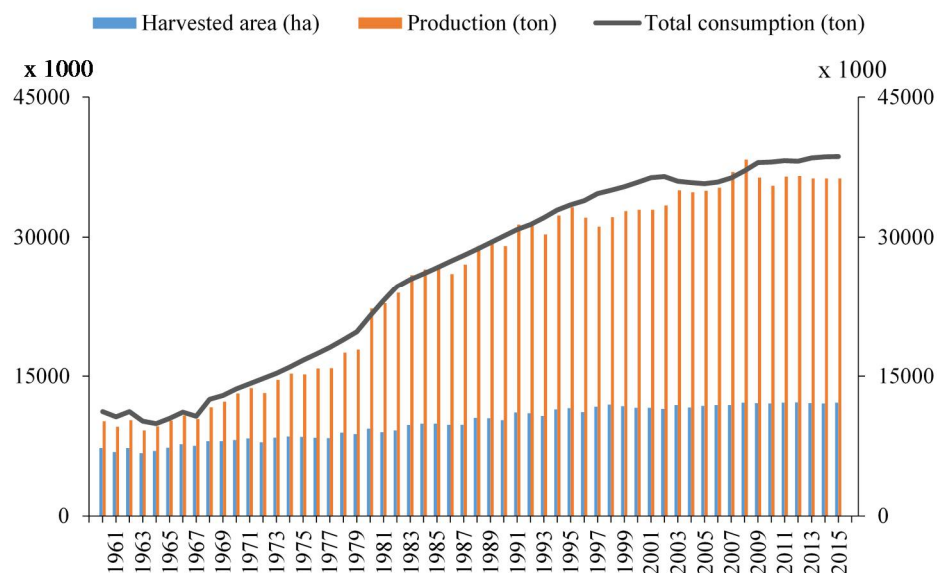
organic matter content, and particle size (Khalil *et al.* 2003). Nitrification is commonly defined as the biological oxidation of  $\text{NH}_4^+$  to  $\text{NO}_3^-$  with nitrite ( $\text{NO}_2^-$ ) as an intermediate (Bremner 1997). Ammonium produced by mineralization of soil organic matter (SOM) which is utilized by soil microorganisms and plants, this condition make the  $\text{NH}_4^+$  concentration in agricultural soils is generally low. However, the application of urea and  $\text{NH}_4^+$  make nitrification becomes the most active process in soils and quickly emit  $\text{N}_2\text{O}$  after N fertilizer is applied in the soil (Nishio and Fujimoto 1990; Cheng *et al.* 2002). The availability of oxygen ( $\text{O}_2$ ) in soil is one of the main factors regulating nitrification, denitrification and the release of  $\text{N}_2\text{O}$ . Denitrification occurs when  $\text{NO}_3^-$  is present in anaerobic microsites developed wherever microbial demand for  $\text{O}_2$  exceeds diffusion-mediated supply (Arah and Smith 1989). Denitrification in soils also consumes  $\text{N}_2\text{O}$  through the reduction of  $\text{N}_2\text{O}$  to  $\text{N}_2$ . Hence, this bacterial process may serve either as a source or as a sink of  $\text{N}_2\text{O}$ . Nitrification also contributes to  $\text{N}_2\text{O}$  emission due to fertilizer addition to soils during the oxidation of  $\text{NH}_4^+$  or hydroxylamine ( $\text{NH}_2\text{OH}$ ) to  $\text{NO}_3^-$  (Pathak 1999).



**Figure 1.3.**  $\text{N}_2\text{O}$  emission from wetland, <http://geology.usgs.gov/postdoc/profiles/moseman/>

### 1.3.2. Rice

Rice (*Oryza sativa* (L.)) is cereal food for feeding of more than half of the world's population, the most important food crop in many developing countries, and has also become a major crop in many developed countries where its consumption is predicted to increase by about 24% in the next 20 years (Van Nguyen and Ferrero 2006; Patel *et al.* 2010). In 2008, the rice consumption in Asia reached around 90% of the world rice consumption (IRRI 2009). Agricultural sector has a strategic position in Indonesia's economy. In 1999, Indonesia was the third-largest rice producing country in the world and contributed at approximately 8% of total of world rice production (Coats 2003). Although as third largest rice producing country, Indonesian rice production still not enough to fulfill its consumption. Figure 1.4 shows about Indonesian rice production, total consumption and harvested area, 1960-2015 (Ito 2015). Since 1960, the Indonesian consumption of rice exceeded the rice production likely occurred because Indonesia has very high human population and it increase very fast as well as limited area to change into rice field and dietary change.



**Figure 1.4.** Indonesian rice production, total consumption and harvested area, 1960-2015

(Ito 2015)

Rice is produced in a wide range of locations and under a variety of climatic conditions. The rice plant's environments have been divided into agro-ecological zones (AEZs): tropical regions, subtropical regions with summer or winter rainfall, and temperate regions (Maclean *et al.* 2002). Rice ecosystems can be defined irrigated, rainfed, lowland, upland, and flood-prone (Dobermann *et al.* 2004). Rice ecosystem in Indonesia was 54% irrigated rice area, 35% rainfed and 11% upland (Maclean *et al.* 2002). Irrigated rice production is the leading consumer of water in the agricultural sector, and rice is the world's most widely staple crop consumption, therefore finding ways to reduce the need for water to grow irrigated rice should benefit both producers and consumers. Most rice varieties maintain better growth and produce higher grain yields when grown in a flooded soil than when grown in dry soil (De Datta 1981). Recently, the concern about the sustainability of food production has increased because of the increasing of world population and the growing of environmental issues.

Rice growth duration is 3–6 months, depending on the variety and the environment under which it is grown. During this time, rice completes two distinct growth phases: vegetative and reproductive. The average rice yields in rice-growing countries range from less than 1 to more than 6 t/ha. There are many factors that influence the rice yield. Temperature, solar radiation, and rainfall influence rice yield by directly affecting the physiological processes involved in grain production, and indirectly through diseases and insects (Yoshida 1981). There are 16 essential elements and divided into major and minor elements. The major elements, carbon (C), hydrogen (H), oxygen (O), nitrogen (N), phosphorous (P), potassium (K), calcium (Ca), magnesium (Mg), and sulphur (S), are needed by plants in relatively higher amounts than the minor elements, Fe, Mn, copper (Cu), zinc (Zn), molybdenum (Mo), boron (B), and chlorine (Cl) (De Datta 1981). The nutrients can be made available to plant roots by contact exchange and soil solution. Based on rice disease-causing agents, the diseases of rice are classified into four groups i.e., fungus, bacteria, virus, and nematode (Ou 1979).

### 1.3.2.1. Water scarcity

Water is the primary resources for agriculture and food production. Water is a critical and the most important factor in rice production. Without water, no crops can be grown. About 75% of rice areas in the world are under continuous flooding (Van der Hoek *et al.* 2001). Rice grown under traditional practices in the Asian tropics and subtropics requires between 700-1500 mm of water per cropping season depending on soil texture (Bhuiyan 1992). It is known that in irrigated systems, approximately around 4,000-5,000 litres of water are used to produce 1 kg of rice in many areas (Tabbal *et al.* 1992). Presently, irrigation water efficiency is generally low. Efficiency for the flood irrigation practiced in paddy fields can be as low as 20% (Abdullah 2004). The competition related to the demand for water between agriculture and other sectors such as industry, environment, has become acute. Scarcity of water for agricultural production is becoming a major problem in many countries, particularly the world's leading rice-producing countries. Rainfall patterns in many areas are becoming more unreliable, with extremes of drought and flooding occurring at unexpected times due to climate change. Attempts to reduce water in rice production may result in yield reduction and may threaten food security in Asia. Reducing water input for rice will change the soil from submergence to greater aeration. These shifts may have profound – and largely unknown – effects on the sustainability of the lowland rice ecosystem.

### 1.3.2.2. Water management

Doorenbos and Kassam (1986) stated that water content below 70% of saturation cause yield decreases and at 50% could give a 50-70% yield reduction. At 30% of saturated water contents, zero yield is obtained, while at 20%, the rice plants will wilt and die. The challenge in paddy rice ecosystem to address problems of water scarcity, researchers had been looking for ways to develop the approaches that allow rice production to be maintained or

increased in the face of declining water availability. One such strategy to address this problem in rice field is the use of alternate wetting and drying (AWD) irrigation. Alternate wetting and drying is a water-saving technology in irrigated fields that applied alternately flooded and non-flooded condition in the field. The number of days of non-flooded soil in AWD between irrigations can vary from 1 day to more than 10 days, irrigation is applied after soil water potential has reached  $-10$  to  $-30$  kPa at 15 cm depth or shallow groundwater tables have gone to 10 to 40 cm depth, depending on growth stages of rice (Zhang *et al.* 2009). AWD can lower water use for irrigated rice by 35%, increase rice yield by 10% relative to continuous flooding (Yang *et al.* 2009; Zhang *et al.* 2009). However, other studies reported that AWD resulted slightly lowers yield (Sudhir-Yadav *et al.* 2012; Yao *et al.* 2012). One of the challenges on AWD adoption by the farmer is farmers do not always receive water at the time of demand, since irrigation is executed by a pump operator or other people who in charge on water management in that site (Kürschner *et al.* 2010).

### **1.3.2.3. Soil amendment/ ameliorant**

Soil amendment is the process of modifying soils to provide what the native or existing soils do not naturally provide. The amendment required can vary depending upon the existing soil and the traits of the soil that require alteration, for examples improving the drainage of a heavy clay soil, increasing the nutrient holding capacity of a highly sandy soil or repelling the negative effects of a saline soil near the coast with the application of calcium. Soil amendment mostly are conducted to enhance the nutrient status of the soil and to improve crop yield as well as to reduce GHG emission, e.g., biochar, manure, steel slag fertilizer (Table 1.1).

Steel slag, a by-product of the steel industry, contains of high iron, silica and calcium. Steel slag could enhance canopy photosynthesis, increased biotic and abiotic stress resistance, and contributions to healthy growth and high yields because it contain a high silica

content (Ma and Takahashi 2002). Some studies have shown that the application of steel slag fertilizer with high silica contents reduced CH<sub>4</sub> emissions because of aerenchyma enlargement (Furukawa and Inubushi 2004; Jackel *et al.* 2005; Ali *et al.* 2008). Steel slag could be used as an oxidizing agent to suppress CH<sub>4</sub> emissions from rice fields. Electron acceptors such as NO<sub>3</sub><sup>-</sup>, Mn<sub>4</sub><sup>+</sup>, Fe<sub>3</sub><sup>+</sup> and SO<sub>4</sub><sup>2-</sup> can decrease CH<sub>4</sub> production because of inhibitory and competitive effects with different microorganisms for common electron donors (Jakobsen *et al.* 1981; Achtnich *et al.* 1995). According to Ali *et al.* (2014), the application of silicate fertilizer with organic amendments decreased CH<sub>4</sub> flux, increased grain yield and improved soil quality. However, the application of steel slag as a soil ameliorant results in a trade-off between N<sub>2</sub>O and CH<sub>4</sub> emitted from paddy fields. Some studies have shown that applications of steel slag reduced CH<sub>4</sub> emissions (Furukawa and Inubushi 2004; Ali *et al.* 2008) and stimulated N<sub>2</sub>O emissions (Huang *et al.* 2009; Singla and Inubushi 2013). Based on Huang *et al.* (2009), using the same product could stimulate N<sub>2</sub>O emissions by enhancing the nitrification rate. Conversely, steel slag applications suppressed N<sub>2</sub>O emissions from rice fields (Susilawati *et al.* 2015).

Biochar application to soil is believed to improve soil fertility as well as sequester C to mitigate climate change (Lehmann *et al.* 2011). Biochar applications significantly decrease N<sub>2</sub>O emission and increase CH<sub>4</sub> emission from paddy fields in mineral soil (Zhang *et al.* 2012; Singla and Inubushi 2014). Existing reviews show conflicting results with respect to GHG emission following biochar application and the mechanisms remain debatable (Lehmann and Sohi 2008; Wardle *et al.* 2008; Mukherjee and Lal 2013). Another suggestion for improving soil fertility is the application of organic matter (Glaser *et al.* 2002). Manure apparently has a strong ability to mitigate soil acidity and enhance Ca uptake in a tropical forage legume (Hue 1992). However, organic matter can be mineralized very rapidly under tropical conditions (Tiessen *et al.* 1994).

**Table 1.1.** The emission of CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O from soil ameliorants in different land use in field

| Type of ameliorants                           | Application rate (Mg ha <sup>-1</sup> ) | Indicator plant | Country    | Soil types       | Range of   |                 |   | References           |
|---|---|-----------------|------------|------------------|--|-----------------|---|----------------------|
|   |   |                 |            |                  | CO <sub>2</sub><br>mg C m <sup>-2</sup> hour <sup>-1</sup> | CH <sub>4</sub> | N <sub>2</sub> O<br>μg N m <sup>-2</sup> hour <sup>-1</sup> |                      |
| Compost                                       | 12                                      | Paddy           | Japan      | Gley soil        |  | 5.07            |   | Yagi and Minami 1990 |
|   |   |                 |            | Andosols         |  | 2.53            |   |                      |
| Rice straw                                    | 6                                       | Paddy           |            | Gley soil        |  | 12.80           |   |                      |
|   |   |                 |            | Andosols         |  | 4.27            |   |                      |
|   |   |                 |            | Peat soil        |  | 21.73           |   |                      |
| Biochar (wheat straw)                         | 10-40                                   | Paddy           | China      | Entic Halpudept  | 54.4-65.9  | 1.02-1.82       | 83.33-129.17  | Zhang et al. 2012    |
| Green manure                                  | 20                                      | Paddy           | India      | Typic Ustochrept |  | 2.99            |   | Khosa et al. 2010    |
| Rice straw compost                            | 10                                      |                 |            |                  |  | 0.80            |   |                      |
| Wheat straw                                   | 10                                      |                 |            |                  |  | 7.08            |   |                      |
| Farmyard manure                               | 20                                      |                 |            |                  |  | 3.41            |   |                      |
| Wheat straw                                   | 3.75-4.8                                | Paddy           | China      | Typic Epiaquepts |  | 4.77-20.54      | 1.54-8.92   | Ma et al. 2009       |
| Steel slag                                    | 2-8                                     | Paddy           | China      | No information   |  | 1.59-2.29       | 12.14-28.54   | Wang et al. 2015     |
| Urea, rice straw compost                      | 0.22; 2, respectively                   | Paddy           | Bangladesh | Clay loam        |  | 3.90            |   | Ali et al. 2014      |
| Urea, rice straw compost, silicate fertilizer | 0.17; 2; 0.3, respectively              |                 |            |                  |  | 3.71            |   |                      |
| Urea, sesbania, silicate slag                 | 0.17; 2; 0.3, respectively              |                 |            |                  |  | 3.79            |   |                      |
| Urea, azolla anabaena, silicate slag          | 0.17; 2; 0.3, respectively              |                 |            |                  |  | 3.55            |   |                      |
| Urea, cattle manure compost, silicate slag    | 0.17; 2; 0.3, respectively              |                 |            |                  |  | 3.67            |   |                      |
| Manure  | 4                                       | Paddy           | Indonesia  | Peat soil        | 162.5  | 9.21            |   | ICCTF 2011           |
| Steel slag fertilizer (Pugam A)               | 0.75                                    |                 |            |                  | 163.6  | 9.39            |   |                      |
| Steel slag fertilizer (Pugam T)               | 0.75                                    |                 |            |                  | 218.2  | 8.52            |   |                      |
| Mineral soil                                  | 4                                       |                 |            |                  | 179.5  | 11.66           |   |                      |
| Biochar                                       | 20-40                                   | Maize           | China      | Inceptisols      | 55.4-62.2  | -0.03 to -1.88  | 22.00-47.73   | Zhang et al. 2015    |
| Coated fertilizer                             | 0.046                                   | Oil palm        | Indonesia  | Peat soil        | 153  |                 | 272   | Sakata et al. 2015   |



### 1.3.3. Peat soil

#### 1.3.3.1. Peat soil characteristic

Peat soils cover more than 420 million ha worldwide (Clymo 1987) which is around 3% of the earth's surface and contain around 10% of the total C stored in peatlands (Immirzi *et al.* 1992; Maltby and Proctor 1996). Around 36 million ha of it can be found in the tropics and subtropics (Andriessse 1988). Peat consists of dead, partially decomposed plant remains that that has accumulated over thousands of years in waterlogged environments that lack oxygen (Wösten *et al.* 2008). The plant material from which the peat is derived does have some influence on the chemical composition of the peat. Aboveground plant production is believed to be the primary source of peat (Clymo 1983). Tropical peatlands are predominantly forested with no moss cover (Rieley and Ahmad-Shah 1996). The high temperatures and more aerobic conditions on the surface of tropical peatlands may speed up the decomposition of leaves and wood, suggesting that roots could be more important to peat accumulation in the tropics. The accumulation rates range of tropical peatland approximately 4–5 mm year<sup>-1</sup> until 5–10 mm year<sup>-1</sup>, significantly faster than in most temperate and boreal peatlands because tropical peatlands occur in consistently hot and often humid conditions. However the rate of accumulation temperate and boreal peatlands approximately around 0.5–1 mm year<sup>-1</sup> (Gorham 1991, Maas 1996 and Gorham *et al.* 2003).

Important physical properties of the peat soils are bulk density, porosity, water holding capacity, subsidence, and irreversible shrinkage. Normally, higher degree of peat maturity (fibrists < hemists < saprists) will be followed by the high degree of bulk density (BD) and C organic content. Intensive management on peat land and environmental drainage could significantly influenced on BD and C density of peat. According to Wahyunto *et al.* (2010), the types of mineral materials substratum – non-sulfidic clay, sulfidic (marine) clay or sandy substratum– underlying the peat determine the peat fertility. The non-sulfidic clay are

commonly found in inland peats, and considered as the better substratum. However, the quartz sand substratum determine low to very low peat soil fertility. Moreover, the substratum peat soil in shallow peats, sulfidic and exposed to aerobic condition, there is a possibility of acid sulphate soils formation and a very poor peat for crop cultivation (Wahyunto *et al.* 2010). Peat is source of organic materials and particular it contains sizeable quantities of lignin, bitumens and humic acids (Delicato 1996). Tropical peatlands are widely distributed under waterlogged and acidified conditions. Tropical peatlands have a high porosity and, as a consequence, a high water-holding capacity that provides them with an important water regulation function with respect to downstream tropical lowlands.

#### **1.3.3.2. Peat soil and greenhouse gas emissions**

Peatlands play important roles in the global cycling of C as they are net sinks of atmospheric CO<sub>2</sub> and under natural decomposition release by-product such as N<sub>2</sub>O, CH<sub>4</sub> and CO<sub>2</sub> (Gorham 1991). The total C stored in tropical peatland is about 83.3 Gt where 44.5 Gt or about 53.1% is found in Indonesia across the three main islands, i.e., Sumatra, Kalimantan and Papua (West Papua) with total C stored of 18.3 Gt (41.1%), 15.1 Gt (33.8%) and 10.3 Gt (23%), respectively (Saharjo 2011). It has been estimated that peat and forest degradations contribute to about 45% of total GHG emissions from Indonesia (Ridlo 1997). This considerable contribution of peat on total GHG emissions has put peat soil as a target for C emission reduction. Peat management is targeted to reduce 9.5-13% of GHG emissions from Indonesia by year 2020 (Las and Surmaini 2010).

**Table 1.2.** The emission of CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O from different land use in Indonesia.

| Location  | Land-use  | Range (min-max) of   |                 |   | References                       |
|---|---|--|-----------------|---|----------------------------------|
|   |   | CO <sub>2</sub><br>mg C m <sup>-2</sup> hour <sup>-1</sup> | CH <sub>4</sub> | N <sub>2</sub> O<br>µg N m <sup>-2</sup> hour <sup>-1</sup> |                                  |
| South Kalimantan, Kalimantan  | Abandoned upland crops field, abandoned paddy fields, secondary forest  | 113 to 176   | 0.07 to 0.22    | -12.56 to -0.42   | Inubushi <i>et al.</i> 2003      |
| Jambi, Sumatera   | Drained forest, cassava field, upland paddy field, lowland paddy field  | 30 to 266  | 0.10 to 4.24    | 3.77 to 62.21   | Furukawa <i>et al.</i> 2005      |
| South Kalimantan, Kalimantan  | Secondary forest, paddy field, upland crops field, abandoned paddy field, abandoned upland, rice-soybean rotation field | 30 to 804  | -0.08 to 8.01   | -30 to 1040   | Hadi <i>et al.</i> 2005          |
| Riau, Sumatera<br>(1°30'N, 103°40'E)  | Sago palm   | 24 to 150  | -0.04 to 0.99   | nd  | Watanabe <i>et al.</i> 2009      |
| Central Kalimantan, Kalimantan<br>(02°21'S, 114°02'E and 2°19'S,<br>114°01'E )    | Natural forest, regenerated forest, burned forest, grassland cropland   | nd   | nd              | 5 to 2.957  | Takakai <i>et al.</i> 2010       |
| Central Kalimantan, Kalimantan<br>(2°20'27.74"S, 114°2'16.48"E)                   | Ferns, sedges, pulp wood  | 10 to 455  | 0 to 158.63     | -1.05 to 292.29   | Jauhiainen and Silvennoinen 2012 |
| South Kalimantan, Kalimantan  | Paddy, oil palm, vegetable  | -0.38 to 1.30  | 0.02 to 0.19    | -7.78 to 52.34  | Hadi <i>et al.</i> 2012          |
| Kalampangan, Kalimantan<br>(2°17'-2°21'S, 113°54-114°01'E)                        | Undrained natural forest, drained forest, burned forest, cropland   | 80 to 349  | -0.02 to 0.36   | 0.01 to 13.13   | Arai <i>et al.</i> 2014a, 2014b  |
| Central Kalimantan, Kalimantan<br>(02°12'26"S, 55°00"E)                           | Flooded forest, drained forest, flooded burnt site, drained burnt site  | 108 to 340   | 0.01 to 5.75    | -2.40 to 8.10   | Adji <i>et al.</i> 2014          |
| Tatau, Malaysia (02°57.924N,<br>112°45.851'E)                                     | Oil palm  | 90 to 223  | nd              | 131 to 523  | Sakata <i>et al.</i> 2015        |
| Serawak, Malaysia<br>(2°49'N, 111°51'E; 2°47'N, 111°50'E<br>and 2°49'N, 111°56'E) | Sago, oil palm, forest  | 46 to 533  | nd              | -3.4 to 176.3   | Melling <i>et al.</i> 2005, 2007 |

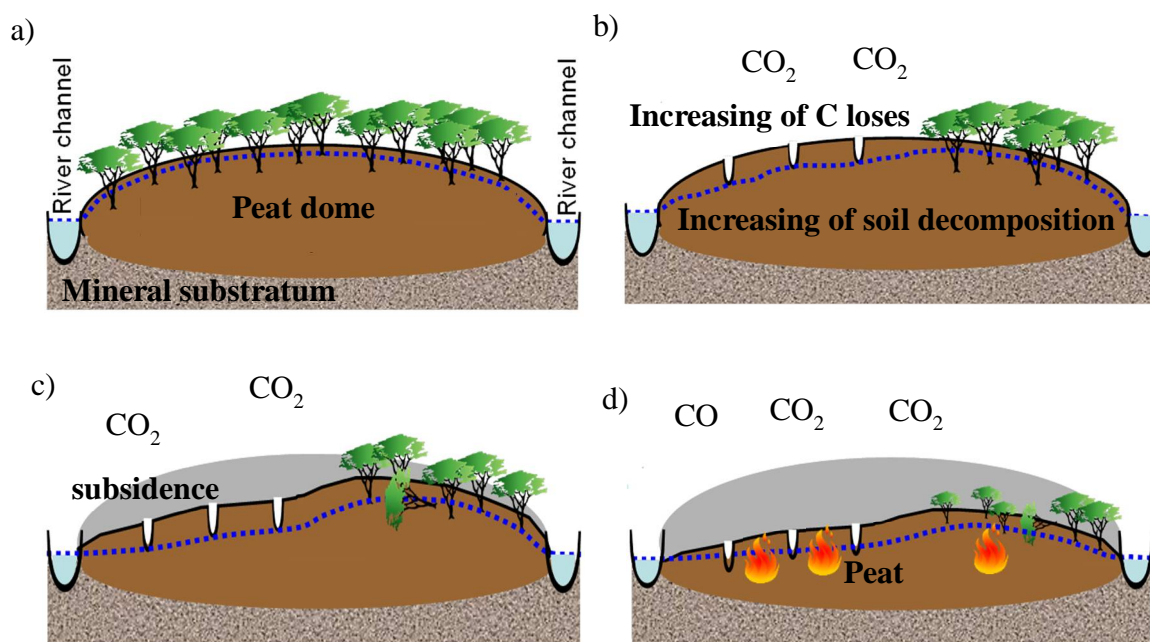
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Decomposition of drained peatlands in Indonesia is estimated to cause 632 Tera gram (Tg) year<sup>-1</sup> CO<sub>2</sub> emissions (range 355–874 Tg year<sup>-1</sup>), which will likely increase every year for the first decades after 2000 unless peatland use practices are changed (Hooijer *et al.* 2006). The cultivation of tropical peatlands has been considered to be a large source of N<sub>2</sub>O emissions and the IPCC (2000) collected the data from boreal and temperate regions and estimated that direct total N<sub>2</sub>O emissions from mineralization of soil organic nitrogen in cultivated organic soils approximately 16 kg N ha<sup>-1</sup> year<sup>-1</sup>. The CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O emission from different land uses in peat soil in Indonesia has been observed by many studies (Table 1.2)

#### **1.3.3.3. Peatland for agriculture**

Peat soil are fragile ecosystems with important biological and hydrological functions. Demand to expand agriculture are likely to lead to further deforestation. This conditions resulted the water adjustment and soil improvement that suitable for agriculture or for other land use. The process of drainage peatland is shown in Figure 1.5. Natural peatlands is a sequester carbon, waterlogged dome and mostly water tables lies near soil surface. Agricultural use of peatlands need to lowering of the water table, increased aeration, and changes in plant. When drainage starts as implication of land use change, it has led to a number of effects including increasing of soil decomposition, c losses, and CO<sub>2</sub> emission. Then, if drainage continued lead to peat subsidence and CO<sub>2</sub> emission, which is sign of carbon loss due to increased organic matter decomposition (Jauhainen and Silvennoinen *et al.* 2012). It has known that peat soil is irreversible soil, thus lowering water causes peat shrinkage, biological oxidation, loss of carbon stock and high risk of smouldering peat fire. Sometimes, agricultural use of peatlands often destroyed their ecological character and the ecosystem services that go with it. This is the consequences of using pristine peatland as agricultural functions. Figure 1.6

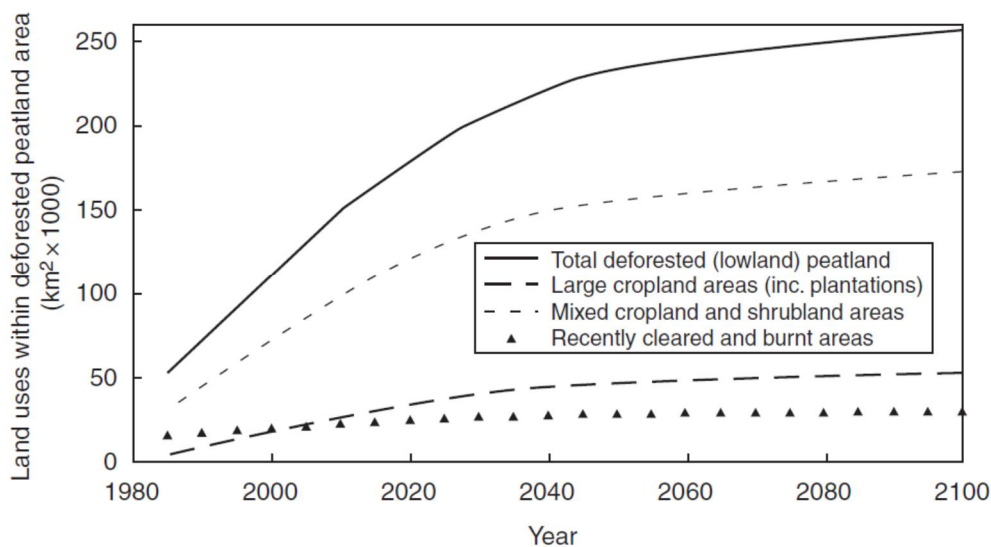
shows that large number of deforested peatland in South-East Asia and still remaining peatlands can be expected to be logged and drained in the next few decades. According to Hooijer *et al.* (2006) that the area used for oil palm plantations, currently estimated at 20 000 km<sup>2</sup>, is expected to more than double to a surface area of 50 000 km<sup>2</sup> by 2050.



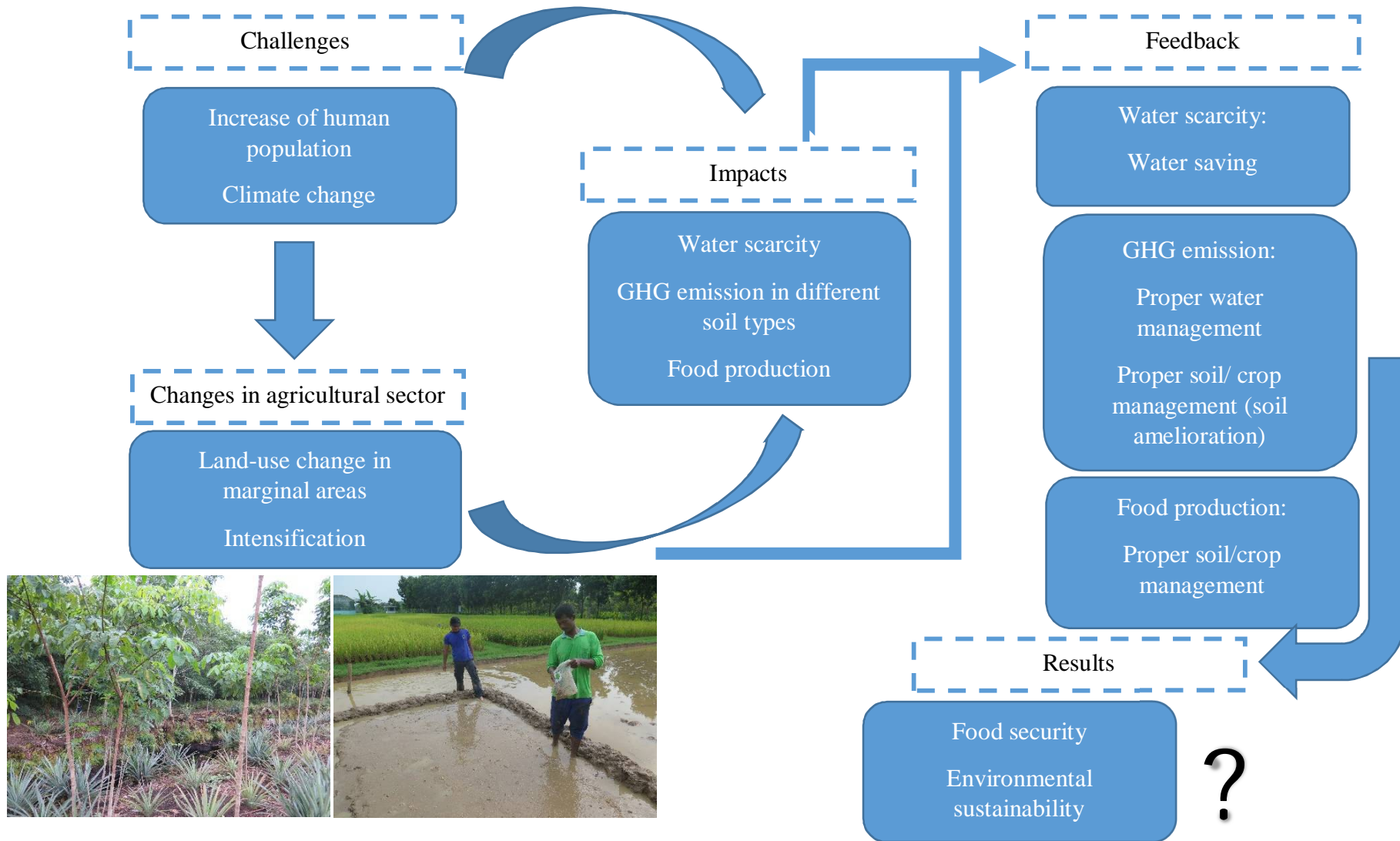
**Figure 1.5.** The process of drainage tropical peatland, a) natural peatland; b) when drainage starts; c) and d) when drainage continue (source: modify from <http://blogs.helsinki.fi/jyjauhia/peat-in-agriculture-and-forestry>)

Several management strategies such as soil conservation practices, incorporation of crop residues, use of composts, minimum tillage and others soil-crop managements hold promise for achieving GHG mitigation and adaptation (Hobbs 2007; Delgado *et al.* 2011). Murdiyarso *et al.* (2010) reported that intensifying existing production of oil palm and locating new plantations in degraded secondary forests and grasslands in peat swamp forest in Indonesia can both satisfy demands and reduce greenhouse gas emissions. Thornton and Herrero (2010) observed that extensive tropical livestock systems and conclude that management options could

mitigate a maximum of ~7% (417 Mt CO<sub>2</sub> eq or 0.417 Pg CO<sub>2</sub> eq) of the global agricultural mitigation potential to 2030 without reducing production. Management options to enhance food production commonly involve trade-offs among multiple objectives. Synergistic options to meet multiple objectives although less common, when and where they do exist should be done to achieve not only food security and sustainability but also environmental friendly.



**Figure 1.6.** Current trends and future projections of land use within deforested peatlands in South-East Asia (Verhoeven and Setter 2009).



**Figure 1.7.** The framework study of water managements and soil ameliorations in mineral soil and peat soil

## Chapter 2

### METHANE AND NITROUS OXIDE PRODUCTION CAPACITIES FROM DIFFERENT RICE SOILS UNDER DIFFERENT WATER TREATMENTS

#### 2.1. Introduction

Methane and N<sub>2</sub>O are considered the major sources of GHG, emitted mainly from flooded rice fields because the coexistence of aerobic and anaerobic condition (Reddy *et al.* 1989). Methane production in soils generally occurs under strictly anaerobic condition. On the other hand, N<sub>2</sub>O is produced from nitrification under aerobic conditions, and denitrification under moderately anaerobic conditions. Generally, there is a trade-off between CH<sub>4</sub> and N<sub>2</sub>O production in rice soils and this condition makes more challenge on reducing the production of one gas but not to increase the production of the other.

The process of wetting and drying of the soil change soil structure therefore likely to affect the biological processes that lead to C and N transformations and the biogenic gases production (Beare *et al.* 2009). Flooding and unflooding field influence root activity, photosynthesis and respiration of rice plants. In a soil profile, CO<sub>2</sub> is produced by respirations of plant root and microorganisms. The CO<sub>2</sub> is partly leached (Minamikawa *et al.* 2010). However, many of the data obtained so far are not sufficiently detailed to examine CO<sub>2</sub> exchange in rice paddies (Liu *et al.* 2013).

There are important factors in soil that control CH<sub>4</sub> and N<sub>2</sub>O production from rice fields. Methanogenesis and N<sub>2</sub>O production are influenced by physical and biochemical factors in the soil, such as soil pH, redox potential, organic matter content, temperature, and soil moisture content. The content of soil oxidants (O<sub>2</sub>, NO<sub>3</sub><sup>-</sup>, Mn<sup>4+</sup>, Fe<sup>3+</sup>, SO<sub>4</sub><sup>2-</sup> and CO<sub>2</sub>) used as electron acceptors for organic matter degradation contributes significantly to these processes (Yu *et al.* 2001). The reduction of various oxidants in homogeneous soil suspensions occurs



sequentially at corresponding soil redox potential values (Ponnamporuma 1972). A better understanding of this relationship is needed in order to be able to possibly mitigate the emission of these important GHGs through changes in agricultural practices.

The objective of this study was to investigate CH<sub>4</sub> and N<sub>2</sub>O production of five paddy soils of Indonesia in different water treatments under laboratory conditions.

## **2.2. Materials and Methods**

### **2.2.1. Soil samples collection**

Soil samples were collected from irrigated rice areas in Central Java, Indonesia (Figure 2.1). The soil samples were taken from surface layer (0 to 15 cm) at all sites. The samples were collected in March 2014. Soil maps with a scale of 1:250,000 from Indonesian Center for Agricultural Land Resources Research and Development (ICALRRD) were used to determine the soil sampling sites. Five sites of paddy fields, (i) Klaten, (ii) Boyolali; (iii) Grobogan; (iv) Demak and (v) Pati district, were classified as typic dystrudepts, typic hapludants, typic epiaquepts, typic epiaquepts and aeric endoaquepts, respectively. The soil samples were brought to Indonesian Agricultural Research Institute (IAERI), Jakenan, Indonesia. Characteristics of soil used are listed in Table 2.1.



**Figure 2.1** Map of Java Island. Red dot colour is the location of soil sampling.

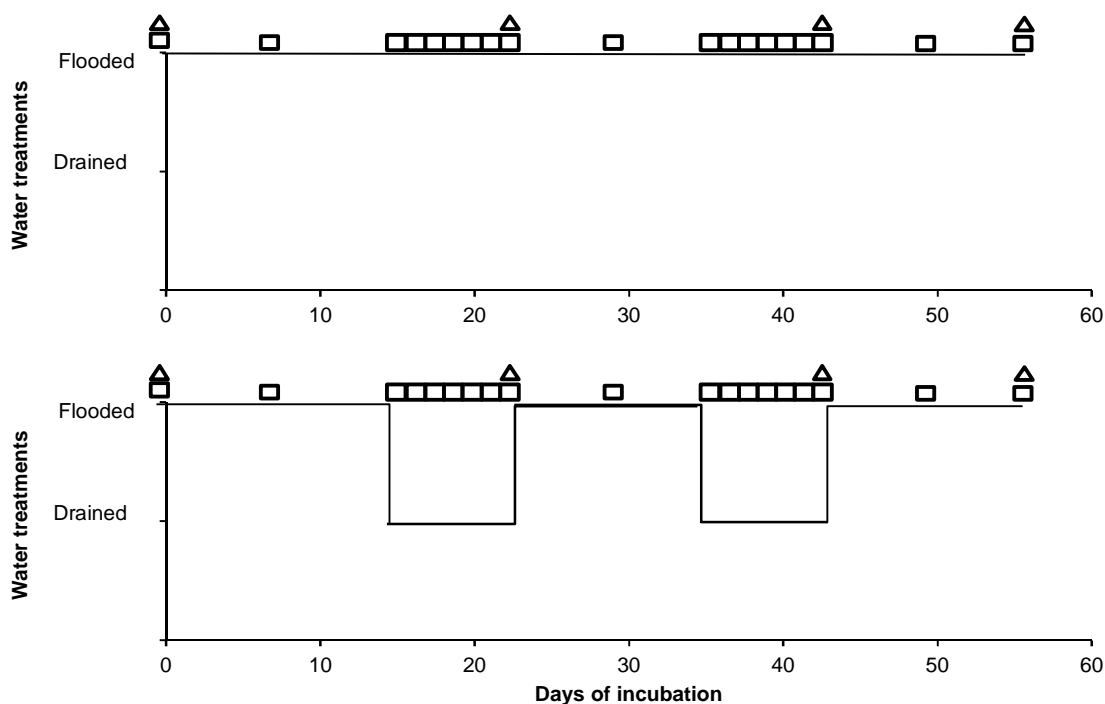
### 2.2.2. Soil incubation

Soil samples were air-dried, and ground to pass through a 2-mm stainless steel sieve with removal of visible plant residues. Each of the soils was taken in a glass bottle and pre-incubated for 7 days. Pre-incubation of the soil samples were carried out in the dark at 30°C and in aerobic condition. After pre-incubation finished, 20 grams of soil sample were put into a 120-ml beaker and 40 ml of distilled water was added to keep the soil saturated. Magnetic bar beakers bottles were closed tightly with butyl rubber stoppers and silicone grease sealant. The stoppers had 4 holes fitted with stoppered glass tubing to facilitate flushing of beaker headspace with N<sub>2</sub>, collection of gas samples through septa via syringes and monitoring of pH-Eh by inserting suitable electrodes. Headspace was flushed with N<sub>2</sub> at a rate of 300 ml min<sup>-1</sup> for at least 10 min to stimulate soil reduction. The beakers were kept in incubator and maintained at temperature of 30°C for 57 days of incubation (Mitra *et al.* 2002). Two sets of bottles were used for gas sampling: one set was used as continuous flooded (CF) and the other was used as flooded-drained-flooded (FDF). All measurements were carried out in triplicate.

During the drying phase, water were removed from the FDF incubation bottles for 7 days (Figure 2.2). Gas samples were collected every day for both treatments. After 7 days, soil incubations were rewetted and gas samples were collected once a week. The NO<sub>3</sub>, NH<sub>4</sub>, Fe total and Mn total analyses were repeated once a week on the remaining soil incubation at the termination of the experiment.

**Table 2.1.** Characteristics of the soils used

| <b>Parameters</b>                      | <b>Klaten</b>             | <b>Boyolali</b>           | <b>Grobogan</b>           | <b>Demak</b>              | <b>Pati</b>               |
|--|---------------------------|---------------------------|---------------------------|---------------------------|---------------------------|
| Soil order                             | Typic dystrodepts         | Typic hapludants          | Typic epiaquepts          | Typic epiaquepts          | Aeric endoaquepts         |
| Latitude                               | 07 <sup>0</sup> 36'23.4"  | 07 <sup>0</sup> 26'39.1"  | 07 <sup>0</sup> 03'13.0"  | 06 <sup>0</sup> 52'10.6"  | 06 <sup>0</sup> 46'42.1"  |
| Longitude                              | 110 <sup>0</sup> 42'24.3" | 110 <sup>0</sup> 41'41.9" | 110 <sup>0</sup> 46'17.0" | 110 <sup>0</sup> 42'58.6" | 111 <sup>0</sup> 11'52.0" |
| Sand (%)                               | 29                        | 15                        | 11                        | 15                        | 43                        |
| Silt (%)                               | 34                        | 19                        | 24                        | 20                        | 40                        |
| Clay (%)                               | 37                        | 67                        | 65                        | 65                        | 17                        |
| pH (H <sub>2</sub> O)                  | 6.4                       | 7.8                       | 6.8                       | 7.4                       | 5.6                       |
| CEC (cmol kg <sup>-1</sup> )           | 11.38                     | 35.66                     | 30.64                     | 20.68                     | 7.93                      |
| Active Fe (%)                          | 1.25                      | 0.56                      | 1.05                      | 0.4                       | 0.2                       |
| Available P (mg kg <sup>-1</sup> soil) | 27.1                      | 10.0                      | 22.7                      | 28.8                      | 4.8                       |
| Total N (g kg <sup>-1</sup> soil)      | 1.1                       | 1.1                       | 0.7                       | 0.7                       | 0.2                       |
| Total C (g kg <sup>-1</sup> soil)      | 10.8                      | 12.5                      | 8.2                       | 2.0                       | 1.6                       |



**Figure 2.2.** Schematic diagram of water treatments during incubation experiment

### 2.2.3. Gas sampling

Gas samples were taken and analysed periodically at 1, 8, 15, 16, 17, 19, 22, 30, 36, 37, 38, 40, 50 and 57 days of incubation (DOI). During time 0 ( $C_0$ ), the beaker were stirred and flushed with  $N_2$  for 2 minute before gas sampling to flush out accumulated gas from the beaker headspace, then the beaker were closed and the gas samples in the headspace of the beaker were taken using a syringe. After 24 hours ( $C_{24}$ ), the beakers were stirred again and gas samples were taken by syringe from the headspace of the beaker headspace (Wang *et al.* 1999). Ten millilitres of gas in the syringe was then injected into an auto-sampler vial. The concentrations of  $CH_4$  and  $N_2O$  in the vial were simultaneously analysed with a gas chromatograph equipped with flame ionization detector (FID) and electron capture detector (ECD). The incubation lasted 57 days. The Eh in the slurry was monitored and measured with a pH-Eh meter connected to a platinum electrode.

Production potential of CH<sub>4</sub> or N<sub>2</sub>O were calculated based on the equation from Lantin *et al.* (1995) as follows:

$$E = (C_{24} - C_0) \times \frac{V_h}{20 \text{ g}} \times \frac{mW}{mV} \times \frac{273}{(273 + T)}$$

E : CH<sub>4</sub> or N<sub>2</sub>O production (mg CH<sub>4</sub> or N<sub>2</sub>O g<sup>-1</sup>soil)

C<sub>0</sub> : CH<sub>4</sub> or N<sub>2</sub>O concentration in time 0 (ppm)

C<sub>24</sub> : CH<sub>4</sub> or N<sub>2</sub>O concentration after 24 hours (ppm)

V<sub>h</sub> : Volume of headspace of beaker (ml)

mW : Molecular weight of CH<sub>4</sub> or N<sub>2</sub>O (g)

mV : Molecular volume of CH<sub>4</sub> or N<sub>2</sub>O (22.41 litre at *stp, standard temperature and pressure*)

T : Temperature of incubator (°C)

The total CH<sub>4</sub> and N<sub>2</sub>O production potential were calculated as follows:

$$\text{Total CH}_4 \text{ or N}_2\text{O production} = \sum_i^n (R_i \times D_i)$$

Where R<sub>i</sub> is the production potential rate of CH<sub>4</sub> or N<sub>2</sub>O (mg g<sup>-1</sup> soil hour<sup>-1</sup>) in the *i*<sup>th</sup> sampling interval, D<sub>i</sub> is the number of days in the *i*<sup>th</sup> sampling interval and n is the number of sampling intervals.

#### 2.2.4. Statistical analyses

The data were analysed using SAS software (SAS Institute 2003). The mean comparison between treatments was established by Tukey HSD test. Simple and multiple correlation analysis between total production of CH<sub>4</sub>-N<sub>2</sub>O and soil parameters was established to find out the effect of water treatments on production CH<sub>4</sub>-N<sub>2</sub>O from different soil types.

## 2.3. Results and discussion

### 2.3.1. Methane production

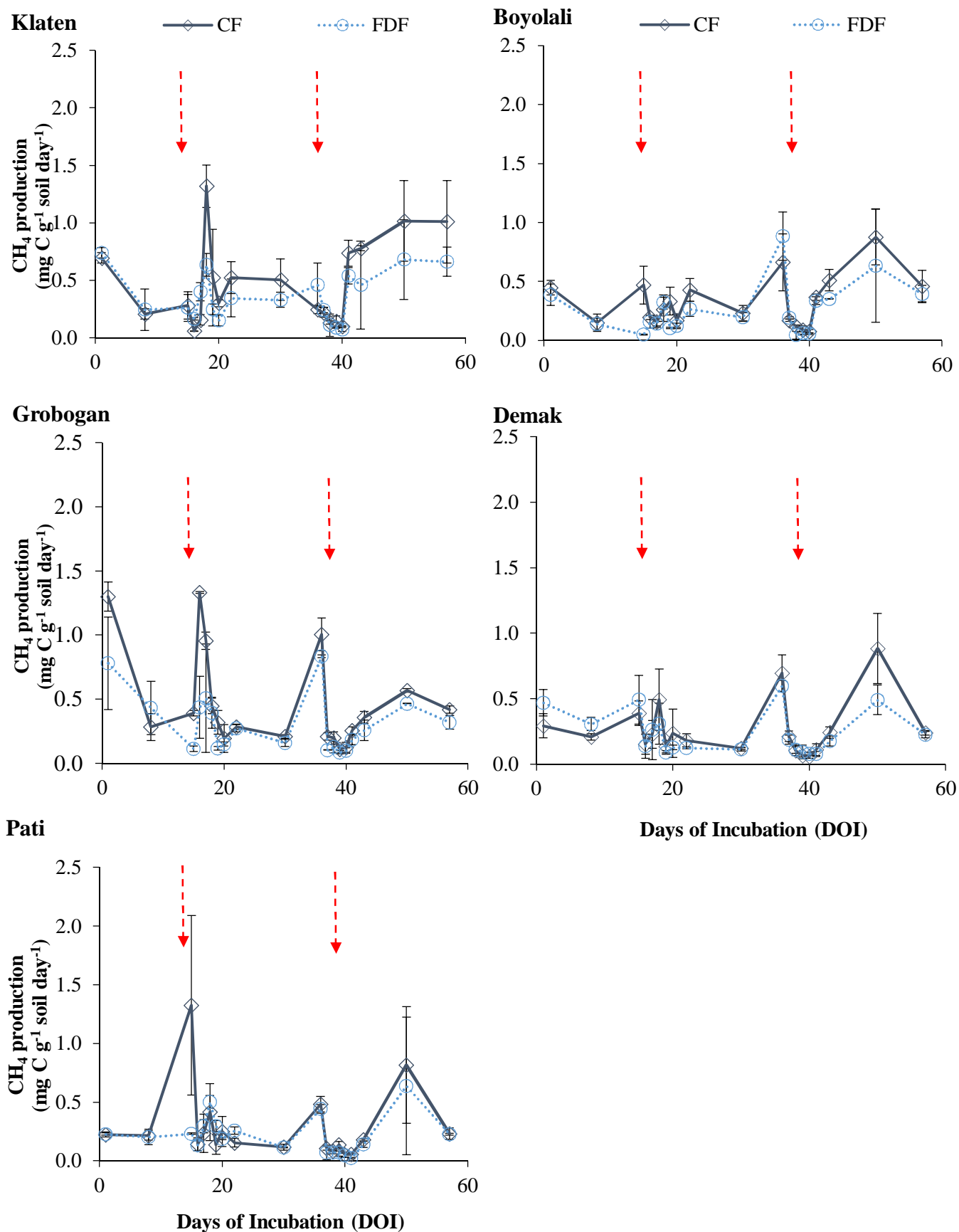
The time course of CH<sub>4</sub> production differed between the soils (Figure 2.3). Early in the experiment, the CH<sub>4</sub> productions were high in all of the treatments, likely occurs because of all soils produced CH<sub>4</sub> immediately after submergence (indicated by a flush). The production potential of CH<sub>4</sub> slightly decreased in second measurement. The drainage in FDF treatment was conducted twice. Drainage was started 15 and 36 Days of Incubation (DOI), but then the soils were re-flooded at 20 and 41 DOI. Watanabe *et al.* (2010) reported that wetting and drying of the soil could change the composition, population and transcriptional activities of the methanogenic archaea. Before drainage *Rice cluster I*, *Methanomicrobiales* and *Methanosarcinales* exist, but after drainage only the *Methanomicrobiales* were detected (Sugano *et al.* 2005). It means that the CH<sub>4</sub> is still produced in flooded or in drained condition, the difference is only amount of CH<sub>4</sub> production. During the measurement period, there were different peaks of CH<sub>4</sub> production in each of the soil likely occurs because of the existence of methanogenic bacteria. According to Roy *et al.* (1997), the reason for different initiation time of CH<sub>4</sub> production in different soils likely occurred because of the difference in abundance of viable methanogenic archaea in the air-dried. Unfortunately, in this study did not observe the activity of microorganism.

In all cases, the CF treatment had the highest CH<sub>4</sub> production compared to the FDF treatment. There were huge amounts of CH<sub>4</sub> degassing from FDF treatment when the water was drained, therefore CH<sub>4</sub> production from FDF were comparably low. Somehow, the trend of CH<sub>4</sub> production between CF and FDF treatments almost similar probably due to effect stirring of the soil. The ratio between soil and water should be carefully adjusted based on the soil texture. More clay content in the soil reduced the velocity of the magnetic bar inside of the soil. The velocity of magnetic bar is important aspect in this incubation experiment because

stirring can release the gas that entrapped in micro-pore of the soil and purify the gas in the micro-pore of the soil with wash out the soil using N<sub>2</sub>. Based on the procedure for gas sample collection are described by Mitra *et al.* (2002) that incubation experiments were conducted by placing a magnetic bar inside the incubation bottle to stir the soil during gas sampling to release gas which entrapped between the micro-pore of the soil. Besides release gas, soil stirring also increase the chance for oxygen diffusion into soil. The oxygen concentration influence the production of CH<sub>4</sub> in both treatments. So, although in CF or FDF, both of the treatments could release the gases which entrapped between the soils. The CH<sub>4</sub> production in the CF treatment from Klaten, Boyolali, Grobogan, Demak and Pati ranged approximately 0.06-1.32; 0.08-0.88; 0.11-1.30; 0.06-0.88 and 0.05-1.33 mg C g<sup>-1</sup> soil, respectively. However, the CH<sub>4</sub> production in the FDF treatment from Klaten, Boyolali, Grobogan, Demak and Pati ranged approximately 0.07-0.74; 0.04-0.88; 0.09-0.78; 0.08-0.60 and 0.02-0.64 mg C g<sup>-1</sup> soil day<sup>-1</sup>, respectively.

Figure 2.4 shows the cumulative CH<sub>4</sub> production in each of the soils. This figures show that the first water drainage resulted the lower CH<sub>4</sub> production in all soil. Compare to CF treatment, the FDF treatment in Klaten, Boyolali, Grobogan, Demak and Pati's soils reduced the cumulative of CH<sub>4</sub> production by approximately 23.7; 21.6; 34.7; 11.4 and 21.9%, respectively. This result is similar to the finding in paddy field studies reported by Cai *et al.* (2003); Kang *et al.* (2002); Zhang *et al.* (2011), i.e., that CH<sub>4</sub> fluxes were higher during flooded condition rather than drained during the fallow condition. High CH<sub>4</sub> production in CF may have caused by soil water conditions. It is one of the factors that control CH<sub>4</sub> production in paddy soil, due to methanogenesis takes place under strict anaerobic reducing conditions. FDF treatments improved soil aeration and facilitated O<sub>2</sub> diffusion from the atmosphere into the soil. This condition inhibit the formation of CH<sub>4</sub> due to O<sub>2</sub> availability is the major factor limiting methanotrophy bacteria. King *et al.* (1990) proved the importance of O<sub>2</sub> availability in Florida swamps where gas diffusion is easy, therefore methanotrophic bacteria was significant in peat.





**Figure 2.3.** The time course of CH<sub>4</sub> production rate during soil incubation. Red arrow means

drainage the water in FDF treatment

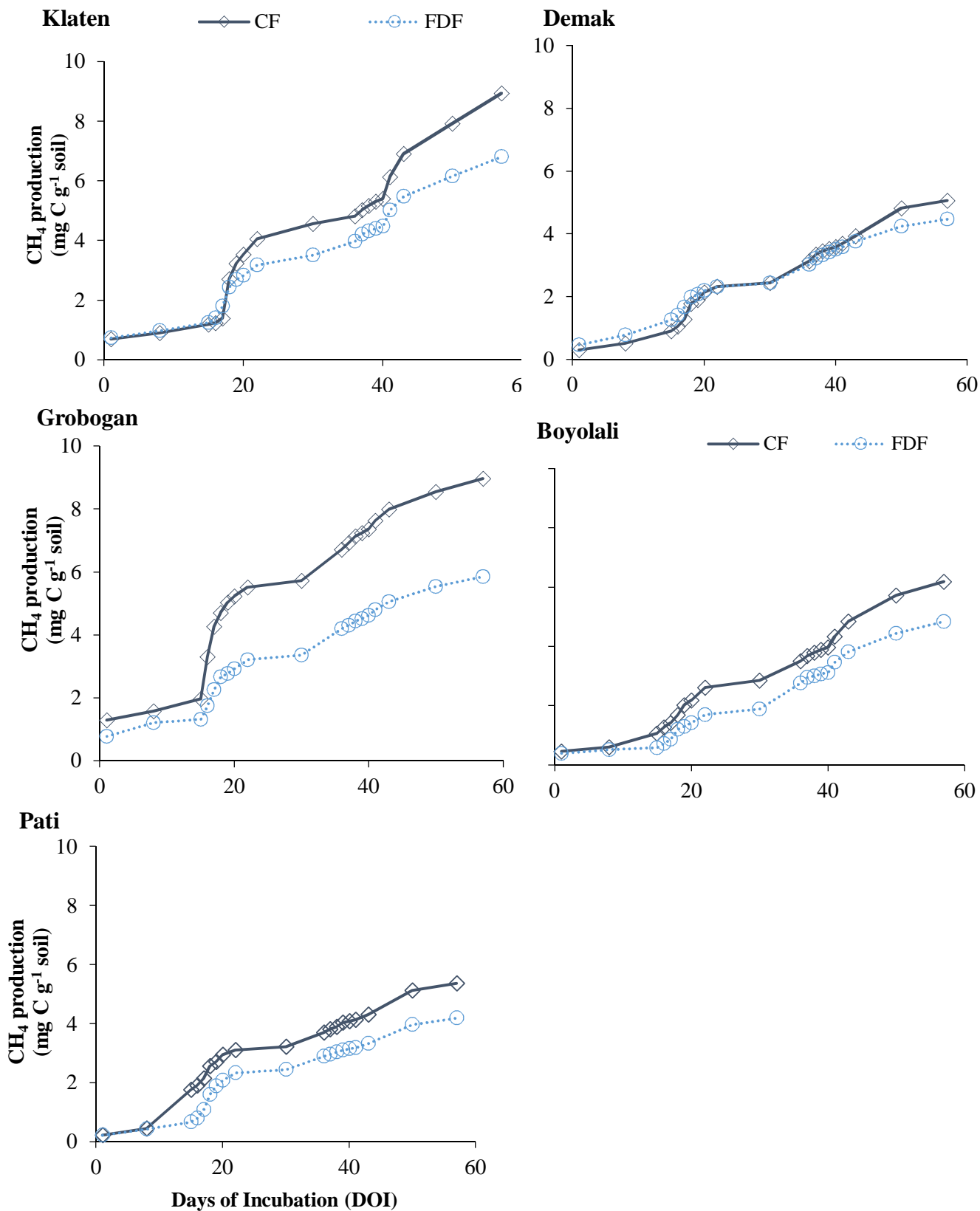
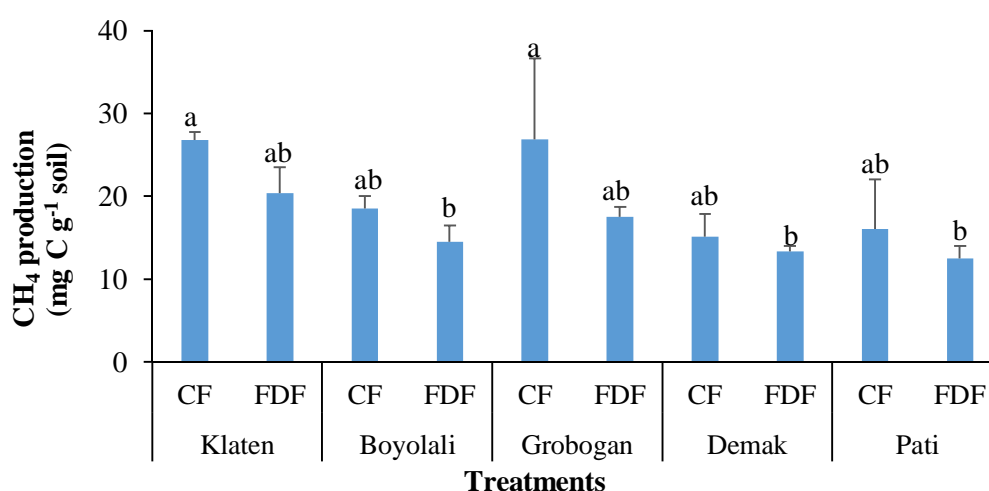


Figure 2.4. The pattern of cumulative CH<sub>4</sub> production during soil incubation

Although after re-flooded, the CH<sub>4</sub> production from FDF treatment almost in all soils were lower than CF treatment. Li *et al.* (2011) reported that drainage in longer period reduces oxidants in the soil, therefore soil redox (Eh) drop to favourable level for CH<sub>4</sub> production and it is difficult for methanogenesis to survive and CH<sub>4</sub> emission becomes lowered. The duration of the reduction processes varied greatly between the soils. According to Yao *et al.* 1999, NO<sub>3</sub><sup>-</sup> was reduced first, then reduction of Fe(III) and SO<sub>4</sub><sup>2-</sup> were completed later.



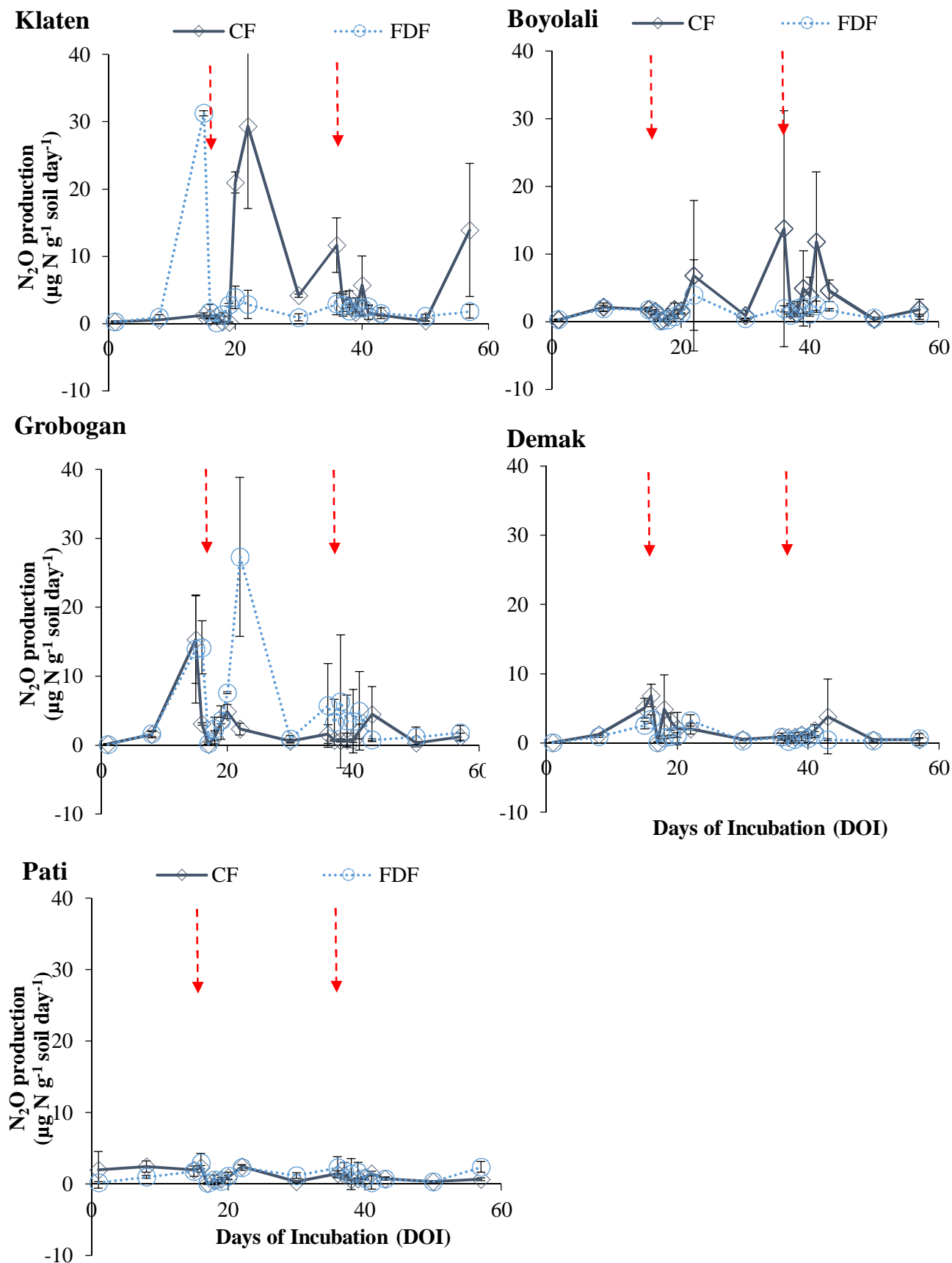
**Figure 2.5.** The total of CH<sub>4</sub> production during soil incubation,  $P < 0.05$

There were significant effect of different soil types and water treatments on the CH<sub>4</sub> production. The highest CH<sub>4</sub> production was from Grobogan soil under CF treatment approximately 26.92 mg C g<sup>-1</sup> soil and the lowest CH<sub>4</sub> production was from Pati under FDF treatment approximately 12.24 mg C g<sup>-1</sup> soil (Figure 2.5). Klaten's soil produced highest CH<sub>4</sub> followed by Grobogan, Boyolali, Pati and Demak's soil were approximately 27.57; 26.92; 18.53; 16.53; 14.33 and 14.29 mg C g<sup>-1</sup> soil, respectively. The CH<sub>4</sub> production from Klaten, Grobogan and Boyolali's soil higher compare to Demak and Pati likely occurred because of the chemical characteristic of the soil. Klaten, Grobogan and Boyolali's soil have high C and N content in the soil. Similar observation were reported by Inubushi *et al.* (1990) and Kimura (1992). Those

studies showed that there was positive correlations between the amounts of CH<sub>4</sub> formed in paddy soils and several soil parameters such as the content of C and N. Soil organic matter related with Eh and provide soluble C which favour the formation of both CH<sub>4</sub>.

### 2.3.2. Nitrous oxide production

The patterns of N<sub>2</sub>O production rate from five paddy soils were quite different from those of CH<sub>4</sub> production (Figure 2.6). The N<sub>2</sub>O production fluctuated during soil incubation and the pattern for each treatment were quite different. There was no consistent change in the rate of N<sub>2</sub>O production from CF and FDF treatments. In FDF treatment, sometimes peaks of N<sub>2</sub>O production occurred after re-flood. This finding was similar with Beare *et al.* (2009) that reported N<sub>2</sub>O production in soils was reduced by 93–96% during the drainage phase and the majority (88%) of the N<sub>2</sub>O production occurred after re-flooded from compacted soil. However, effect of drying and re-wetting the soil was inconsistent with those found in previous studies. Cai *et al.* (1997, 2001) reported that peak of N<sub>2</sub>O appeared at the beginning of the disappearance of floodwater in rice fields. The production of N<sub>2</sub>O in the FDF treatment from Klaten, Boyolali, Grobogan, Demak and Pati ranged approximately 0.12-31.27; 0.12-3.94; 0.10-14.18; 0.05-3.26 and 0.22-3.00 mg μg N g<sup>-1</sup> soil, respectively. However, the production of N<sub>2</sub>O in the CF treatment from Klaten, Boyolali, Grobogan, Demak and Pati were approximately 0.28-29.32; 0.09-13.73; 0.12-15.30; 0.04-6.84 and 0.09-2.46 μg N g<sup>-1</sup> soil, respectively.



**Figure 2.6.** The time course of N<sub>2</sub>O production rate during soil incubation. Red arrow means drainage the water in FDF treatment

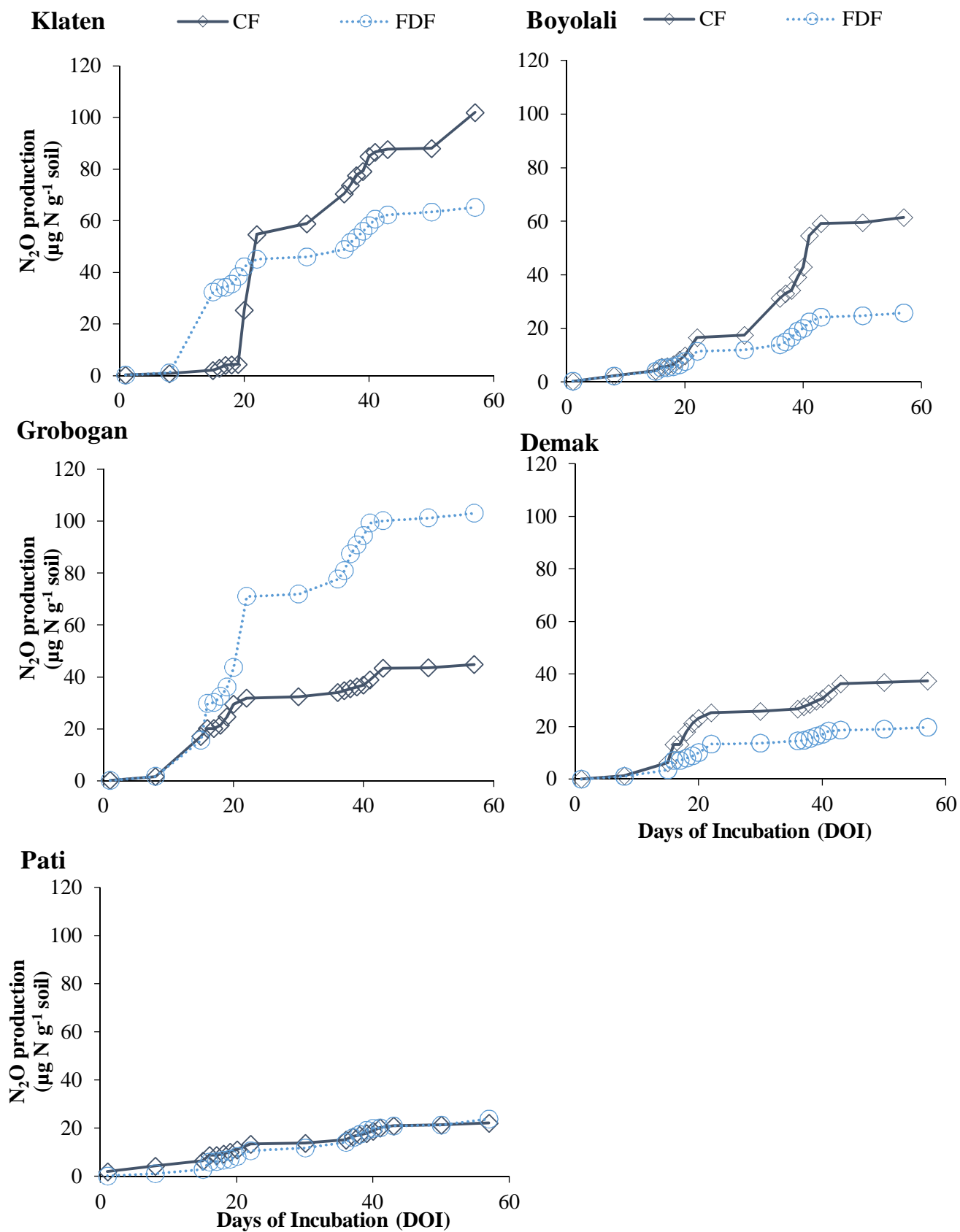
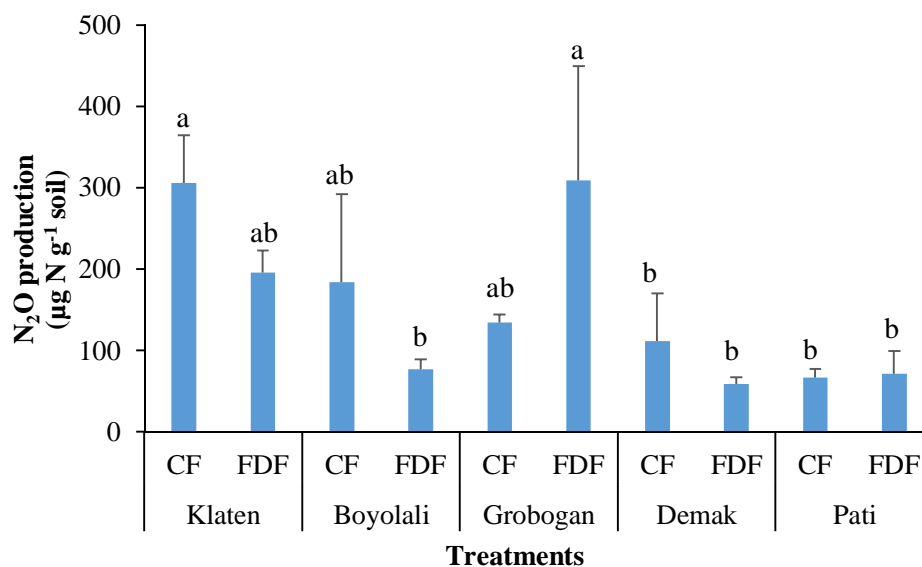


Figure 2.7. The pattern of cumulative N<sub>2</sub>O production during soil incubation

Figure 2.7 show that the  $N_2O$  production reached stationary phase in almost all soil in 43 day of incubation (DOI). Drainage stimulated cumulative production of  $N_2O$  only in Grobogan and Pati's soil, likely occur because FDF triggers interchangeable nitrification of ammonia, denitrification of nitrate and nitrifiers produced more  $N_2O$  than denitrifiers. Cai *et al.* (1997) and Zou *et al.* (2005) found that alternate flooding and drying cycle considerably increased  $N_2O$  emission. The highest  $N_2O$  production rate by nitrifiers was observed at 90 % WHC, when the soil had become partly anaerobic, as indicated by the high denitrification rate (Klemedtsson *et al.* 1988). While  $N_2O$  production was lower in Klaten, Boyolali and Demak's soil due to the drainage. During the CF period, less  $N_2O$  emission occurred as nitrification process was inhibited by flooding water and  $N_2$  was the main product of denitrification; also  $N_2O$  transport was retarded in water saturated soil (Freney and Denmead 1992; Granli and Bockman 1994).

Although there were very large variations in the  $N_2O$  production during soil incubation, there was interaction between soils and water treatments. The production of  $N_2O$  from the FDF treatment in Klaten, Boyolali and Demak's soil were lower than those from the CF treatment approximately 36.1; 58.0 and 46.9%, respectively. FDF treatment resulted in large increased in  $N_2O$  production relative to the CF treatments on Grobogan and Pati's soil approximately 130 and 7.6%, respectively, but in Pati's soil showed no significant difference between both treatments. On the other hand, FDF treatment stimulated  $N_2O$  production in Grobogan's soil likely occur because there was one big peak during 22 DOI in FDF treatment. This peak likely occur because the Eh reached the lowest value at that time. This condition could be happened likely occur because of Grobogan's soil contain high clay. When the soil was stirred the stirrer cloud not stir properly due to heavy soil. Therefore, the gas that entrapped in soil micro-pore cannot release and accumulated until the next gas sampling. Although  $N_2O$  is a by-product of nitrification and an immediate product of denitrification, the different effect

on N<sub>2</sub>O production due to water treatment in soil likely occurs because of length of incubation and differences in soil moisture of different soil types which affect N<sub>2</sub>O production (Klemedtsson *et al.* 1988).



**Figure 2.8.** The total of N<sub>2</sub>O production during soil incubation,  $P < 0.05$

Nitrous oxide patterns were greatly affected by soils (Xiong *et al.* 2007). The N<sub>2</sub>O production from incubation of Klaten, Boyolali, Grobogan, Demak and Pati's soil were approximately 251; 130; 221; 85 and 69 µg N g<sup>-1</sup> soil, respectively (Figure 2.8). Klaten's soil resulted the highest N<sub>2</sub>O production probably because the high of total N in Klaten's soil. According to Baggs *et al.* (2000) that more N will be available for nitrification and denitrification processes and higher N<sub>2</sub>O emissions may occur.

### 2.3.3. Soil redox potential

The difference of soil Eh between CF and FDF in each of the soil can be seen clearly in Figure 2.9. In all the soil FDF treatments could reach higher soil Eh compare to CF. Soil Eh from CF treatment in Klaten, Boyolali, Grobogan, Demak and Pati's soil were



approximately -238 to 156; -410 to 76; -163 to 54; -308 to 52 and -228 to 1 mV. However, Eh from FDF treatment in Klaten, Boyolali, Grobogan, Demak and Pati's soil ranged approximately 146 to 164; -256 to 13; -145 to 97; -166 to 46 and -227 to 55 mV. Wang *et al.* (1993) and Neue *et al.* (1994) reported that rice field management and indigenous soil characteristics, such as  $\text{Fe}_2\text{O}_3$ ,  $\text{SO}_4$ ,  $\text{MnO}_4$ , silt and carbon content, affect the potential  $\text{CH}_4$  production of soils. These properties could affect the redox status of the soils in a reduced condition; which in turn may influence to the production of  $\text{CH}_4$  by methanogenic bacteria. A rapid decrease in Eh after flooded due to high carbon content in soil clay, i.e, in Klaten, Boyolali and Grobogan, appears to explain the greater  $\text{CH}_4$  production potential.

Soils like Boyolali with high C content but low active Fe content attain Eh values less than  $-200$  mV soon after submergence. Similar observation was reported earlier by Ponnampurna (1972, 1981). Methanogenesis occur under strictly anaerobic condition. In this study, it can be seen clearly that Eh in CF lower than FDF, especially in Grobogan's soil (Figure 2.10). The low of soil Eh made  $\text{CH}_4$  production potential higher in Grobogan soil. A sufficient low redox (Eh) potential is required for  $\text{CH}_4$  production. Once the soil is flooded, organic matter starts decomposing accompanied with a stepwise biochemical reduction of the soil which is indicated by a lowering of the redox potential (Ottow 1981; Inubushi *et al.* 1984; Neue 1985). Soil Eh in water-logged condition is primarily related to the amount and kind of organic matter in soil (Ponnampurna 1972). Soil Eh was found negatively correlated with  $\text{CH}_4$  production in and emission from the flooded soil. However, Eh development alone may not be a good indicator for the onset of methanogenesis and should only be used when the soil and its  $\text{CH}_4$  production behavior have been characterized (Yagi *et al.* 1996; Sigren *et al.* 1997; Yao *et al.* 1999). Soil Eh is also an important factor affecting  $\text{N}_2\text{O}$  emissions from paddy fields.

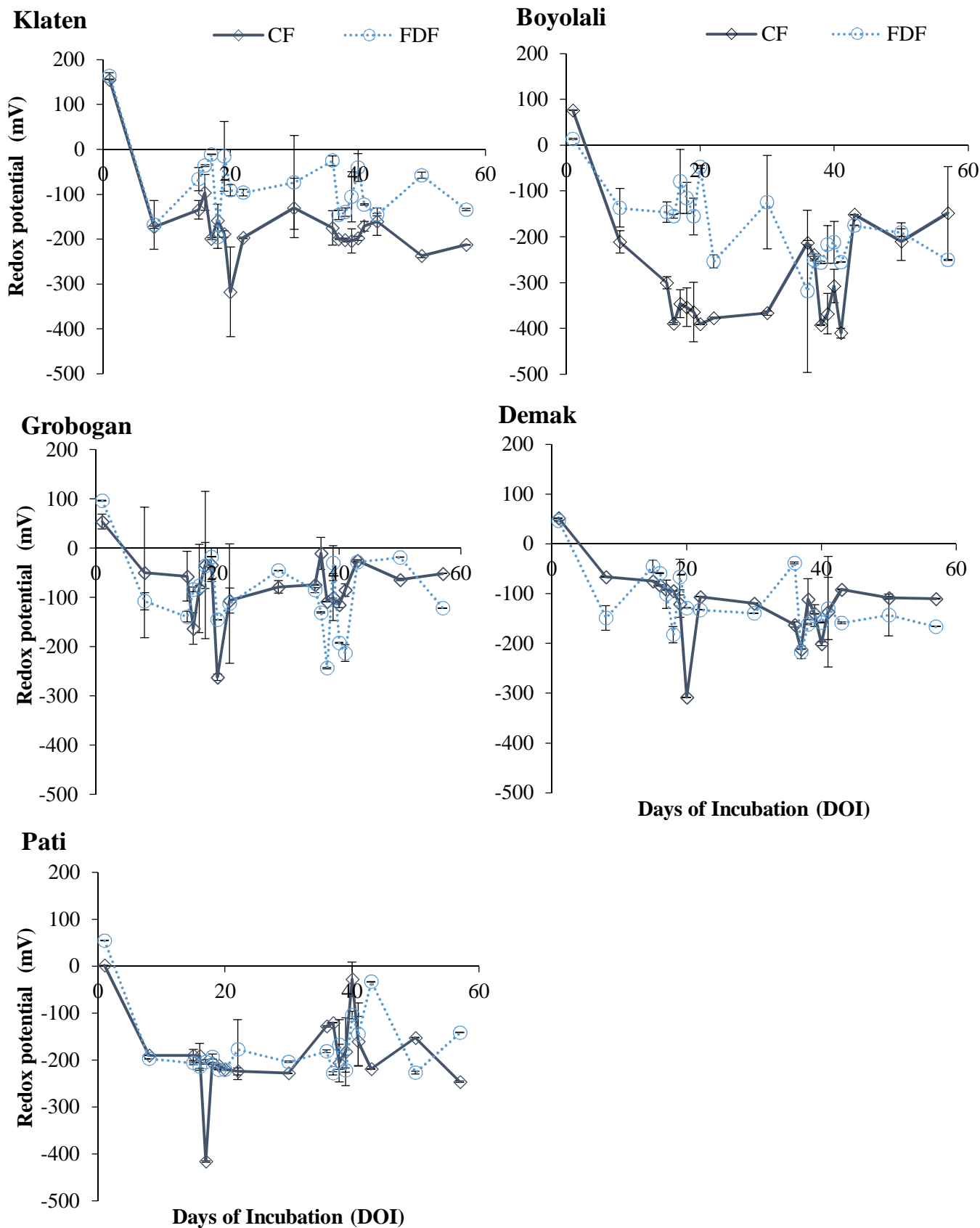
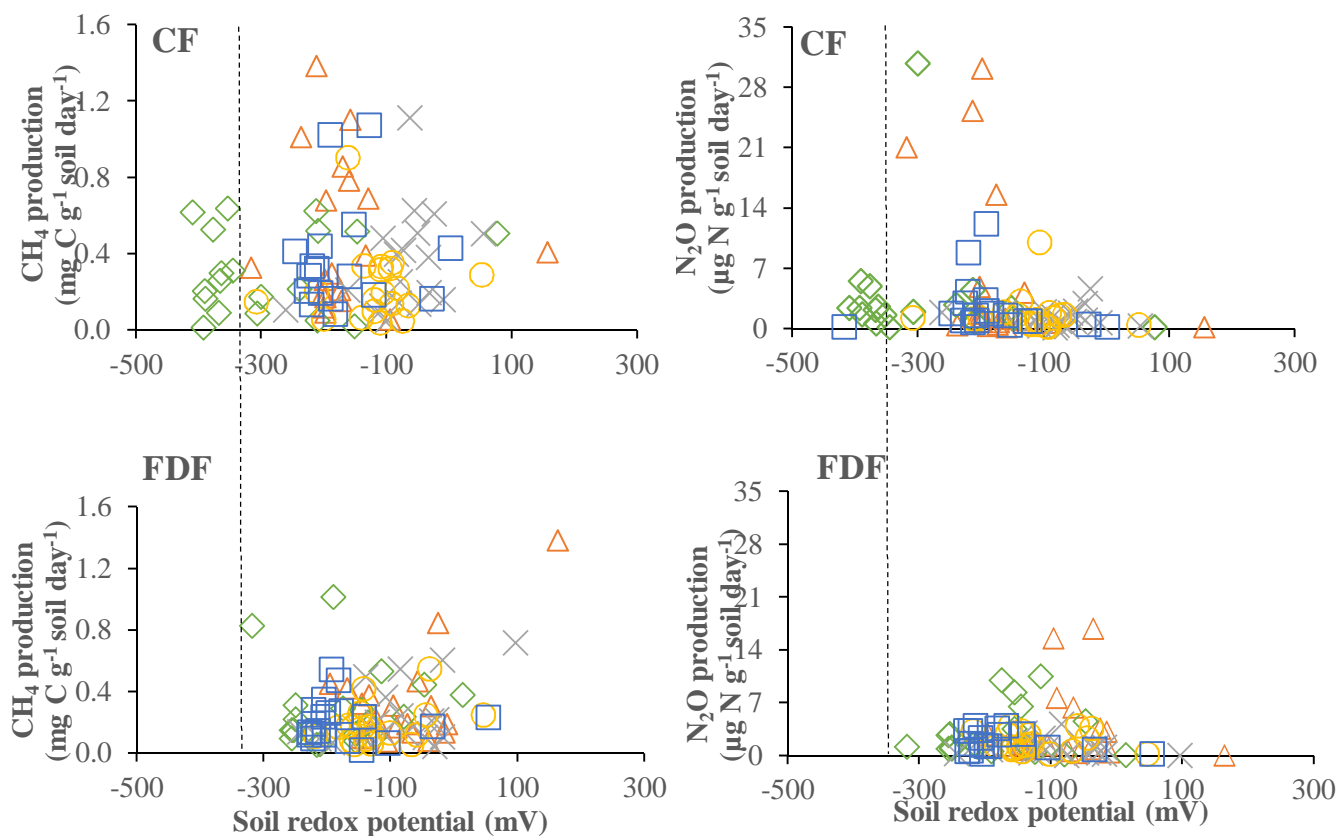


Figure 2.9. Redox potential during soil incubation



**Figure 2.10.** Correlation between CH<sub>4</sub>-N<sub>2</sub>O and soil Eh during soil incubation

### 2.3.4. Effect of soil properties on CH<sub>4</sub>-N<sub>2</sub>O emission

CH<sub>4</sub> production potentials showed pronounced variations among the different soils. Results of simple regression correlation analysis between different soil properties and CH<sub>4</sub>-N<sub>2</sub>O production indices are presented in Table 2.2. Soil properties like soil organic carbon, NO<sub>3</sub><sup>-</sup>, NH<sub>4</sub> and total Fe significantly affected the CH<sub>4</sub> production potentials. However, NO<sub>3</sub><sup>-</sup>, total Fe and total Mn significantly affected the N<sub>2</sub>O production potentials. In this study, simple regression correlation analysis showed that soil organic C had significant effect on CH<sub>4</sub> production. It could be due the dominating acetoclastic pathways for CH<sub>4</sub> production rather than hydrogenotropic pathway under waterlogged incubation (Conrad and Klose 1999). Wang *et al.* (1993) also observed no correlation between soil organic C and CH<sub>4</sub> production. Earlier study, Inubushi *et al.* (1990) and Kimura (1992) observed positive correlations between the

amount of CH<sub>4</sub> formed in paddy soils and several soil parameters such as the content of organic-C, water-soluble organic-C and mineralizable-N. The content of soil organic C lower the Eh and provide soluble C which favour the formation of both CH<sub>4</sub> and N<sub>2</sub>O. However, Yagi *et al.* (1990), on the contrary, found no correlation between CH<sub>4</sub> production rates and total C contents in soils.

**Table 2.2.** Correlation between soil characteristics and CH<sub>4</sub>-N<sub>2</sub>O production under CF and FDF treatments

| Variables                     | r values        |                  |
|-------------------------------|-----------------|------------------|
|                               | CH <sub>4</sub> | N <sub>2</sub> O |
| Soil organic carbon           | 0.443**         | 0.235            |
| SiO <sub>2</sub>              | 0.156           | 0.012            |
| P <sub>2</sub> O <sub>5</sub> | 0.055           | 0.093            |
| NO <sub>3</sub> <sup>-</sup>  | -0.404**        | 0.426**          |
| NH <sub>4</sub> <sup>+</sup>  | 0.355*          | 0.093            |
| Total Fe                      | -0.404**        | 0.584**          |
| Total Mn                      | -0.292          | 0.337*           |
| SO <sub>4</sub> <sup>-</sup>  | 0.124           | 0.222            |

\* $P < 0.05$ , \*\* $P < 0.01$ , without \* or \*\*: not significant.

The presence of NO<sub>3</sub><sup>-</sup> in the soil is one of the important factors controlling CH<sub>4</sub>-N<sub>2</sub>O production. It has been demonstrated that existence of NO<sub>3</sub><sup>-</sup> can inhibited CH<sub>4</sub> and stimulated N<sub>2</sub>O production. The presence of NH<sub>4</sub> in the soil is also one of the important factors controlling the CH<sub>4</sub>, NH<sub>4</sub> can stimulate CH<sub>4</sub> emission from rice paddy fields due to the competition of NH<sub>4</sub> for the oxidation with CH<sub>4</sub> by methanotrophs (Mosier *et al.* 1991). The NH<sub>4</sub> leads to an increase in nitrified population relative to methanotrophs and thus the overall CH<sub>4</sub> oxidations reduces, as nitrifiers oxidize CH<sub>4</sub> less efficiently than methanotrophs (Willson

*et al.* 1995). In the study there was correlation between CH<sub>4</sub>-N<sub>2</sub>O production and active Fe and Mn contents. The finding is synergy with those of Wang *et al.* (1993) who reported that high Fe and Mn in soils inhibited CH<sub>4</sub> production. Takai and Wada (1990) also postulated that the content of bio-active Fe may be the most important controlling factor in CH<sub>4</sub> production. Iron is an important oxidant for biological and chemical reactions that use oxidizing or reducing agents. Nitrite can be reduced by the presence of iron oxide at a near-neutral pH, and the end product is N<sub>2</sub>O and NO as an intermediate (Van Cleemput and Baert 1983; Van Cleemput 1998). In this study there was no significant correlation between CH<sub>4</sub> production and P<sub>2</sub>O<sub>5</sub> (available P). The findings do not comply with those of Adhya *et al.* (1998) who reported that addition of K<sub>2</sub>HPO<sub>4</sub> in soil had a stimulatory effect while Mussorie rock phosphate and single super phosphate had an inhibitory effect on CH<sub>4</sub> production, due to the presence of sulphur (S) in them.

### **2.3.5. Global warming potential and contribution of each gas**

The range of total GHG emission from CF and FDF treatment were approximately around 11-21 and 9-15 mg CO<sub>2</sub> eq g<sup>-1</sup> soil, respectively (Table 2.3). The highest GHG emission reduction was found in Boyolali followed by Klaten, Pati, Demak and Grobogan, were approximately around 30, 27, 19, 18 and 14%, respectively. High GHG emission reduction in Boyolali and Klaten's soil likely occur because high soil carbon. The content of soil organic C lower the Eh and provide soluble C which favour the formation of both CH<sub>4</sub> and N<sub>2</sub>O. The difference of Eh between CF and FDF treatments in Boyolali and Klaten were wider than the difference of Eh in other soils. Inubushi *et al.* (1990) and Kimura (1992) found positive correlations between the amount of CH<sub>4</sub> formed and the content of organic-C in paddy soils. To evaluate the climate implication of the cultivation practices, it is desirable to have relative contribution of each gas to global warming. In this study, CH<sub>4</sub> emission from different water

treatments account for 67-90% of the contribution to global warming, while N<sub>2</sub>O emission is only contributed 10-33%. Water treatments influence GHG emission by changing soil water content, which determines aerobic and anaerobic conditions in the soil. Aerobic and anaerobic conditions were related with soil redox potential (Eh), which has been used as one of the most indicative soil parameters for CH<sub>4</sub> and N<sub>2</sub>O from irrigated rice fields (Hou *et al.* 2000).

**Table 2.3.** Global warming potential and contribution of each gas

| Soil name | Water treatments | CH <sub>4</sub> | N <sub>2</sub> O | GHG emission<br>mg CO <sub>2</sub> eq g <sup>-1</sup> soil | Reduction (%) | Contribution each treatment to GHG emission |                  |
|-----------|------------------|-----------------|------------------|--|---------------|---|------------------|
|           |                  |                 |                  |  |               | CH <sub>4</sub>                             | N <sub>2</sub> O |
| Klaten    | CF               | 16              | 5.0              | 21   |               | 76  | 24               |
|           | FDF              | 12              | 3.2              | 15   | 27            | 79  | 21               |
| Boyolali  | CF               | 11              | 3.0              | 14   |               | 78  | 22               |
|           | FDF              | 8               | 1.3              | 10   | 30            | 87  | 13               |
| Grobogan  | CF               | 16              | 2.2              | 18   |               | 88  | 12               |
|           | FDF              | 10              | 5.1              | 15   | 14            | 67  | 33               |
| Demak     | CF               | 9               | 1.8              | 11   |               | 83  | 17               |
|           | FDF              | 8               | 1.0              | 9  | 18            | 89  | 11               |
| Pati      | CF               | 9               | 1.1              | 11   |               | 90  | 10               |
|           | FDF              | 7               | 1.2              | 9  | 19            | 86  | 14               |

### 2.3.6. Contribution of soil to national emissions

According to soil maps from ICALRRD that fifth of paddy soil from Klaten, Boyolali, Grobogan, Demak and Pati district, were classified as Inceptisols. Inceptisols is the largest paddy soil in Indonesia and the area was approximately around 59.69 million ha or around 32% of Indonesian's paddy soil (ISRI 2006). The average of CH<sub>4</sub> and N<sub>2</sub>O production from fifth soils were approximately 11.37 g and 2.66 g CO<sub>2</sub> eq ha<sup>-1</sup> year<sup>-1</sup>, respectively. Total emission from Indonesia for CH<sub>4</sub> and N<sub>2</sub>O ranged approximately 236,388 and 28,341 Gg

CO<sub>2</sub>eq (SNC 2010). Based on the area and the average of CH<sub>4</sub> and N<sub>2</sub>O production from Inceptisols, therefore Inceptisols contributed to national CH<sub>4</sub> and N<sub>2</sub>O production were approximately 0.68 and 0.16 Gg CO<sub>2</sub>eq, respectively (Table 2.4).

**Table 2.4.** National emission (SNC 2010) and contribution of soil to national emissions

|                  | <b>National emission</b>   | <b>Inceptisols</b> | <b>Contribution</b> |
|------------------|----------------------------|--------------------|---------------------|
| <b>Gases</b>     | <b>Gg CO<sub>2</sub>eq</b> |                    | <b>%</b>            |
| CH <sub>4</sub>  | 236,388                    | 0.68               | 0.000287132         |
| N <sub>2</sub> O | 28,341                     | 0.16               | 0.000560551         |

#### 2.4. Conclusions

Five rice soils from different locations in Central Java were incubated anaerobically for 57 days to determine methane (CH<sub>4</sub>) and nitrous oxide (N<sub>2</sub>O) production potentials and to establish relationships between chemical properties of soils and CH<sub>4</sub>-N<sub>2</sub>O production potential based on different water treatments. Compare to CF treatment, the FDF treatment in Klaten, Boyolali, Grobogan, Demak and Pati's soils reduced the CH<sub>4</sub> production by approximately 23.7; 21.6; 34.7; 11.4 and 21.9%, respectively. However, FDF treatment could not reduce N<sub>2</sub>O production in all soil types. Thus, we need to be cautious while recommending a particular irrigation regime for rice cultivation in order to avoid substantial emission of one or the other greenhouse gas. Soil organic carbon showed significant correlation with CH<sub>4</sub> production while total Mn showed significant correlation with N<sub>2</sub>O production. Total Fe, NO<sub>3</sub><sup>-</sup> and NH<sub>4</sub><sup>+</sup> have significant correlation with CH<sub>4</sub> and N<sub>2</sub>O production.

## CHAPTER 3

### EFFECT OF WATER MANAGERMENTS ON GREENHOUSE GAS EMISSION FROM PADDY FIELD IN INDONESIA

#### 3.1. Introduction

In previous chapter 2, evaluating CH<sub>4</sub> and N<sub>2</sub>O production under laboratory-scale has made clear that chemical characteristics of the soil is important factor that determine CH<sub>4</sub> and N<sub>2</sub>O production from the soil. It has been observed that water treatment could reduce CH<sub>4</sub> production in different soil types. There is a trade-off between CH<sub>4</sub> and N<sub>2</sub>O production in paddy soils. Somehow, reducing water from the soils stimulate N<sub>2</sub>O production. Based on previous study in chapter 2, the experiment related with water treatments should be conducted in field which use indicator plant to determine how CH<sub>4</sub> and N<sub>2</sub>O emissions from these types of treatments can be reduced and to examine how much water can be saved. In this chapter used one of the soils that was observed in previous chapter.

Rice field is an important source of CH<sub>4</sub> and N<sub>2</sub>O, but it can also be a sink for CO<sub>2</sub> through C sequestration into biomass products and soil organic matter (Johnson *et al.* 2007). Miyata *et al.* (2000) observed that net CO<sub>2</sub> flux from the rice paddy significantly larger when the field was drained than when it was flooded due to no diffusion barrier by the floodwater. While according to Alberto *et al.* (2009), soil CO<sub>2</sub> efflux is reduced due to limitation of diffusion of oxygen and suppression of CO<sub>2</sub> emissions in flooded fields. Ruser *et al.* (2006) found that CO<sub>2</sub> production from a fine-loamy soil fertilized with nitrate was not strongly effected by soil moisture. Liu *et al.* (2013) reported that there was a negative rate of CO<sub>2</sub> flux in the daytime and a positive throughout the night most likely because during the daytime plant photosynthesis uptake of CO<sub>2</sub> from both the atmosphere and from respired CO<sub>2</sub> emitted by the soil and floodwater. Respiration at night leads to an efflux of CO<sub>2</sub> to the atmosphere. In this



study, gas samples were taken in the morning and soil type in field experiment is loam, therefore CO<sub>2</sub> emission from this study was omitted.

Many cultivation practices have been improved in order to decrease environmental burdens of paddy rice production and to improve yield potential. The concern about the sustainability of food production has increased because of the increasing of world population and the growing of environmental issues. Due to increasing scarcity of freshwater resources, water-saving regimes are needed to reduce water use from rice field. Alternate wetting and drying (AWD) is a method in irrigated rice cultivation to save water, the rice fields are allowed to dry intermittently but still adequately supplied with water for the roots of rice plant to maintain rice yields. AWD is conducted by drying and re-flooding of the rice field and the time intervals between dry and wet conditions appear to be too short to facilitate the shift from aerobic to anaerobic soil conditions (Wassmann *et al.* 2000). Groundwater table are used in these system. The ponded water on the field is allowed to drop to 15–20 cm below the soil surface before irrigation is applied (Rejesus *et al.* 2011). On the other hand, oxic/anoxic boundary of the soil has important effects on GHG production (Dinsmore *et al.* 2009). CH<sub>4</sub> emissions are high under strictly anaerobic conditions (Moore and Dalva 1993), while N<sub>2</sub>O emissions are high in intermediate conditions (Davidson *et al.* 2000). It has been reported that mid-season drainage could mitigate CH<sub>4</sub> emission conversely it could lead to an increase in N<sub>2</sub>O emission (Bronson *et al.* 1997). Flooded rice fields are not a potent source of atmospheric N<sub>2</sub>O because N<sub>2</sub>O is further reduced to N<sub>2</sub> under the strong anaerobic conditions (Granli and Bockman 1994).

Simultaneous mitigation options are different for CH<sub>4</sub> and N<sub>2</sub>O emission, and minimizing one gas may increase the emission of the other, since the production of these two gases take place under contrasting conditions (Ghosh *et al.* 2003). So, the trade-off both of the emission should be well prepared for a balanced set of mitigation options, which optimize the

emission trade-off in minimum cumulative radiative forcing of the two gases on global warming potential (GWP), thus having a lowest possible greenhouse effect. On the other hand, this option have to be carefully sorted out if the mitigation option should not come in the way of achieving high crop yields. The objective of this chapter is to investigate the effects of AWD, site specific AWD and continuous flooded on CH<sub>4</sub>-N<sub>2</sub>O emission, yield and water productivity from paddy field in Indonesia

## **3.2. Material and methods**

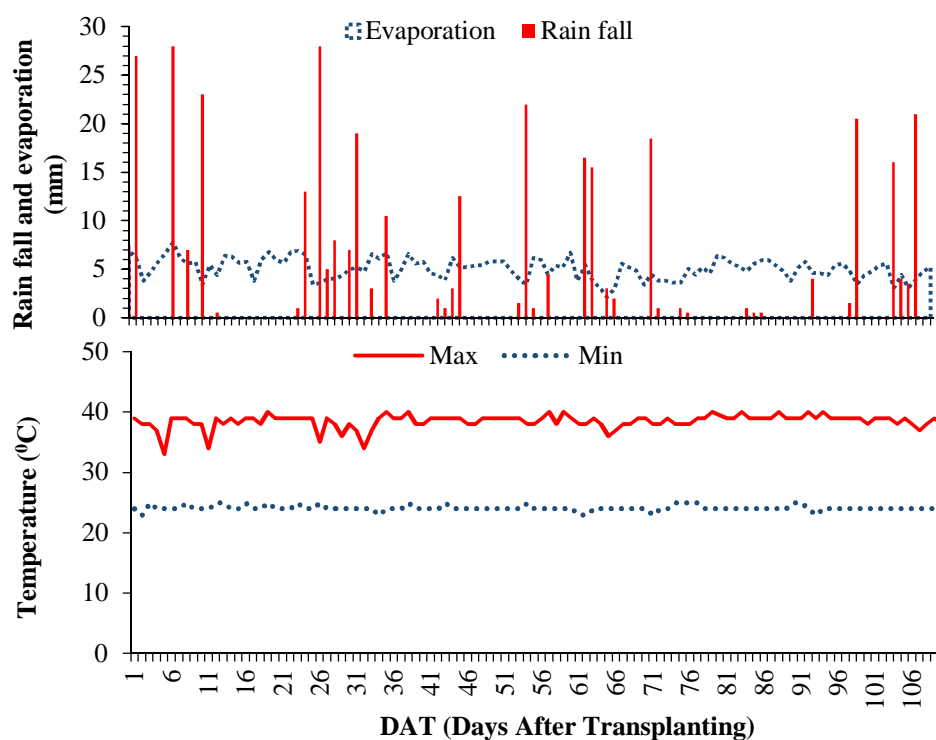
### **3.2.1. Site description**

The experimental field was located at the experimental farm of Indonesian Agricultural Environment Research Institute (IAERI), Jakenan (06<sup>o</sup>46'42.1" S, 111<sup>o</sup>11'52.0" E), in the Pati district, Central Java, Indonesia. The soil type at IAERI experimental farm is an Inceptisols (Aeric endoaquept) classified by USDA (2014). The physicochemical characteristics of the soil are listed in Table 3.1. The soil pH is 5.6 and soil texture is medium loam. The total carbon and nitrogen contents of the soil are 1.6 and 0.2 g kg<sup>-1</sup>, respectively.

The study were conducted during the rainy season (RS) 2014 which ran from March to June 2014. According to meteorological data that was collected from IAERI weather station, the mean annual air temperature in 2014 was 27.7°C, and the mean air temperature during March to June 2014 was 31.3°C. The annual rainfall in 2009 was 2000.3 mm, and the rainfall during March to June 2014 was 365.50 mm (Figure 3.1).

**Table 3.1.** Physicochemical characteristics of the experimental soil

| Parameters   | Values |
|--|--------|
| Texture (%)  |        |
| Sand   | 43     |
| Silt   | 40     |
| Clay   | 17     |
| pH   |        |
| H <sub>2</sub> O   | 5.6    |
| KCl  | 4.9    |
| EC (dS/m)  | 0.035  |
| Total (g kg <sup>-1</sup> )  |        |
| C  | 1.6    |
| N  | 0.2    |
| Available P (mg kg <sup>-1</sup> )                                       | 4.8    |
| Available K (mg kg <sup>-1</sup> )                                       | 14.1   |
| 1M NH <sub>4</sub> OAc extractable (cmol <sup>+</sup> kg <sup>-1</sup> ) |        |
| Ca   | 10.98  |
| Mg   | 0.85   |
| K  | 0.03   |
| Na   | 0.13   |

**Figure 3.1.** Meteorological data during the experimental period RS

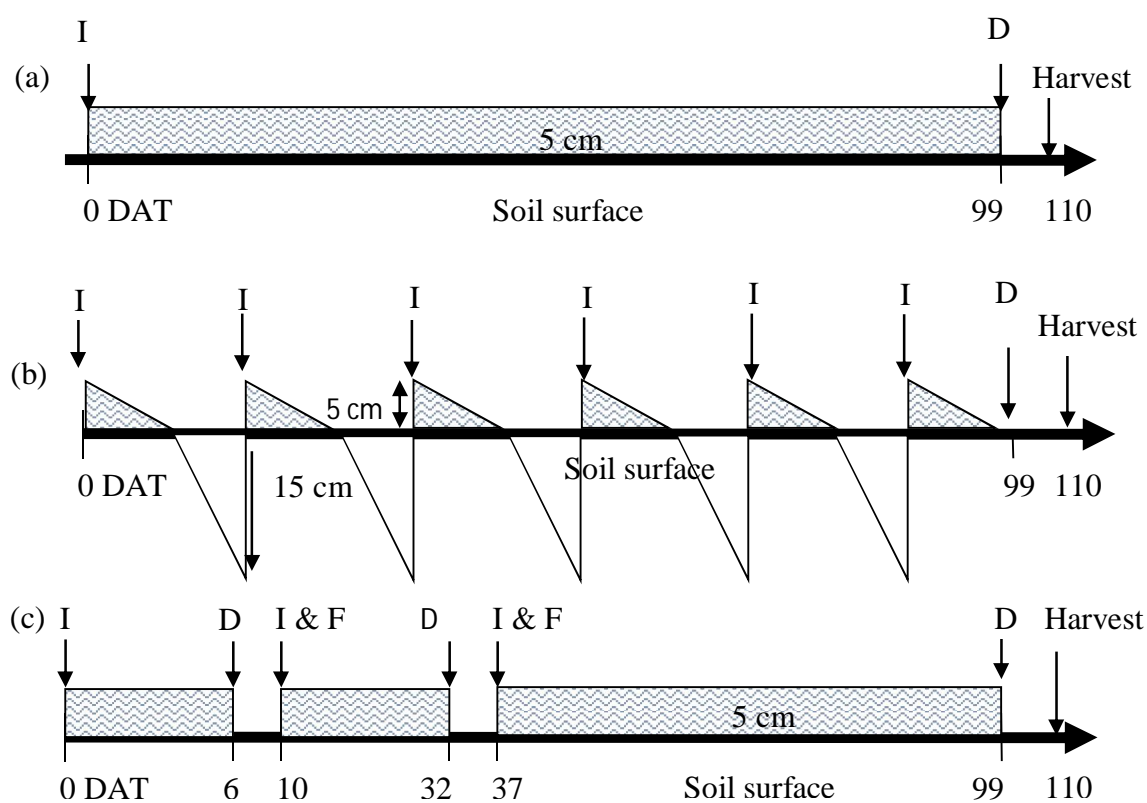
### 3.2.2. Experimental design and culture practices

The crops were established by transplanting, fourteen-day-old seedlings of the Cisadane rice variety were transplanted into each 5 m x 7 m plot, with 20 cm x 20 cm plant spacing and two-three seedling per hill. The physiological and agronomic characteristic of Cisadane is shown in Table 3.2. The days from sowing to harvest for this variety are 135 to 140 days. The fields were plowed and puddled thoroughly to a 10 cm depth 5 days before transplanting. Every experimental plot received fertilization at rates of 120 kg N ha<sup>-1</sup> (urea), 60 kg P<sub>2</sub>O<sub>5</sub> ha<sup>-1</sup> (super phosphate) and 90 kg K<sub>2</sub>O ha<sup>-1</sup> (potassium chloride). Super phosphate at 60 kg ha<sup>-1</sup> was applied 1 day before transplanting as the basal dose. Urea and K<sub>2</sub>O fertilizers were broadcast as three split applications at rates of 40 and 30 kg ha<sup>-1</sup> for each application, respectively. Urea and K<sub>2</sub>O fertilizers were applied 11, 38, and 56 days after transplanting (DAT).

**Table 3.2.** Physiological and agronomic characteristic of Cisadane

| Parameters             | Cisadane              |
|------------------------|-----------------------|
| Date release           | 18-Feb-80             |
| Origin                 | Pelita I-1/B2388      |
| Group                  | Cere (indica)         |
| Growth period          | 135 – 140 days        |
| Plant height           | 105 – 120 cm          |
| Productive tillers     | 15 – 20 hills         |
| Weight per 1000 grains | 29 g                  |
| Productivity           | 5 Mg ha <sup>-1</sup> |
| Yield potential        | 7 Mg ha <sup>-1</sup> |
| Plant shape            | Straight              |
| Foot colour            | Green                 |
| Auricle colour         | Colourless            |
| Leaf tongue colour     | Colourless            |
| Leaf colour            | Green                 |
| Leaf surface           | Coarse                |
| Leaf position          | Straight              |
| Grain colour           | Yellow                |
| Amylose content        | 20%                   |

The experiments were arranged in a randomized block design with 3 treatments replicated three times. The plots were comprised of water management: continuous flooding (CF), alternate wetting and drying (AWD) and site specific of AWD (S-AWD) (Figure 3.2). Piezometer was made from PVC pipes with 4 cm in diameter and 100 cm in length. Piezometer were installed in the field keeping 20 cm above the soil and the remaining 80 cm which was perforated underneath to measure the depletion of soil water in the field. Each of the plot was installed 2 pieces of piezometer. Water irrigation was controlled every day. In CF treatment, standing water was kept 5 cm above soil surface until 2 week before harvest. In AWD treatment, irrigation water was applied when depleting water table inside the pipe reached a 15 cm below soil surface. However, in S-AWD the water allowed to drain 1 week before first and second fertilization as long 7 days. The inner borders of the experimental plots were lined with plastic sheets up to 40 cm soil depth to prevent lateral water flow between plots.



Noted: I = irrigation, D = drainage and F = fertilization

**Figure 3.2.** Schematic diagram of different water managements (a) CF, (b) AWD and (c) S-AWD

### 3.2.2. Measurement of yield and yield component

Plants were harvested when they completely matured. The plants were harvested on 107 DAT. The harvested area of each plot was approximately 3 x 3 m for determination of yield per unit area. The information related to yield and all the yield component, i.e., plant height, effective tillers, length of the panicle, number of spikelet per panicle, number of filled and unfilled grains per panicle, 1000 grain weight, grain yield, straw yield were collected and harvest index were calculated. Grain yield was adjusted to 14 % seed moisture content. The biomass was dried at 70°C for 48 hours. According to Fageria *et al.* (2011), the grain harvest index was calculated by using the following formula:

$$\text{Grain harvest index} = \frac{(\text{Grain yield})}{(\text{Grain yield} + \text{rice straw})}$$

### 3.2.3. Measurement of water saving and water productivity

Water discharge from the irrigation pipe was calculated as the volume of water (m<sup>3</sup>) flowing through the pipe and measured as cubic meter per second (m<sup>3</sup> s<sup>-1</sup>). The time required to maintaining appropriate water levels in the main plots during each irrigation was noted and summed to calculate the total volume of water applied to the plots throughout the cropping season. The percentage of water saving was calculated as follows:

$$\text{Water savings (\%)} = \frac{\text{Water supply in flooded plot} - \text{water supply in AWD plot}}{\text{Water supply in flooded plot}}$$

Furthermore, water loss was also calculated based on the amount of water supplied in each plot. A simple measuring scale was used to determine the level of water (cm) lost each day during wetting period. Water Productivity Index (WPI) is water-use efficiency is intrinsically ambiguous in relation to crop production (Sharma 1989; Bhuiyan *et al.* 1995). WPI is

calculated as the ratio of crop yield ( $\text{kg h}^{-1}$ ) per unit water ( $\text{m}^3 \text{h}^{-1}$ ) supplied as defined by Jaafar *et al.* (2000). It includes irrigation and rainfall.

$$\text{Water productivity index (kg m}^{-3}\text{)} = \frac{\text{Grain yield (kg ha}^{-1}\text{)}}{\text{Total water supply (m}^3\text{ha}^{-1}\text{)}}$$

### 3.2.4. Gas sampling

The  $\text{CH}_4$  and  $\text{N}_2\text{O}$  fluxes were measured by closed chamber method and collected by using Plexiglas chambers during the rice-growing period (IAEA 1992). Each experimental plot had removable chambers for gas collection, which measured 50 cm x 50 cm x 100 cm for  $\text{CH}_4$  sampling and 40 cm x 20 cm x 30 cm for  $\text{N}_2\text{O}$  sampling. Access to the chambers in paddy fields was provided by small footbridges to avoid gas bubbles/ebullition in the field. Four hills of rice plants were covered in each sampling  $\text{CH}_4$  chamber, and no rice plants were covered in an  $\text{N}_2\text{O}$  chamber. There were 2 hills of rice plants that were left unplanted to leave space for  $\text{N}_2\text{O}$  chamber placement between the rice plants. Gas samples were collected once a week. However, when the plots of S-AWD was drainage, the gas sampling was collected every day. In each of the plot, gas sample of  $\text{CH}_4$  was measured in triplicate, while  $\text{N}_2\text{O}$  was measured once. Five gas samples from each chamber of  $\text{CH}_4$  and  $\text{N}_2\text{O}$  were collected with interval time at 0, 6, 12, 20 and 30 minutes started at 08:00 in the morning on each sampling day. Gas samples inside the chambers were collected using a 10-mL syringe fitted with a stopcock and transferred into 10 ml of vacuum vial then brought directly to the laboratory. The  $\text{CH}_4$  and  $\text{N}_2\text{O}$  concentrations were directly analyzed with a gas chromatograph (GC), which was equipped with a flame ionization detector (FID) for  $\text{CH}_4$  analysis and an electron capture detector (ECD) for  $\text{N}_2\text{O}$  analysis.

Soil pH and redox potential (Eh) were measured simultaneously with gas sampling. Platinum-tipped electrodes for determining the redox potential were inserted into the soil of

each plot to a depth of 0.1 m and remained there for the whole rice-growing period. In each plot, the electrode was set up in four replicates by the quadrat after transplanting. The soil pH-Eh was measured by using a portable pH-millivolt meter.

### 3.2.5. Calculation of GHG emission and GHGI

The emission rate was calculated based on the equation from IAEA (1992) as follows:

$$E = \frac{Bm}{Vm} \times \frac{\Delta C}{\Delta t} \times \frac{V}{A} \times \frac{273.2}{T + 273.2}$$

where E is the flux ( $\text{mg m}^{-2} \text{ min}^{-1}$ ),  $Bm/Vm$  ( $\rho$ ) is the density of gas ( $\text{mg m}^{-3}$ ),  $\Delta c/\Delta t$  is the average rate of concentration change with time ( $\text{ppmv min}^{-1}$ ), V is the volume of the chamber ( $\text{m}^3$ ), A is the base area of the chamber ( $\text{m}^2$ ), and T is the temperature in the chamber ( $^{\circ}\text{C}$ ). The total  $\text{CO}_2$ ,  $\text{CH}_4$  and  $\text{N}_2\text{O}$  emissions were calculated according Singh *et al.* (1999):

$$\text{CO}_2, \text{CH}_4 \text{ or } \text{N}_2\text{O emissions} = \sum_i^n (R_i \times D_i)$$

Where  $R_i$  is the rate of  $\text{CO}_2$ ,  $\text{CH}_4$  or  $\text{N}_2\text{O}$  flux ( $\text{mg m}^{-2} \text{ min}^{-1}$ ) in the  $i^{\text{th}}$  sampling interval,  $D_i$  is the number of days in the  $i^{\text{th}}$  sampling interval and n is the number of sampling intervals.

GHG emission equal with total  $\text{CO}_2$  emission equivalent were calculated by using the following formula (IPCC 2007b):

$$\text{GHG emission (kg CO}_2 \text{ eq ha}^{-1}) = (25 \times \text{CH}_4) + (298 \times \text{N}_2\text{O})$$

Greenhouse gas intensity (GHGI) was calculated as described in Shang *et al.* (2010):

$$\text{GHGI (ton CO}_2 \text{ eq ton}^{-1} \text{ grain yield)} = \text{GHG emission/yield}$$

### 3.2.6. Statistical analysis

The effects of the treatments were analyzed with SAS software (SAS Institute 2003). The significant effects of different sites and steel slag applications were examined by using a two-way analysis of variance (ANOVA). While the effects of water managements were



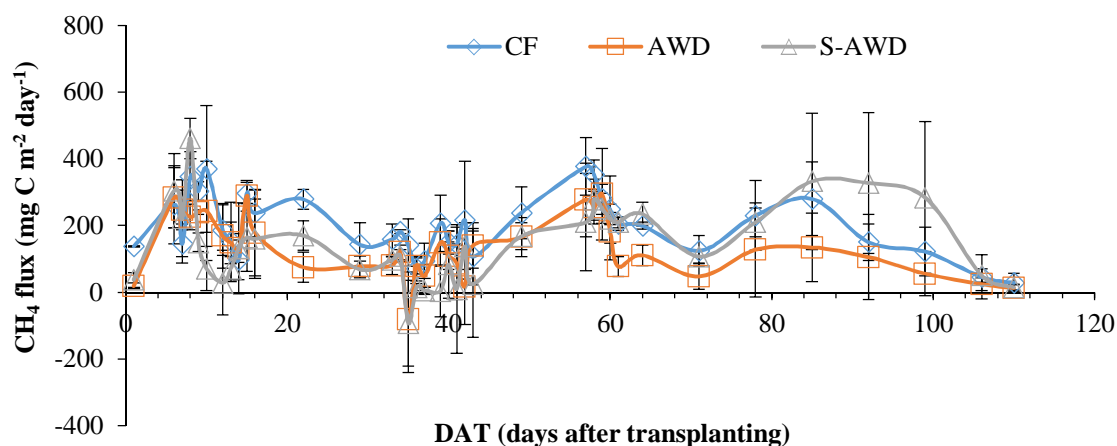
examined by using one-way ANOVA. When significant differences were detected at  $P = 0.05$ , the mean values were compared by using Tukey's HSD test (SAS Institute 2003).

### 3.3. Result and discussion

#### 3.3.1. Methane flux

The CH<sub>4</sub> fluxes under different water managements are shown in Figure 3.3. The CH<sub>4</sub> fluxes of AWD and S-AWD were comparably low, there were quite huge amounts of CH<sub>4</sub> degassing from the AWD and S-AWD when the water was drained. The CH<sub>4</sub> fluxes from CF, AWD and S-AWD were approximately 205, 134 and 147 mg C m<sup>-2</sup> day<sup>-1</sup>, respectively. During the measurement period, the peak of CH<sub>4</sub> flux was in CF at 57 DAT (468 mg C m<sup>-2</sup> day<sup>-1</sup>). Total CH<sub>4</sub> emission during rice growing period from CF, AWD and S-AWD approximately ranged 219.4; 143.8 and 157.4 kg ha<sup>-1</sup> season<sup>-1</sup>, respectively. In this study, there were CH<sub>4</sub> reduction approximately around 34.5 and 28.4 % by application of AWD and S-AWD, respectively. Sass *et al.* (1992) reported that multi-aeration decreased CH<sub>4</sub> emission by 12% compared to continuous flooding, without any decreases in rice yield. Yagi *et al.* (1996) reported that CH<sub>4</sub> emission with intermittent irrigation decreased to 45% of continuous flooding. Katayanagi *et al.* (2012) reported that alternate wetting and drying has the potential to reduce CH<sub>4</sub> emission by 73 % compared with traditional flooded rice. In the paddy soil ecosystem, CH<sub>4</sub> is produced by microbial activities in the extremely anaerobic conditions that resulted from flooding soils by irrigation or rain water. A part of CH<sub>4</sub> produced in the anaerobic layer of soil is re-oxidized at the oxidized zones in soil, and the rest is transported to the atmosphere, mainly via plants. Water management practise have a strong influence on the processes involved in CH<sub>4</sub> emission on rice paddy fields. The presence of surface standing water is essential for the development of the anaerobic conditions paddy soil by limiting the

transport of atmospheric oxygen into soil. Drying give to the higher Eh and to decrease the amount of CH<sub>4</sub> emitted.

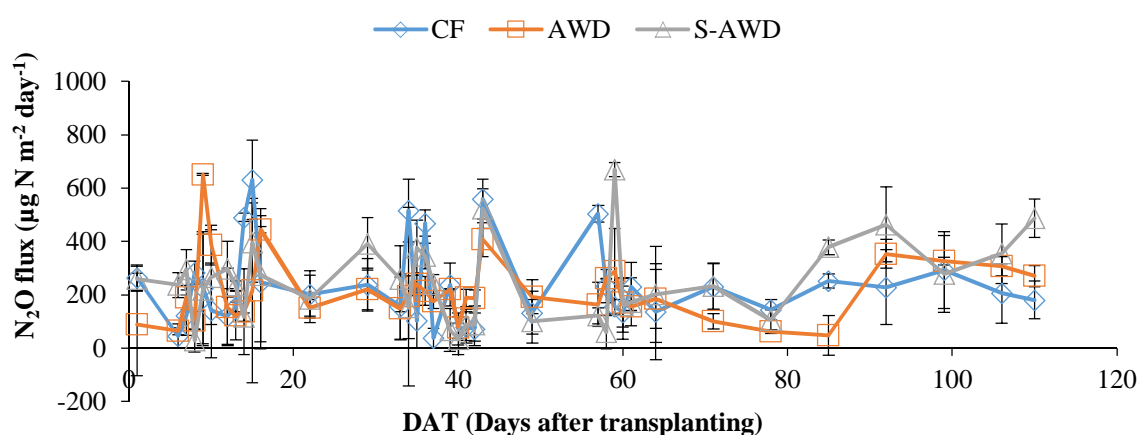


**Figure 3.3.** Dynamic changes of CH<sub>4</sub> fluxes under different water managements

### 3.3.2. Nitrous oxide emission

In general, the N<sub>2</sub>O fluxes at different water managements were quite low. Statistics revealed that there is no difference in N<sub>2</sub>O fluxes. N<sub>2</sub>O fluxes from CF, AWD and S-AWD range approximately 234; 218 and 248  $\mu\text{g N m}^{-2} \text{ day}^{-1}$ , respectively. The three treatments showed pronounced emission peak at different time (Figure 3.4). As already mentioned in previous studies, continuous flooded paddy field have less significant N<sub>2</sub>O emissions but emit great amounts of CH<sub>4</sub> (Smith and Patrick 1983; IPCC 1992, Neue and Sass 1998; Zou *et al.* 2005a; Yan *et al.* 2009; Cai *et al.* 1997), while there is a trend towards water-saving irrigation practices with less flooding which on the other hand reduce CH<sub>4</sub> emissions but increase N<sub>2</sub>O emissions (Smith and Patrick 1983; Cai *et al.* 2001; Zou *et al.* 2005b; Johnson-Beebout *et al.* 2009; Liu *et al.* 2010; Peng *et al.* 2011). In this results were partly in contrast to the other findings as we found reduced CH<sub>4</sub> emissions under conditions of less flooding, indeed, but we also found less N<sub>2</sub>O emissions with less flooding. AWD resulted lower N<sub>2</sub>O emission, while S-AWD showed higher N<sub>2</sub>O emission compare to CF. This result was similar to Berger *et al.*

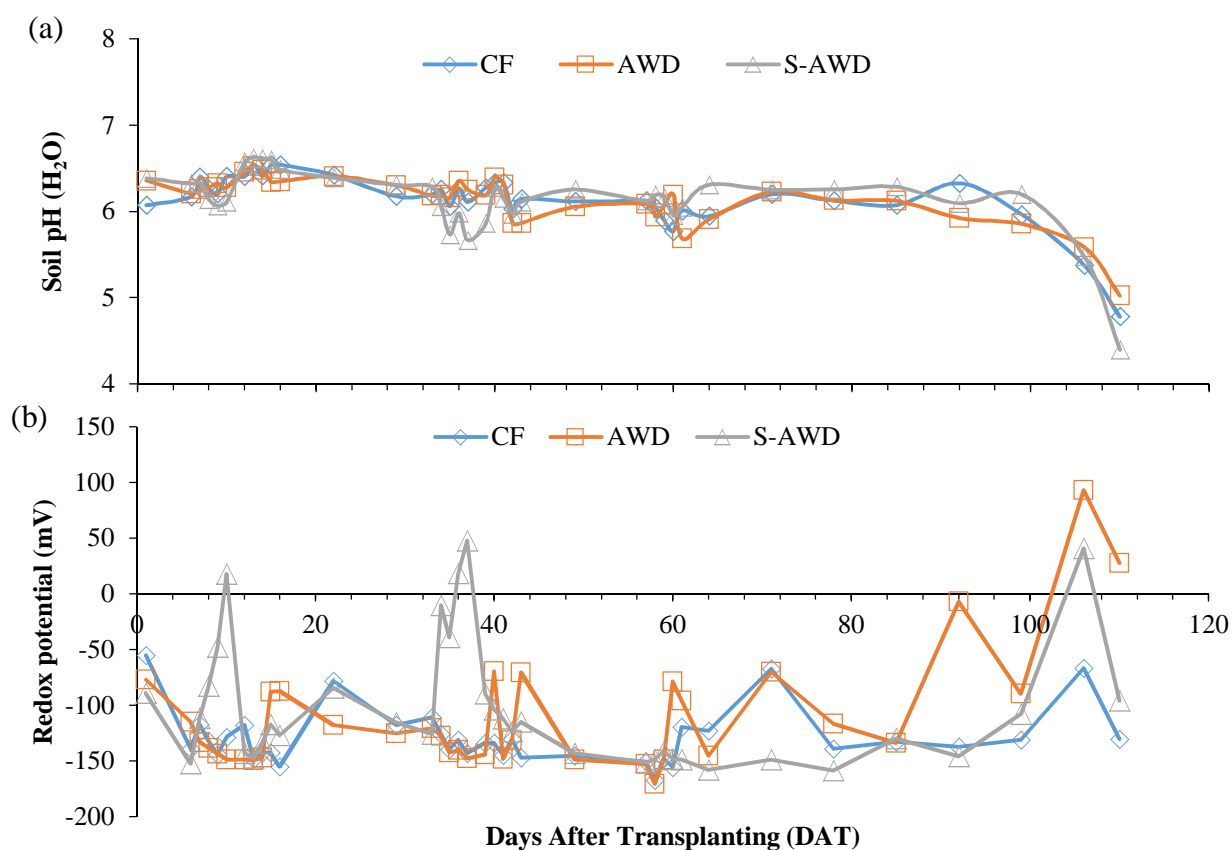
(2013) that AWD could reduce  $\text{CH}_4$  as well as  $\text{N}_2\text{O}$  emission from paddy field. The  $\text{N}_2\text{O}$  emission from CF, AWD and S-AWD were approximately 0.251; 0.233 and 0.265  $\text{kg ha}^{-1}$  season $^{-1}$ . As possible reasons for this finding, soil property such as the sandy substrate which most likely caused strong  $\text{NO}_3^-$  leaching, which may have made a huge  $\text{N}_2\text{O}$  production impossible. But this results also suggest that, with having  $\text{N}_2\text{O}$  production in the soil because short term environmental changes such as less flooding or short term fluctuations of the water table height during one vegetation period, may not have a strong effect on  $\text{N}_2\text{O}$  production in short term. While,  $\text{N}_2\text{O}$  is formed primarily from nitrification and denitrification in soil, depending on the aerobic and anaerobic conditions of soil (Mosier *et al.* 1998). Emissions of  $\text{N}_2\text{O}$  during intermittent irrigation periods strongly depended on the status of water logging in the fields. Different water regimes in rice fields caused a sensitive change in  $\text{N}_2\text{O}$  emissions (Zou *et al.* 2005). AWD give varies results in  $\text{N}_2\text{O}$  emission. Smith and Patrick (1983) observed that alternate anaerobic and aerobic cycling considerably increased  $\text{N}_2\text{O}$  emission relative to constant aerobic and anaerobic conditions. While Monteny *et al.* (2006) reported that drainage can suppress  $\text{N}_2\text{O}$  emissions by improving aeration.



**Figure 3.4.** Dynamic changes of  $\text{N}_2\text{O}$  fluxes under different water managements

### 3.3.3. Soil pH and soil redox potential

There was no effect of different water management on soil pH. The pattern of soil pH is similar among to the treatments. The soil pH values on CF were around 4.78-6.54, while pH values from AWD and S-AWD ranged approximately 5.02-6.48 and 4.40-6.60, respectively (Figure 3.5a). Mostly, increasing soil pH near to neutral enhance  $\text{CH}_4$  emission, i.e., the increase in the soil pH may have enhanced the activity of soil microorganisms, including that of methanogens, and this activity accelerated the decomposition of organic matter with an increase in  $\text{CH}_4$  emissions. In this study, the reduction of  $\text{CH}_4$  emission was not influenced by soil pH.



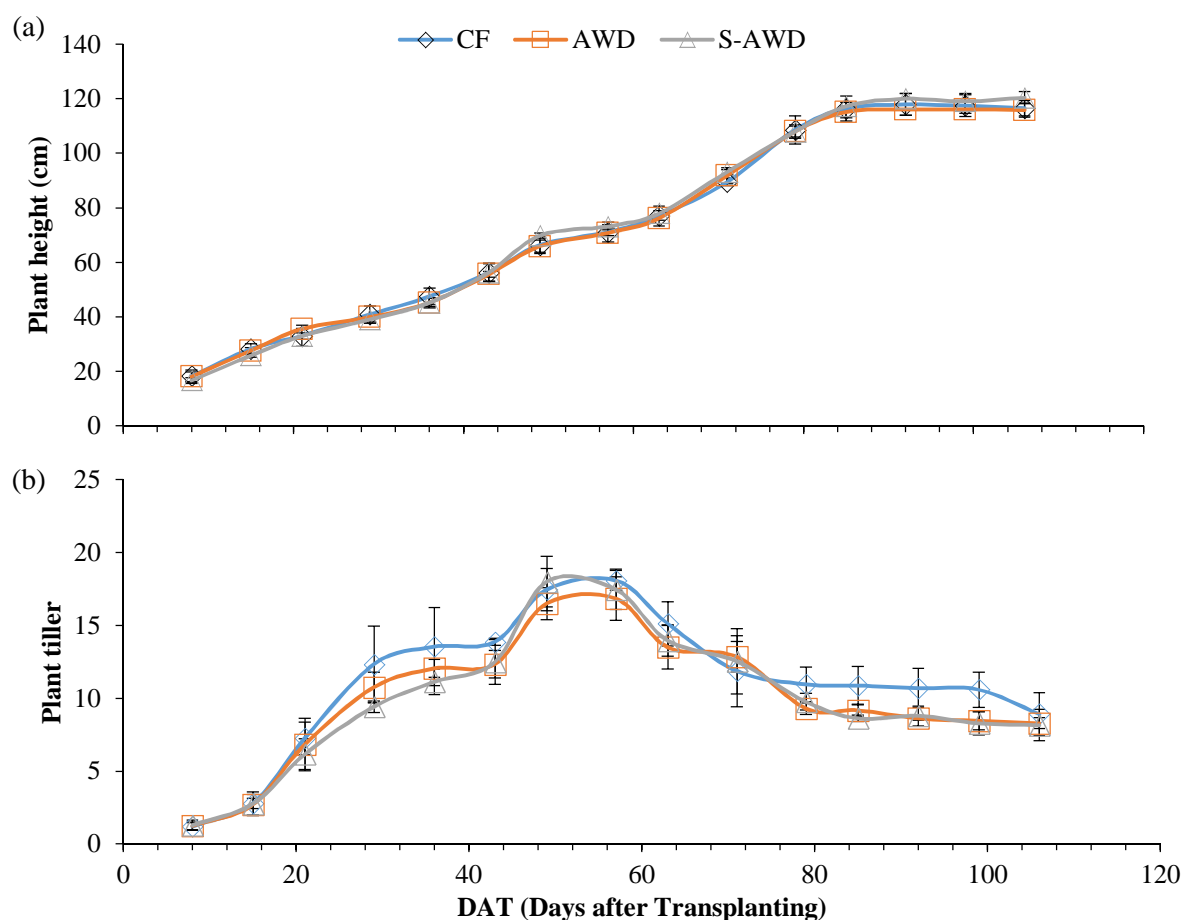
**Figure 3.5.** Dynamic changes of soil pH (a) and redox potential (b) under different water managements

Redox potential (Eh) pattern of the soil shows that water management influenced soil Eh. When the field was drained, soil Eh showed higher values (Figure 3.5b). Soil redox potential increased rapidly following the drainage of the field. AWD and S-AWD have higher values compare to CF. The range soil Eh values from CF were around -55 to 167 mV, while soil Eh from AWD and S-AWD were approximately -149 to 92 and -158 to 47 mV, respectively. Water managements influence GHG emission especially CH<sub>4</sub> by changing soil water content, which determines aerobic and anaerobic conditions in the soil. Aerobic and anaerobic conditions were related with soil Eh, which has been used as one of the most indicative soil parameters for CH<sub>4</sub> and N<sub>2</sub>O from irrigated rice fields (Hou *et al.* 2000). Conrad (1999) identified three sequential phases in soil reduction dynamics: (i) H<sub>2</sub>-dependent methanogenesis at positive Eh (360–510 mV), then (ii) sulfate or Fe (III) reductions when the first phase methanogenesis is becoming thermodynamically unfavourable, and (iii) vigorous acetate-dependent methanogenesis with a constant rate. The patterns in relation to soil Eh shows that CH<sub>4</sub> emission was highest when redox potential was lowest and N<sub>2</sub>O emission was at a minimum at the same time (Hou *et al.* 2000). The critical soil redox potential for CH<sub>4</sub> and N<sub>2</sub>O production has been demonstrated in laboratory studies to be below about -150 mV for CH<sub>4</sub> and above about -250 mV for N<sub>2</sub>O (Wang *et al.* 1993; Masscheleyn *et al.* 1993).

#### **3.3.4. Plant height and plant tiller**

Water management practice showed similar response in vegetative and reproductive characteristics. No significant effect of AWD, S-AWD and CF were observed in plant height and plant tiller. Maximum plant tiller were 18 (Figure 3.6a), while plant height were 120 cm (Figure 3.6b). This condition most likely because the application of water management still give a sufficient amount of nutrient in root zone to secure a high photosynthetic rate (Osaki *et al.* 1997). Yang and Zhang (2010) that reported AWD improves

water use efficiency and can improve yield by increasing the proportion of tillers that are productive, reducing the angle of the topmost leaves, (thus allowing more light to penetrate the canopy) and modifying shoot and root activity, implying altered root-to-shoot signalling of phyto-hormones such as abscisic acid (ABA) and cytokinins. While according to Koma and Sinv (2003) reported that reducing water in field lead to reducing adequate soil moisture that make less number of tiller per m<sup>2</sup>.



**Figure 3.6.** Plant height (a) and plant tiller (b) during rice growth under different water managements

### 3.3.5. Yield components

Yield components in all parameter measured between CF, AWD and S-AWD were not different. Among the yield components, no large different between the treatments that

caused yield one treatment higher or lower than another. These included 1000 weight grain (30.5; 30.1 and 29.6 g), % filled grain (67.9; 67.4 and 63.2%), and harvest index (0.38; 0.38 and 0.35) (Table 3.3). The highest panicle number per m<sup>2</sup> was for CF (2408) and no significant differences with AWD and S-AWD. The total biomass ranged between 0.929-1.073 kg m<sup>-2</sup>. No significant differences were observed among the water treatments on total biomass. Providing enough water for rice growth would promote percentage of filled grain as long as nutrient supply is sufficient and climate is favourable for rice plant growth (Yoshida 1981). In contrast with what was mentioned by Yoshida (1981), this study showed that water management do not affect the yield components. Tuong *et al.* (2009) stated that no yield penalty was observed when safe AWD was practiced. Bouman and Tuong (2001) summarized 31 field experiments on AWD and they found the yield reductions of 0–70% in AWD treatments compared with CF controls in 92% of the experiments. Hatta (1967); Tabbal *et al.* (1992), and Singh *et al.* (1996) reported that maintaining a very thin water layer, at saturated soil condition, or AWD can reduce water applied to the field by about 40–70 percent compared with CF, without a significant yield loss. Sato and Uphoff (2007) reported from Indonesian experience that CF was not essential for achieving high rice yields. The large variability in the performance of AWD was caused by differences in the irrigation interval, soil properties and hydrological conditions across the experiments. In addition, variety is a major factor that influences the performance of AWD (Peng and Bouman 2007). Contrary to Vizier (1990) and Sahrawat (2000), under CF condition rice yields tend to be very low due to detrimental to rice root growth, limited rice growth during the vegetative phase of rice and the chemical changes of paddy soil that affect the transformation of nutrient.

**Table 3.3.** Yield components of Cisadane under different water managements

| Yield components                       | Treatments     |                |                |
|--|----------------|----------------|----------------|
|  | CF             | AWD            | S-AWD          |
| 1000 weight grain (g)                  | 30.5 ± 0.17    | 30.1 ± 0.71    | 29.6 ± 0.27    |
| Unfilled grain                         | 354 ± 91.9     | 354 ± 95.4     | 453 ± 91.9     |
| Filled grain                           | 761 ± 184.6    | 741 ± 161.1    | 803 ± 233.3    |
| % filled grain                         | 67.9 ± 6.6     | 67.4 ± 8.8     | 63.2 ± 7.4     |
| Panicle number per m <sup>2</sup>      | 2075 ± 736     | 1742 ± 245     | 2408 ± 1054    |
| Panicle length (cm)                    | 23.0 ± 0.40    | 23.5 ± 0.71    | 22.8 ± 0.62    |
| Root dry weight (kg m <sup>-2</sup> )  | 0.116 ± 0.0294 | 0.115 ± 0.0361 | 0.116 ± 0.0376 |
| Shoot dry weight (kg m <sup>-2</sup> ) | 0.844 ± 0.2345 | 0.814 ± 0.1216 | 0.957 ± 0.3077 |
| Total biomass (kg m <sup>-2</sup> )    | 0.960 ± 0.2639 | 0.929 ± 0.1577 | 1.073 ± 0.3453 |
| Grain harvest index                    | 0.38 ± 0.084   | 0.38 ± 0.013   | 0.35 ± 0.062   |

### 3.3.6. Nutrient uptake

Table 3.4 shows nutrient uptake by Cisadane under different water managements. Water treatments did not significantly affect the uptake of N, P, K, Ca, Mg, Fe, Mn, Zn and Si by rice plants during their growth. According to Yang *et al.* (2004) that AWD methods enhance nutrient uptake because it can improve root morphology and root activity (Yang *et al.* 2004). Intermittent irrigation is believed to improve oxygen supply to rice root system with potential advantages for nutrient uptake (Stoop *et al.* 2002). Bonkowski (2004) has indicated that under more aerobic soil conditions, there will be larger populations of soil fauna that contribute to biological processes for supplying N needs of plants. Paddy soils characterized by high amount of Fe- and Mn-oxides and low cation exchange capacity, aerobic condition reduce the accumulation of soluble ferrous iron and manganese after submerging, which are toxic to rice plants under the continuously flooded conditions (Olaleye *et al.* 2001). However, contrary to their studies, Levit (1980) reported that nutrient uptake by crop plants is generally decreased



under water-stress conditions owing to a substantial decrease in transpiration rates and impaired active transport and membrane permeability and resulting in a reduced root-absorbing power of crop plants nutrient uptake from the soil solution is also closely linked to the plant root and soil water status. A decline in the soil moisture content is associated with a decrease in the diffusion rate of nutrients from the soil matrix to the absorbing root surface. Rice need silicon (Si) in large amounts for vigorous growth and high production. Silicon deposited in the leaves, stem and husk. The function of Si in plant are to mitigate fungal infection and pest attack, alleviates lodging and other abiotic stress, improves the light-interception ability by plants in a community and minimizes transpiration losses (Ma and Takahashi 2002).

**Table 3.4.** Nutrient uptake by Cisadane under different water managements

| Water managements | N                             | P | K  | Fe                             | Mn  | Zn | Si                 |
|-------------------|-------------------------------|---|----|--------------------------------|-----|----|--------------------|
|                   | -----g kg <sup>-1</sup> ----- |   |    | -----mg kg <sup>-1</sup> ----- |     |    | g kg <sup>-1</sup> |
| CF                | 19                            | 7 | 29 | 39                             | 205 | 10 | 33                 |
| AWD               | 14                            | 9 | 20 | 38                             | 149 | 13 | 32                 |
| S-AWD             | 13                            | 7 | 23 | 39                             | 153 | 8  | 33                 |

### 3.7. Water saving and water productivity index (WPI)

The CF treatment needed supply water approximately around 765.7 m<sup>3</sup> ha<sup>-1</sup> followed by AWD and S-AWD ranged approximately 712.4 and 677.0 m<sup>3</sup> ha<sup>-1</sup>, respectively (Table 3.5). Although no significance effect of water treatment on water supply in paddy field, AWD and S-AWD could save the water approximately 6.96 and 11.59%, respectively. According to Bhuiyan (1992), under traditional practices in the Asian tropics and subtropics rice requires water between 700-1500 mm per cropping season depending on soil texture. Tuong and Bouman (2003) reported that the total water input varies from 700 to 5300 mm for 100 day per season in lowland rice field in tropic, depending on climate, soil characteristics and hydrological conditions. Although no significance different, S-AWD and AWD resulted

higher WPI compare than CF. Study from Chapagain *et al.* (2011) showed that total water required in AWD plot less 29% as compare to conventionally flooded plot in Chiba Japan, while WPI was significantly higher ( $1.7 \text{ kg m}^{-3}$ ) than conventional irrigation ( $1.3 \text{ kg m}^{-3}$ ). Bhuiyan and Tuong (1995) reported that a standing depth of water throughout the season is not needed for high rice yields. They added that about 40–45 percent of the water normally used in irrigating the rice crop in the dry season was saved by applying water in small quantities only to keep the soil saturated throughout the growing season, without sacrificing rice yields. Alternate drying and wetting of the fields allows for good aeration of the soil and better root growth thereby increasing rice yield and water use efficiency in Indonesia (Uphoff, 2006).

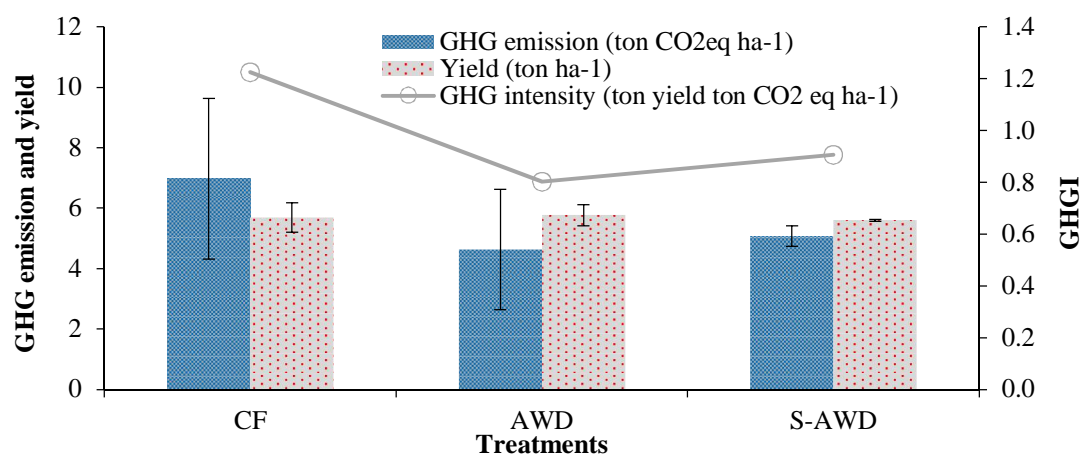
**Table 3.5.** Water supply, water saving and water productivity index (WPI) under different water managements

| Treatments | Water supply<br>( $\text{m}^3\text{ha}^{-1}$ ) | Water saving<br>(%) | WPI<br>( $\text{kg m}^{-3}$ ) |
|------------|--|---------------------|-------------------------------|
| CF         | 765.7  |                     | 7.44                          |
| AWD        | 712.4  | 6.96                | 8.10                          |
| S-AWD      | 677.0  | 11.59               | 8.27                          |

### 3.3.8. Total GHG emission and GHGI

The effects of water managements on the GHG emission, rice yield, and GHGI are presented in Figure 3.7. GHG emissions from CF, AWD and S-AWD were 6.97, 4.63 and 5.08 ton  $\text{CO}_2 \text{ eq ha}^{-1} \text{ season}^{-1}$ , respectively. Although, there was no statistical difference, there were tendency of GHG emission reductions approximately 33.6 and 27.2% due to application of AWD and S-AWD, respectively. Hadi *et al.* (2010) and Feng *et al.* (2013) reported 34 and 54% less GWP ( $\text{CH}_4$  and  $\text{N}_2\text{O}$ ) of intermittent irrigation as compared with traditional flooding. The AWD and S-AWD did better than the CF treatment in term of GHG emission. Therefore, it can

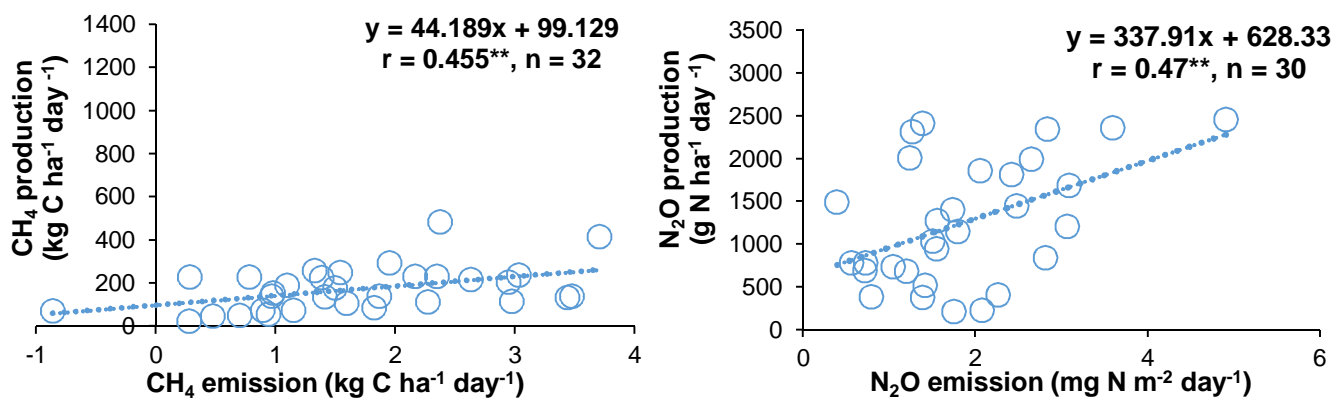
be expected that AWD and S-AWD would usually produce less GHG emission than CF. To evaluate the climate implication of the cultivation practices, it is desirable to have relative contribution of each gas to global warming. In this study, CH<sub>4</sub> emission from different water management account for 95-97% of the contribution to global warming, while N<sub>2</sub>O emission is only contributed 3.4-4.9%. No significance difference on yield but calculated yields were highest under AWD which produced better than under CF conditions. Yield of AWD was 1.3% higher than under CF conditions. It has been reported by Tuong *et al.* 2005; Yang *et al.* 2007; Zhang *et al.* 2008 that AWD can maintain or even increase grain yield because of the enhancement in root growth, grain-filling rate, and remobilization of carbon reserves from vegetative tissues to grains, when it compared with CF conditions. The lower value of the GHGI from AWD and S-AWD compared to control means that the treatments give more advantages to mitigate GHG emission and produce more rice.



**Figure 3.7.** Effects of water managements on GHG emissions, yield and GHG intensity at paddy field during rice growing season

### 3.3.9. Relationship between CH<sub>4</sub>-N<sub>2</sub>O production and CH<sub>4</sub>-N<sub>2</sub>O emission from Pati's soil

Regression analysis was done between CH<sub>4</sub>-N<sub>2</sub>O production from chapter 2 as *y* axis and CH<sub>4</sub>-N<sub>2</sub>O emission from this chapter as *x*. Pati's soil was used in this regression analysis. The relationship between CH<sub>4</sub>-N<sub>2</sub>O production and CH<sub>4</sub>-N<sub>2</sub>O emission is shown in Figure 3.8. Results show that there were significant relationship ( $P \leq 0.01$ ) between potential CH<sub>4</sub>-N<sub>2</sub>O production and CH<sub>4</sub>-N<sub>2</sub>O emission. This results can be used to predict CH<sub>4</sub>-N<sub>2</sub>O emission from rice field using CH<sub>4</sub>-N<sub>2</sub>O production from the same soil under incubation experiment. The equation from linear regression for CH<sub>4</sub> is  $y = 44.189x$ , with coefficient determination (*r*) of 0.46, while for N<sub>2</sub>O is  $y = 337.91x$ ,  $r = 0.47$ . Thus, based on the equation, CH<sub>4</sub> emission equals to 0.023 (2.3%) of the potential CH<sub>4</sub> production and, N<sub>2</sub>O emission equals to 0.003 (0.3%) of the potential N<sub>2</sub>O production. The CH<sub>4</sub>-N<sub>2</sub>O production resulted higher value than CH<sub>4</sub>-N<sub>2</sub>O emission from rice field.



**Figure 3.8.** Relationship between potential CH<sub>4</sub> – N<sub>2</sub>O production (*y*) and measured CH<sub>4</sub> – N<sub>2</sub>O emission (*x*) from rice soil in Pati, Central Java, Indonesia

### 3.4. Conclusions

Field experiments on water management method of cultivating rice have demonstrated the utility of AWD and S-AWD for water saving in irrigated rice farming. This

experiment also indicated that Water Productivity Index increased from continuous flooded irrigation. This field experiment confirms that AWD and S-AWD is a promising method in irrigated rice cultivation with benefits on water saving and maintaining the productivity comparable to continuous flooded irrigation. The increased productivity of water and its resource saving aspects are likely to be the critical factors that will make farmers and other stakeholders adopt AWD in water-scarce areas. However, it is difficult to draw general conclusions as AWD and S-AWD methods adopted in a certain area may not transfer to other areas because of variability in topography, soil, and climatic conditions across the rice agro-ecological domains. Therefore, it is important that comparative studies be conducted in different environments to verify this practice as a way to conserve water under conditions of water scarcity while maintaining, or increasing, crop yields. Moreover, long-term experiments are required to predict water management impacts on soil organic matter and provide leading indicators of sustainability, which can serve as an early warning system to detect impairments that threaten future productivity

## CHAPTER 4

### INFLUENCE OF WATER TABLES AND SOIL AMELIORATIONS ON GREENHOUSE GAS EMISSIONS FROM INDONESIAN PEAT SOIL COLOUM

#### 4.1. Introduction

In our previous chapter, it has been observed that of AWD and S-AWD resulted lower GHGI as well as higher WPI compare to CF irrigation. This field experiment confirms that AWD and S-AWD is a promising method in irrigated rice cultivation with benefits on reduce GHG emission, water saving and maintaining the productivity comparable to continuous flooded irrigation. However, the area of rice cultivation in irrigated area is expected to reduce due to the land conversion. The option to fulfil the increasing food demand is looking toward the areas of new arable land including peatland. However, utilization of natural peatland cause changes on ecosystem. Therefore, utilization of peatland for agriculture is better conducted in degraded peatland, i.e., ex-burned peatland because peat fire is a major cause of peatland degradation that leads to loss of biodiversity and carbon stocks. Sustainable agriculture means increase the production as well as ecology adaptive to environment. Thus, the experiment in this chapter used degraded peat soil to examine water table and soil ameliorant on GHG emission for future agriculture usage in peatland.

Large areas of tropical forest peatland in Indonesia have been converted into agricultural and non-agricultural sectors because of human population growth and economic development. Approximately 14.9 million ha of peatlands are found in Indonesia, which are estimated to account for 47% of the total tropical peatland area (Ritung *et al.* 2011; Page *et al.* 2011). Peatland has huge amount of carbon stock and nitrogen which could be a source of carbon dioxide (CO<sub>2</sub>), methane (CH<sub>4</sub>) and nitrous oxide (N<sub>2</sub>O) (Inubushi *et al.* 2003). To be

used for agricultural activities, peat soils need to be drained, limed and fertilized due to excess water, low nutrient content and high acidity. Regarding agricultural practices in wetland including peatland in Indonesia, there is indigenous knowledge of organic matter management that combines minimum tillage known as *tajak*, *puntal*, *hambur* systems, followed by application of ash, salt, and manure (Noor 2012). Therefore, water adjustment and soil amelioration are needed for suitable cultivation in peatland area due to the excess water and low soil fertility. However, lowering the water table increases carbon mineralization and CO<sub>2</sub> and N<sub>2</sub>O emissions but decreases CH<sub>4</sub> emission (Moore and Dalva 1993; Regina *et al.* 1996; Berglund and Berglund 2011). Soil ameliorants are applied not only to enhance the nutrient status of the soil and to improve crop yield but also to reduce GHG emission. Most of the studies on soil ameliorations/amendments have been conducted in mineral soil; however, there are few publications on the effects of soil amelioration in peat soil.

Most studies in tropical peatland were based on remote sensing data (Jaenicke *et al.* 2008). In the field, daily GHG emissions vary depending on climate condition and hydrologic regime. Therefore, we conducted this study by investigating the influence of water depth and soil amelioration on peat soil columns adjusted to the same conditions. The objective of this chapter is to discuss the effect of water table and soil amelioration on GHG emissions from Indonesian peat soil columns.

## **4.2. Materials and methods**

### **4.2.1 Site and soil sampling**

A soil sampling site was selected in Jabiren, Pulang Pisau district, Central Kalimantan, Indonesia (S 02°30'52.5"; E 114°10'11.6"). The soil sampling area is bounded by Sungai (river) Jabiren and Sungai Kahayan. The water level of this area is controlled by small canals that are approximately 50 cm deep and 50-75 cm wide, respectively. The canals were

constructed from 1995 as part of the Mega Rice Project (Firmansyah *et al.* 2013). The water table at this site is approximately 30-60 cm below the soil surface. The peatlands in this area are dome shaped (Hikmatullah *et al.* 2013). This area was burned in 2005. The peatland areas are extensive and there are various types of land-use managements; however the soil was sampled in a fallow area under natural vegetation and near rubber plantations which were established in 2006. The soil sampling location was covered with fern and grasses.

**Table 4.1.** Chemical characteristic of peat soil collected from Central Kalimantan, Indonesia

| Parameters   | Values |
|--|--------|
| pH   |        |
| H <sub>2</sub> O   | 3.1    |
| KCl  | 2.3    |
| EC (dS m <sup>-1</sup> )                                   | 0.028  |
| Organic (g kg <sup>-1</sup> )                              |        |
| C  | 336.5  |
| N  | 9.7    |
| C/N  | 35     |
| Available P (mg kg <sup>-1</sup> )                         | 3.49   |
| Available K (mg kg <sup>-1</sup> )                         | 22.41  |
| Exchangeable cations (cmol <sub>c</sub> kg <sup>-1</sup> ) |        |
| Ca   | 20.37  |
| Mg   | 2.07   |
| K  | 0.05   |
| Na   | 0.17   |
| CEC  | 88.45  |
| Base saturation (%)  | 26     |
| Ash content (g kg <sup>-1</sup> )                          | 187.9  |
| Silicate (g kg <sup>-1</sup> )                             | 45.2   |
| Humic acid (%)   | 10.91  |
| Pyrite (%)   | 0.06   |

Twenty-seven peat in column (0-100 cm) samples were collected in March 2014. The peat samples were collected approximately 20 m from the small canal. The peat depth at



this site was approximately 5-7 m (Firmansyah *et al.* 2013). The decomposition status of the peat material is dominantly classified as hemic (moderately humified) and fibric (less humified) but in the surface were classified as sapric (most humified) (Hikmatullah *et al.* 2013). The characteristics of the soil properties are listed in Table 4.1. To collect the peat samples, the top of the peat soil was removed using knife approximately 2-3 cm. The soil columns were collected using with polyvinyl chloride (PVC) pipe with a diameter of 21 cm and a length of 100 cm. The PVC pipe was inserted vertically to the soil to a depth of 100 cm and carefully pulled up with the soil inside after sealing the bottom. The litter inside the soil columns was not removed. The base of the soil columns was closed permanently with a tight cap during excavation, while the top of the columns was sealed with removable cap after the excavation. The columns were transported to Indonesian Agricultural Environment Research Institute (IAERI), Jakenan, Central Java, Indonesia.

#### 4.2.2. Experimental design

After been transported to IAERI, the excess water from the soil columns was allowed to drain to change the air-filled porosity of the peat samples. The peat samples were weighed to determine the moisture content. Then, holes were made in each soil column based on the water depth position. Each soil column was dipped into large bucket in the greenhouse to control the water depth. The water depth of each bucket was checked using the transparent tube (diameter of 10 mm) that was installed on the outside wall of each the bucket. The water depths were checked daily to ensure they remained at constant level. The columns for the water depth manipulation are shown in Figure 4.1. Rain water was used to set the water depths. A randomized block design with two factors was used to establish 3 different water depths (15 cm, 35 cm and 55 cm from the soil surface) and 3 different ameliorants (without ameliorant/control, 2.5 Mg ha<sup>-1</sup> biochar + 2.5 Mg ha<sup>-1</sup> compost; 2.5 Mg ha<sup>-1</sup> steel slag + 2.5 Mg

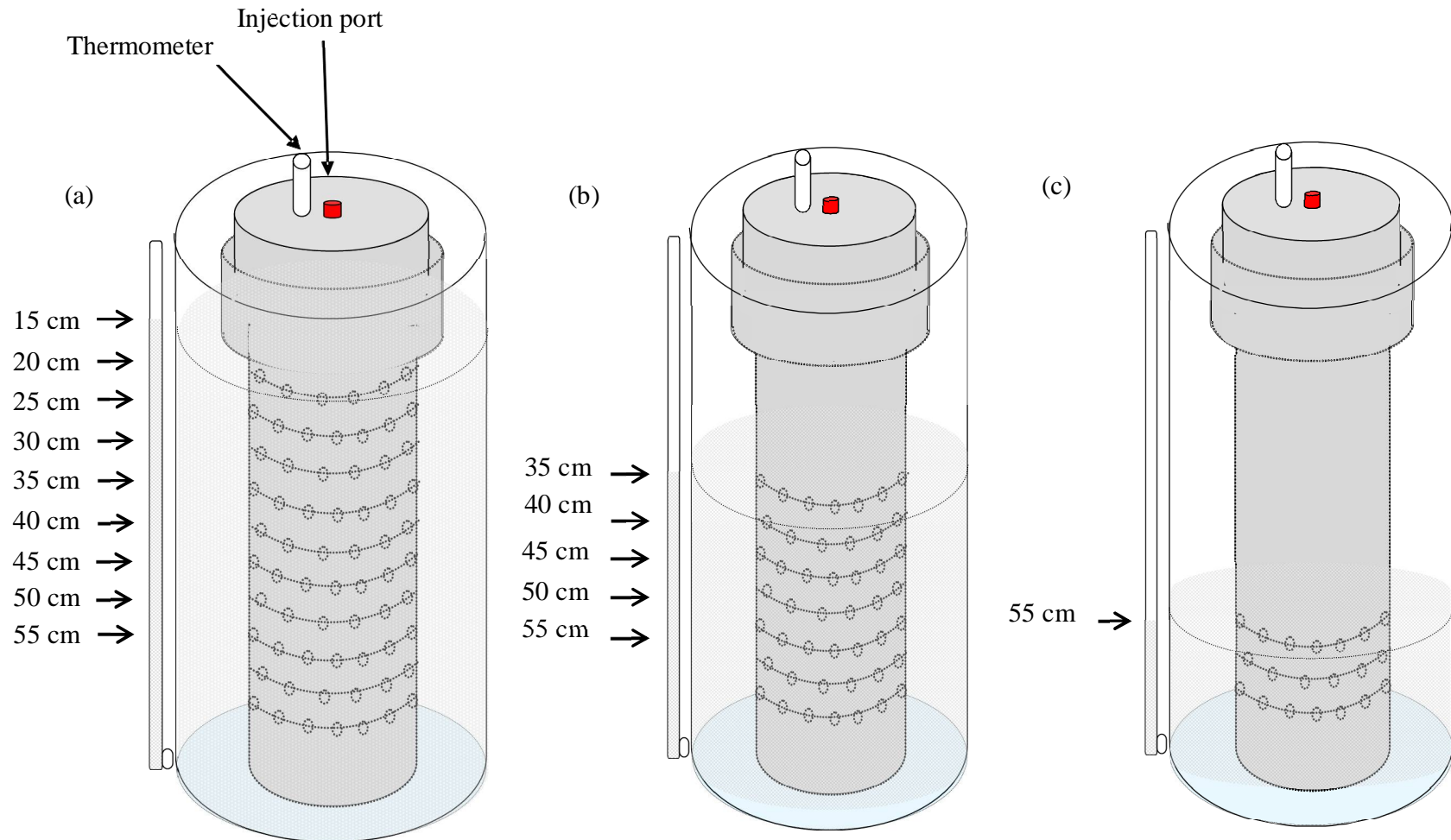
ha<sup>-1</sup> compost), replicated 3 times. According to Susilawati *et al.* (2015) that steel slag applications at 1 Mg ha<sup>-1</sup> and 2 Mg ha<sup>-1</sup> could decrease the CH<sub>4</sub> and N<sub>2</sub>O emissions from paddy rice field but the effect was not statistically significant, thus the rate of steel slag application was increased in this study. The experimental design in this study was adopted from Aerts and Ludwig (1997); Funk *et al.* (1994), and Jungkunst *et al.* (2008); these studies aimed to maintain water in peat soil columns at depths of 0-10 cm; 5-30 cm and 5-40 cm below soil surface for experimental periods of approximately 60, 130 and 141 days, respectively. The water depth treatments in the soil columns were initiated 11 days before the first gas sampling. Data were collected from April-July 2014.

#### **4.2.3. Biochar, steel slag and compost preparation and application**

The biochar used in this experiment was produced by pyrolysis of empty fruit bunches of oil palm. Specifically, the empty fruit bunches of oil palm were placed in pyrolysis reactor and burned at a temperature of approximately 250-300<sup>0</sup>C for 8 hours. Typically, a yield of 20-30% of biochar was achieved. The biochar was then ground to pass through a 2 mm stainless steel sieve and mixed thoroughly to obtain a fine granular consistency. The steel slag that was used in this experiment is a by-product of the steel industry and was obtained from West Java, Indonesia. The steel slag was ground and passed through a 2 mm sieve before use. The compost was a mixture of cow manure, filter cake of sugarcane, miller's bran and lime. The heap of organic materials was maintained under minimum oxygen conditions for 1 month. The heap was watered and mixed thoroughly twice a month. The chemical properties of the ameliorants are shown in Table 4.2. The ameliorants were thoroughly mixed with the water, and these mixtures were applied to the soil surface 1 day before the first gas sampling. The water and ameliorant mixture was used to ensure that the ameliorants were more effectively absorbed at and below soil surface.

**Table 4.2.** Chemical properties of ameliorants used in this experiment

| <b>Parameters</b>               | <b>Compost</b> | <b>Steel slag</b> | <b>Biochar</b> |
|---------------------------------|----------------|-------------------|----------------|
| pH (H <sub>2</sub> O)           | 7.66           | 8.24              | 9.96           |
| Organic C (g kg <sup>-1</sup> ) | 240.8          | 3.9               | 468.2          |
| Total N (g kg <sup>-1</sup> )   | 18.0           | 0.3               | 19.0           |
| Chemical composition (%)        |                |                   |                |
| P <sub>2</sub> O <sub>5</sub>   | 4.38           | 0.27              | 1.05           |
| K <sub>2</sub> O                | 0.29           | 0.07              | 0.22           |
| Al <sub>2</sub> O <sub>3</sub>  | 0.96           | 2.00              | 0.70           |
| SiO <sub>2</sub>                | 13.5           | 29.2              | 6.00           |
| MnO                             | 0.65           | 4.52              | 0.39           |



**Figure 4.1.** Sketch of the equipment used for water table treatments: 15 cm (a), 35 cm (b) and 55 cm(c).

#### 4.2.4. Gas measurement

Gas samples for CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O from the soil columns were measured simultaneously once a week. Gas samples were measured by a closed dark chamber method. The caps of soil column were used as the chambers. The height of the chamber was 20 cm. Each of the caps of the soil column was equipped with rubber septum for taking gas sampling and also equipped with thermometer for measuring temperature inside the chamber. Gas samples from the inner chamber were taken once a week using 10-cm<sup>3</sup> syringes and repeated the sampling 5 times every 5 minute (5, 10, 15, 20 and 25 minutes). Gas samplings were started from 06:00 in the morning on each sampling day. The caps were removed after gas sampling to allow the peat surface to be in contact with ambient air. Gas sample in the syringes were transferred directly to the laboratory and were determined by gas chromatography (GC). A GC is equipped with a thermal conductivity detector (TCD) for CO<sub>2</sub> analysis, a flame ionization detector (FID) for CH<sub>4</sub> analysis and an electron capture detector (ECD) for N<sub>2</sub>O analysis. The calculation of CH<sub>4</sub> or N<sub>2</sub>O fluxes were described already in previous chapter.

Other parameters were measured at the same time with gas sampling. The parameters were soil pH, Eh and soil-water temperature. Soil temperatures were measured, using a digital thermocouple. In each soil column, the redox electrode was permanently installed at a depth of about 5 cm below the peat surface. Redox potential and soil pH were measured weekly using a portable pH-millivolt meter.

#### 4.2.5. Statistical analysis

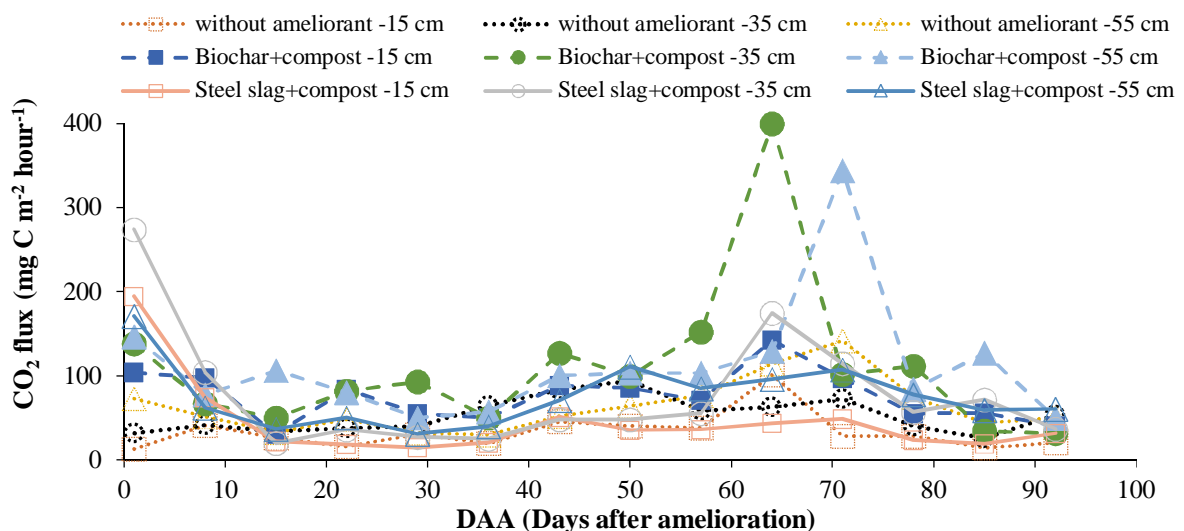
A two-way analysis of variance (water depths and ameliorations) followed Tukey's honestly significant difference (HSD) test was used to compare the mean values of CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O fluxes. The relationship between gas fluxes and each of the treatments, the treatments and the soil parameters were done by simple regression. Statistical considerations were based on  $P < 0.05$  and  $P < 0.001$  significance levels. Statistical analyses were conducted using SAS 9.1.3 portable (SAS Institute 2003).

### 4.3. Results and discussion

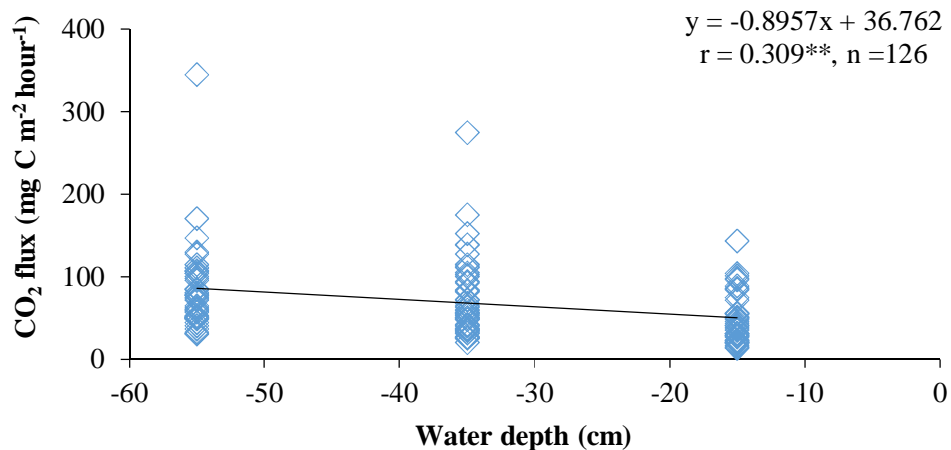
#### 4.3.1. Carbon dioxide emissions

The CO<sub>2</sub> fluxes patterns for the various treatments are shown in Figure 4.2. During the first week of the experiment, the CO<sub>2</sub> fluxes from the peat soils were high for all treatments except without ameliorant in different water depths. These responses likely occurred because of the effect of ameliorants on soil pH. High soil pH measured early on the experiment (Figure 4.11). From that time onwards, there was no difference in CO<sub>2</sub> fluxes between all the treatments until 57 days after ameliorations (DAA) and the ranged from 13 to 275 mg C m<sup>-2</sup> hour<sup>-1</sup>. The CO<sub>2</sub> fluxes from the biochar+compost treatment at water depths of 35 cm and 55 cm exhibited high peaks at 64 and 71 days after amelioration (DAA), approximately 400 and 344 mg C m<sup>-2</sup> hour<sup>-1</sup>, respectively, probably because the decomposition of organic matter was more rapid at lower water depth. After the high peaks and then declined at the end of the experiment. In this study, the mean daily CO<sub>2</sub> emissions ranged from approximately 0.80-2.68 g C m<sup>-2</sup> day<sup>-1</sup>. Moore and Dalva (1993) measured CO<sub>2</sub> emissions of approximately 0.17-3.80 g C m<sup>-2</sup> day<sup>-1</sup> in peat soil columns from subarctic fen, temperate bog and temperate swamp at temperatures of 10 and 22.6°C. According to Funk *et al.*

(1994), the CO<sub>2</sub> emissions from different water depth of taiga bog microcosms were approximately 0.8-3 g C m<sup>-2</sup> day<sup>-1</sup>. The Moore and Dalva (1993) and Funk *et al.* (1994) studies were also conducted using columns and the subsurface of the peat soil was used. CO<sub>2</sub> emissions from tropical peat soil are higher than those from temperate and boreal peat soil due to the temperature and moisture content, which influence the microbial processes leading to the production of these gases (Berglund *et al.* 2010). Based on a field study conducted in Jambi, Indonesia, Furukawa *et al.* (2005) reported that CO<sub>2</sub> emissions from tropical peatlands under different land-use management (coconut field, pineapple field and swamp forest) ranged from approximately around 0.72-6.38 g C m<sup>-2</sup> day<sup>-1</sup>. In addition, field studies conducted in secondary forest and paddy field in tropical peatlands, South Kalimantan resulted in CO<sub>2</sub> emissions of approximately around 3.36-6.62 g C m<sup>-2</sup> day<sup>-1</sup> (Hadi *et al.* 2005). The higher CO<sub>2</sub> emissions recorded in field studies compared with the soil columns likely occur because peat soil in the field have different substrates and microbial populations, which develop in response to long-term differences in water table position and soil managements (Moore and Dalva 1993).



**Figure 4.2.** The changes in CO<sub>2</sub> fluxes from 3 different water depths and ameliorations during the experimental period.



**Figure 4.3.** Relationship between water depths and CO<sub>2</sub> fluxes in peat soil (\*\* indicates  $P < 0.01$ )

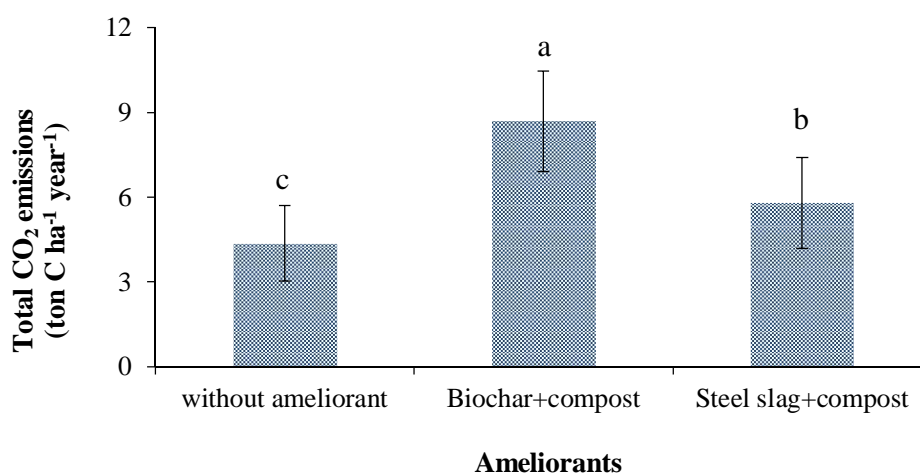
Figure 4.3 shows that there was relationship between CO<sub>2</sub> fluxes and the water depths in the three sets of columns ( $r = 0.309$ ,  $n = 126$ ,  $P < 0.01$ ). This result is similar to the finding reported by Funk *et al.* (1994); Moore and Knowles (1988) and Moore and Dalva (1993), i.e., that CO<sub>2</sub> emission from peat soil are related to water depth. In this study, the CO<sub>2</sub> emissions from peat soil at lowering the water depths (35 and 55 cm) were approximately 56 and 62% higher, respectively, than 15 cm water depth. According to Jungkunst *et al.* (2008), lowering of water depth by 20 and 40 cm from soil surface increased CO<sub>2</sub> emission from temperate forest (in columns) by 33 and 65%, respectively. This likely occurred because oxygen diffusion into soil increases when water depth is lowered; therefore soil become aerated, allowing aerobic decomposition (Silvola *et al.* 1996; Nykanen *et al.* 1998). Consequently, CO<sub>2</sub> fluxes from soils increase under aerobic conditions (Moore and Dalva 1993). This condition increased the soil Eh, and this condition is known to favor microbial activity and nitrogen mineralization (Ueda *et al.* 2000). However, Aerts and Ludwig (1997) found that CO<sub>2</sub> emission from mesotrophic peat with high water-table was higher than from peat with lower water-table probably because the peat layers



were not completely decomposed. Studies conducted by Lafleur *et al.* (2005) and Nieveen *et al.* (2005) indicate that the correlation between water depth and CO<sub>2</sub> emission is poor in temperate low shrub peatland.

The annual CO<sub>2</sub> emission from without ameliorant was approximately 4.4 ton C ha<sup>-1</sup> year<sup>-1</sup>, while the annual CO<sub>2</sub> emissions from steel slag+compost and biochar+compost treatments were approximately 5.8 and 8.7 ton C ha<sup>-1</sup> year<sup>-1</sup>, respectively (Figure 4.4). The CO<sub>2</sub> emission from control, slag+compost and biochar+compost treatments were approximately 50.2; 66.3 and 99.3 mg C m<sup>-2</sup> hour<sup>-1</sup>, respectively. Amelioration had a highly significant ( $P < 0.01$ ) effect on CO<sub>2</sub> emissions which were enhanced by the application of biochar+compost and steel slag+compost. Steel slag+compost and biochar+compost application to the peat soil stimulated CO<sub>2</sub> emissions by approximately 1.4 and 4.3 ton C ha<sup>-1</sup> year<sup>-1</sup>, respectively. The CO<sub>2</sub> emitted from the biochar+compost treatment was nearly 2 fold higher than from without ameliorant likely because of an increased availability of the media as microbial substrates and an increased microbial decomposition and mineralization of organic matter (Smith *et al.* 2010; Jones *et al.* 2011). Although CO<sub>2</sub> emission from biochar+compost and steel slag+compost were quite higher than without ameliorant but the emissions were slightly lower compare the studies in peat soil from ICCTF (2011) and Sakata *et al.* (2015) (Table 1.1). The application of steel biochar+compost and slag+compost increased CO<sub>2</sub> emission. This result contradict with the study from ICCTF (2011), application of biochar from rice husk could reduce CO<sub>2</sub> emission in peat soil because organic matter as an ameliorant in peat soil requires one to consider the quality and type of materials and level of maturity of the organic matter. According to Ali *et al.* (2008), the effect of steel slag fertilizer on CO<sub>2</sub> production exhibited an increasing trend, which is indirect evidence of methanotropic' activity. Another reason for the higher CO<sub>2</sub> emissions in response to amelioration

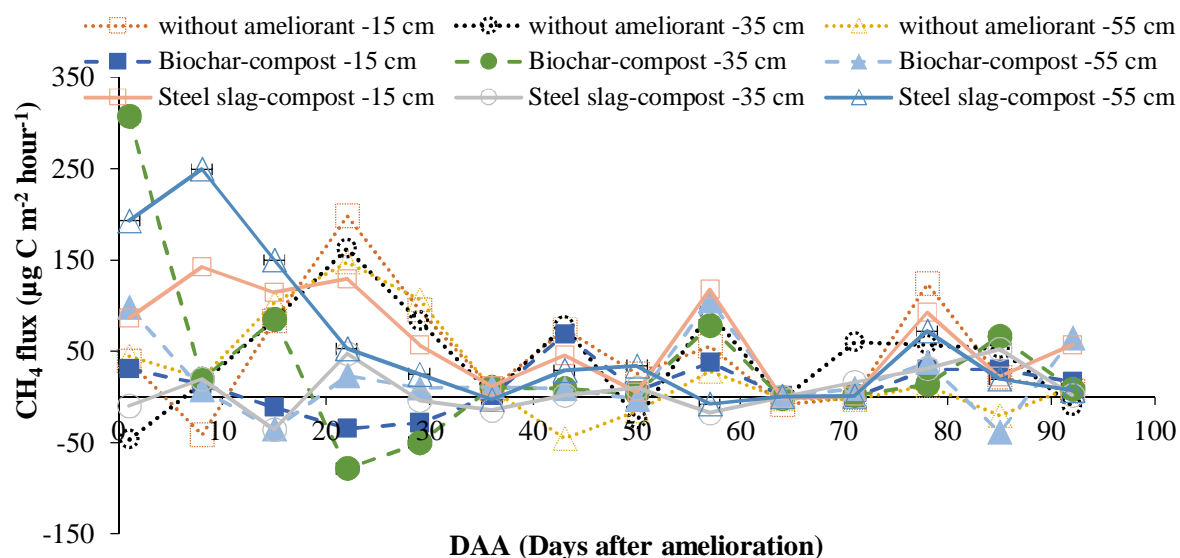
is the soil pH. The biochar+compost and slag+compost treatments increased the soil pH. This finding indicates that there is a decreasing CO<sub>2</sub> emissions trend for water depths closer to the peat surface and increasing CO<sub>2</sub> emissions trend when ameliorations are applied.



**Figure 4.4.** The CO<sub>2</sub> emission from different ameliorants at peat soil columns ( $P < 0.05$ )

#### 4.3.2. Methane emissions

In this study, the CH<sub>4</sub> fluxes were very sporadic and there was no pattern between the treatments was observed throughout the measurements periods (Figure 4.5). The coefficients of variation within treatments were higher than 45% and sometimes exceeded 100%. Consequently, CH<sub>4</sub> emissions were not significantly affected by the treatments. A very high variation in the CH<sub>4</sub> fluxes were observed between the treatments during the first 29 days of measurements. The highest CH<sub>4</sub> fluxes, i.e., approximately around 0.307 mg C m<sup>-2</sup> hour<sup>-1</sup>, were recorded in the biochar+compost treatment at a water depth of 35 cm after which a sharp decrease was observed. The CH<sub>4</sub> fluxes during the 92 days of measurements were rather low and sometimes negative. Negative values indicate net uptake from the atmosphere by the ecosystem.



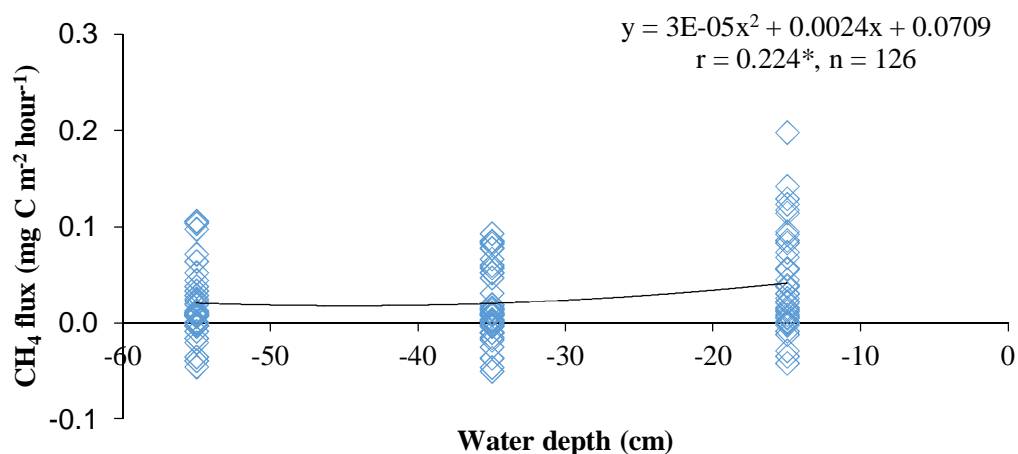
**Figure 4.5.** The changes in CH<sub>4</sub> fluxes from 3 different water depths and ameliorations during the experimental period.

The mean daily CH<sub>4</sub> emissions from all treatments are ranged from 0.17-1.51 mg C m<sup>-2</sup> day<sup>-1</sup>. The CH<sub>4</sub> emissions in this study were very low compared with similar experiments using peat soil columns, e.g., Moore and Dalva (1993). According this authors, the mean CH<sub>4</sub> emissions from subarctic fen, temperate bog and temperate swamp (using soil columns) ranged from 0.53-97.19 mg C m<sup>-2</sup> day<sup>-1</sup> and there was no significant relationship between water depth and CH<sub>4</sub> emissions from the swamp columns because of the small CH<sub>4</sub> emission values. The CH<sub>4</sub> emissions from taiga bog microcosms were approximately 0.02-19.16 mg C m<sup>-2</sup> day<sup>-1</sup> (Funk *et al.* 1994). The low CH<sub>4</sub> production in tropical peat soil likely occurred because most of the supply and the highest quality of decomposable organic matter are restricted to the peat surface and CH<sub>4</sub> oxidation by methanotrophic bacteria under oxic conditions may exceed gas production in a deeper anoxic peat profile (Brady 1997). Many tropical peatlands are covered by forest, in contrast to temperate peatlands which are commonly covered by sedges and moss (Andriess 1988). Woody tropical

peat contains higher levels of recalcitrant materials (e.g., lignin). Wood contain a higher lignin content with less decomposable C compared with cellulose. Williams and Yavitt (2010) reported that the biochemical compositions of lignin affect soil methanogenesis

Regression analysis of the CH<sub>4</sub> fluxes per hour and the water depths indicated a negative relationship ( $P < 0.05$ ) (Figure 4.6). According to this equation, i.e.,  $y = 0.00003x^2 - 0.0024x + 0.0709$ , the lowest CH<sub>4</sub> flux can be reached when the water depth is 40 cm below the soil surface. Methanogenesis may take place at a higher water depth causing higher emissions of CH<sub>4</sub> to the atmosphere. Roulet *et al.* (1992) and Nykanen *et al.* (1998) show that the critical water depth for high CH<sub>4</sub> emissions is approximately 10–20 cm. According to Wosten *et al.* (2006), CH<sub>4</sub> emissions are negligible at water depths more than 20 cm below the surface, and there is an increase in the CH<sub>4</sub> emission at water depths above 20 cm. There was no interaction effect between water depth and amelioration on the CH<sub>4</sub> fluxes, and there was no significant water depths and ameliorations effect on CH<sub>4</sub> emission, which was probably due to the high standard deviation within the treatments. Although there was no significant difference, a CH<sub>4</sub> emission reduction trend was observed as the water depth was lowered. The CH<sub>4</sub> emission measured at the 35 cm and 55 cm water depths were 32 and 12% lower, respectively, than those measured at the 15 cm water depth. According to Jungkunst *et al.* (2008), the CH<sub>4</sub> emissions from temperate forest in peat soil columns were reduced by approximately 3 and 8% at water depths of 20 cm and 40 cm, respectively, compared with 5 cm below the soil surface. The reduction in CH<sub>4</sub> emission due to the lowering of the water depth is probably related to the soil Eh. In our study, the Eh varied with the water depth and influenced the CH<sub>4</sub> emissions ( $r = 0.89$ ). According to Minamikawa and Sakai (2005), the Eh decreased the total CH<sub>4</sub> emission from paddy field in mineral soil by 36%.

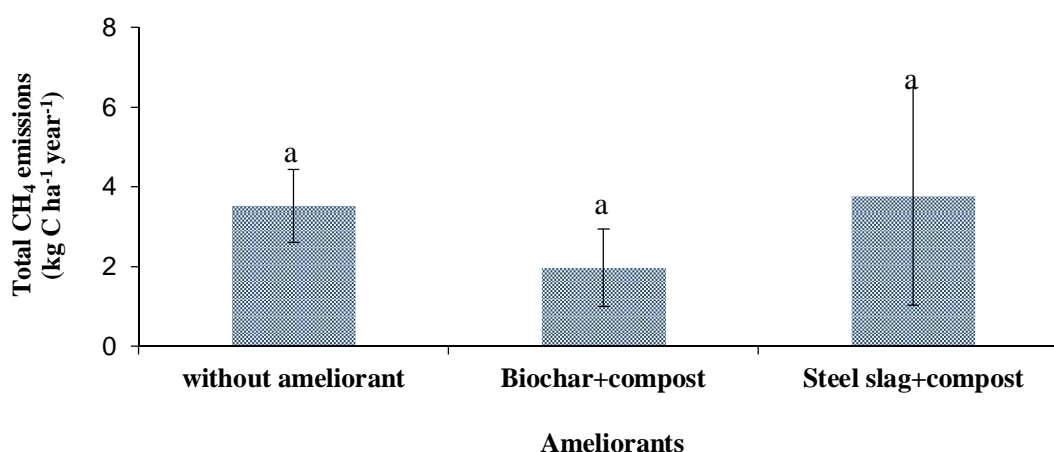
Generally, CH<sub>4</sub> emissions occur at values lower than -150 mV and increase with decreases in soil Eh (Wang *et al.* 1993). This study confirms that temperature controls CH<sub>4</sub> emission.



**Figure 4.6.** Relationship between water depths and CH<sub>4</sub> fluxes in peat soil (\* indicates  $P < 0.05$ )

The annual CH<sub>4</sub> emissions from without ameliorant, the biochar+compost and steel slag+compost applications were approximately 3.5, 1.9 and 3.8 ton C ha<sup>-1</sup> year<sup>-1</sup>, respectively (Figure 4.7). The CH<sub>4</sub> emission from without ameliorant, the biochar+compost and steel slag+compost applications were approximately 40.0; 21.7 and 43.3 mg C m<sup>-2</sup> hour<sup>-1</sup>, respectively. The CH<sub>4</sub> emission from this study is higher compare with those found by ICCTF (2011) approximately 9-12 mg C m<sup>-2</sup> hour<sup>-1</sup>. The application of biochar+compost reduced the total CH<sub>4</sub> emissions by approximately 44%. By contrast, the CH<sub>4</sub> emission were slightly increased (6.7%) by the application of steel slag+compost. The CH<sub>4</sub> emitted from this treatment was approximately 0.2 ton C ha<sup>-1</sup> year<sup>-1</sup> higher. The application of the biochar+compost reduced the CH<sub>4</sub> emissions in this study, similar to the biochar application results reported by Liu *et al.* (2011). The application of biochar+compost can reduce CH<sub>4</sub> emission probably because biochar provides better aeration, higher porosity, a larger surface area and makes the soil more favorable for methanotrophs

compared with soils without biochar (Karhu *et al.* 2011). On the other hand, the steel slag+compost application stimulated CH<sub>4</sub> emissions, most likely because the electron acceptor activity associated with steel slag was not sufficient to accept all of the electrons released from the reduction process due to the high organic matter content (Lee *et al.* 2012). Compost and peat soil have high SOM contents. A similar result was shown by Susilawati *et al.* (2015), i.e., no significant reduction in CH<sub>4</sub> emissions, likely because the low rate of steel slag applications did not provide sufficient electron acceptors. A mechanism used to decrease CH<sub>4</sub> emissions is the addition of electron acceptors such as iron materials, which influence the sequential soil Eh reactions. The electron acceptors are ordered according to their Eh, and the substrate is used at lower concentrations by electron acceptors with a higher Eh (Lovley and Phillips 1988). In this study, the application of the steel slag+compost increased the soil Eh compared with biochar+compost treatment and without ameliorant. Furukawa and Inubushi (2002) reported that CH<sub>4</sub> production activity was decreased with high revolving furnace slag application (20-100 Mg ha<sup>-1</sup>).



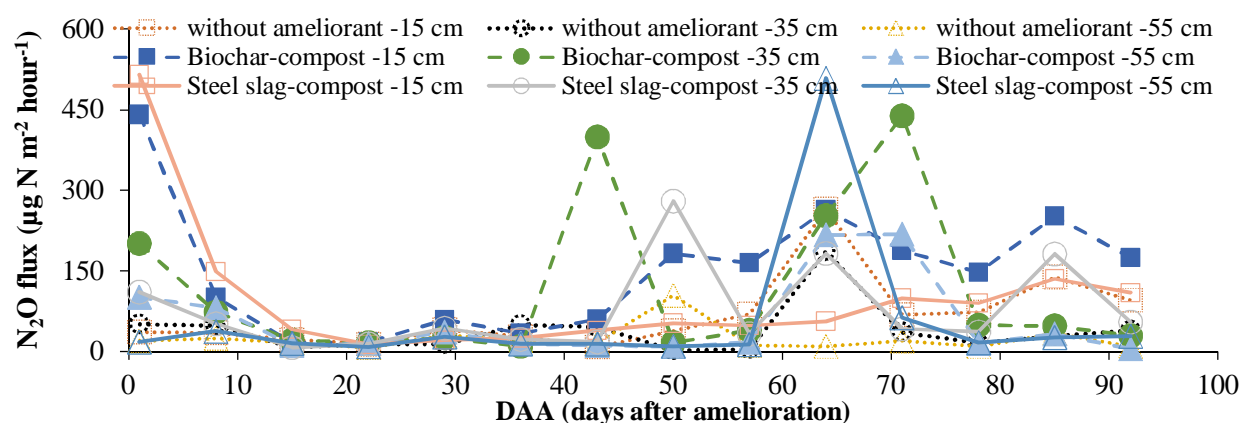
**Figure 4.7.** The CH<sub>4</sub> emission from different ameliorants at peat soil columns ( $P < 0.05$ )

### 3.3.3. Nitrous oxide emissions

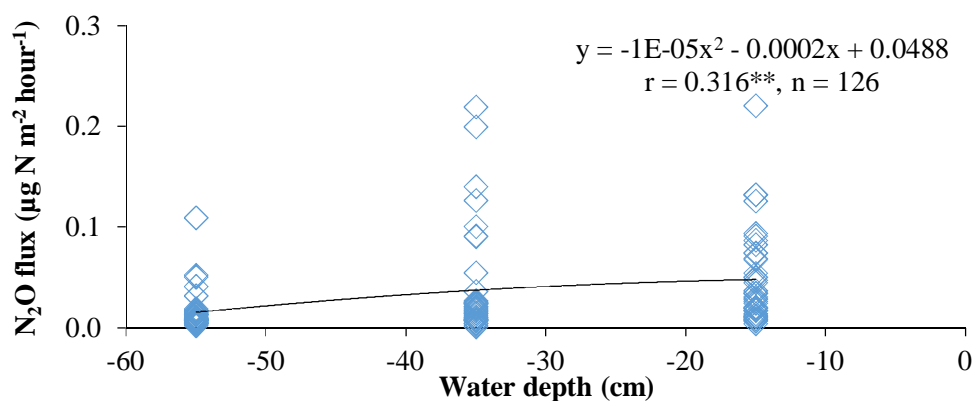
The dynamic changes in the N<sub>2</sub>O fluxes from peat soil columns are presented in Figure 4.8 for various water depths and ameliorants. The N<sub>2</sub>O fluxes varied within and between treatments. At the first measurement at 15 cm water depths, the N<sub>2</sub>O fluxes resulting from the application of steel slag+compost and biochar+compost were very high, likely because of the amelioration effect on soil pH and Eh. The first N<sub>2</sub>O emission measurements were high (similar to CO<sub>2</sub> emissions), probably due to the high soil pH and Eh that occurred early in experiment. In this study, amelioration significantly ( $P < 0.05$ ) affected the N<sub>2</sub>O emission. After the first time measurement conducted in these treatment, N<sub>2</sub>O fluxes at 35 and 55 cm water depths increased for next 5 measurements, i.e., until the end of the measurements. The fluxes from without ameliorant were mostly lower than those resulting from the application of ameliorants. At the first measurement, the N<sub>2</sub>O fluxes at 15 cm water depths in the application of steel slag+compost and biochar+compost treatments were very high. In this study, the N<sub>2</sub>O fluxes from all the treatments ranged from approximately 39-151  $\mu\text{g N m}^{-2} \text{ hour}^{-1}$ . According to Jungkunst *et al.* (2008), N<sub>2</sub>O fluxes from temperate forest were approximately around 20-963  $\mu\text{g N m}^{-2} \text{ hour}^{-1}$ , based on soil columns measurements at different water depths. In addition, N<sub>2</sub>O fluxes from tropical peat soil under secondary forest and paddy field on tropical peat soil in South Kalimantan were approximately 46 and 154  $\mu\text{g N m}^{-2} \text{ hour}^{-1}$ , respectively (Hadi *et al.* 2001).

Figure 4.9 shows that the N<sub>2</sub>O fluxes had a nonlinear, (a quadratic) relationship with water depth;  $y = -0.00001x^2 - 0.0002x + 0.0488$ . Based on this equation, the maximum N<sub>2</sub>O flux occurred when the water depth was 10 cm below the soil surface. This finding similar to that reported by Jungkunst *et al.* (2004) and Furukawa *et al.* (2005), i.e., that the peak N<sub>2</sub>O flux was observed at water depth of 20 cm below soil surface. Jungkunst *et al.* (2008) also observed a quadratic function

between water depths and N<sub>2</sub>O fluxes. This indicates that the N<sub>2</sub>O emitted during both nitrification and denitrification depends on the soil water content. The highest N<sub>2</sub>O emissions are found at intermediate water depths, which allows for both aerobic and anaerobic conditions (Davidson *et al.* 2000). Denitrification rates in soil generally depend on the O<sub>2</sub> concentration, the NO<sub>3</sub><sup>-</sup> concentration and the availability of easily metabolizable organic matter (Stepniewski and Stepniewska 1998). Lowering the water table generally increases the N<sub>2</sub>O emission rate; however, this relationship is rather complex (Regina *et al.* 1996).



**Figure 4.8.** The changes in N<sub>2</sub>O fluxes from 3 different water depths and ameliorations during the experimental period.

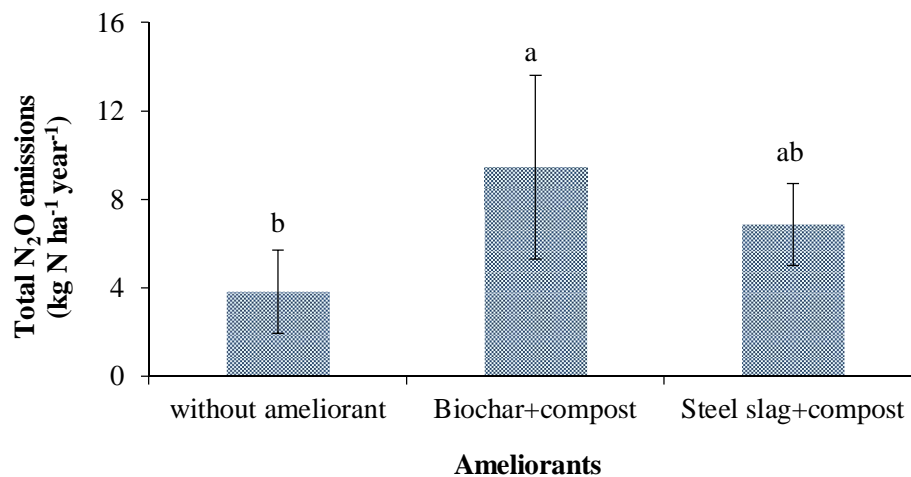


**Figure 4.9.** Relationship between water depths and N<sub>2</sub>O fluxes in peat soil (\*\* indicates  $P < 0.01$ )



The emissions of the  $N_2O$  from the biochar+compost and steel slag+compost treatments were higher than those from without ameliorant (Figure 4.10). The high  $N_2O$  emission from ameliorant treatments most likely occurred because of the increased availability of the nitrate substrates and the easily degradable organic matter for denitrification, which resulted from the compost application (Linn and Doran 1984; Dobbie *et al.* 1999). The availability of organic C is the main factor that influences denitrification under anaerobic conditions (Zou *et al.* 2005). Emissions of  $N_2O$  from the soil amended with biochar depend on the characteristics of the biochar, the addition of exogenous nitrogen and soil properties (Zhang *et al.* 2010). Singh *et al.* (2010) explained that the  $N_2O$  emission increased due to higher microbial activity and high labile N content of the biochar. In addition, soil pH increases in tropical acid soils supplied with composts derived from organic products results in (van der Watt *et al.*, 1991). The mechanism of increasing soil pH through the application of organic matter is not fully understood but likely occurs because of the specific adsorption of organic anions and the corresponding release of hydroxyl ions (Hue 1992). The higher  $N_2O$  emission from the steel slag+compost treatment is similar to the result with steel slag application result obtained by Huang *et al.* (2009); Liu *et al.* (2012) and Singla and Inubushi (2013). The  $N_2O$  emissions were significantly enhanced by the addition of iron. Iron can affect the speciation and mobility of organic and inorganic substances in soils and subsequently alleviate the immobilization of fertilizer N accompanying the decomposition of incorporated crop residue with a high C/N ratio, making more mineral N available for nitrification and denitrification. In addition, the soil pH was increased by the steel slag application. According to Ali *et al.* (2009), the alkaline pH of steel slag contributed to the increase in soil pH. The biochar+compost and steel slag+compost treatments increased of the soil pH. This finding showed that the ameliorations increased the soil pH. The mechanisms of  $N_2O$  production resulting from soil amelioration are

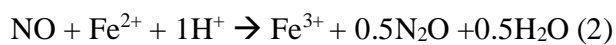
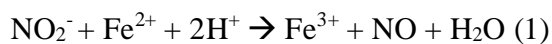
governed by nitrification, nitrifier denitrification and denitrification (Khalil *et al.* 2004). The highest annual N<sub>2</sub>O emission was measured in the biochar+compost treatment followed by the steel slag+compost treatment and without ameliorant, i.e., approximately 9.4; 6.9 and 3.8 ton N ha<sup>-1</sup> year<sup>-1</sup>, respectively (Figure 4.10). The application of Biochar+compost and steel slag+compost to peat soil significantly stimulated N<sub>2</sub>O emissions by approximately 5.6 and 3.0 ton N ha<sup>-1</sup> year<sup>-1</sup>, respectively. In this study, N<sub>2</sub>O emission from without ameliorant, biochar+compost and steel slag+compost were approximately 44.38; 107.31 and 78.77 mg N m<sup>-2</sup> hour<sup>-1</sup>, respectively. The range of N<sub>2</sub>O emission from this study was quite higher compare the result from Sakata *et al.* (2015), i.e., approximately 272 µg N m<sup>-2</sup> hour<sup>-1</sup> (Table 1.1). The difference range of N<sub>2</sub>O emission from these studies could be happened most likely because of many factors, e.g., rain, temperature, fertilization, irrigation, heavy metal accumulation, pH, organic matter content (Khalil *et al.* 2003), degree of peat maturity and experimental conditions.



**Figure 4.10.** The N<sub>2</sub>O emission from different ameliorants at peat soil columns ( $P < 0.05$ )

This study yielded different results from Susilawati *et al.* (2015). In that study, the N<sub>2</sub>O emissions were significantly reduced by steel slag application after second seasons (Table 4.3).

There was a statistical decrease in N<sub>2</sub>O emissions in the rice plantation from steel slag applications; the reduction was approximately 39–49%. Lower N<sub>2</sub>O emissions were found after steel slag application at an 8 Mg ha<sup>-1</sup> rate compared with the control according to Wang *et al.* (2015). The reduced N<sub>2</sub>O emissions can be caused by an increase in the iron (III) oxide concentration, suppressing microbe activities, including N<sub>2</sub>O production (Noubactep 2011). Steel slag is high in iron, which is an important oxidant for biological and chemical reactions that use oxidizing or reducing agents. Nitrite can be reduced by the presence of iron oxide at a near-neutral pH, and the end product is N<sub>2</sub>O and NO as an intermediate (Van Cleemput and Baert 1983; Van Cleemput 1998). The reduction of nitrite will affect the global production of nitric oxide (NO) and N<sub>2</sub>O. Kampschreur *et al.* (2011) described the chemical conversions as follows:



Abiotic denitrification with iron (II) can occur during nitrite accumulation. High nitrite concentrations increased nitrogen availability, especially when fertilizers are applied. During anoxic, an iron (II)/iron (III) reduction may induce biological and chemical nitrate and nitrite reduction to NO and N<sub>2</sub>O (Kampschreur *et al.* 2011). Biological iron oxidation with nitrate releases small amounts of N<sub>2</sub>O accumulation, and under strong anaerobic conditions, N<sub>2</sub>O is further reduced to N<sub>2</sub> as an end product (Nielsen and Nielsen 1998; Granli and Bockman 1994). These reactions indicate that when there are higher iron applications to soil or higher iron contents in soil, this element will influence N<sub>2</sub>O emissions. The response in terms of N<sub>2</sub>O emissions reduction depends on the physicochemical properties of the soil. The availability of iron in soils is influenced by the type of parent material and the land use in different farming systems.

**Table 4.3.** Seasonal N<sub>2</sub>O emissions at two different paddy field sites during rice growing seasons (Susilawati *et al.* 2015)

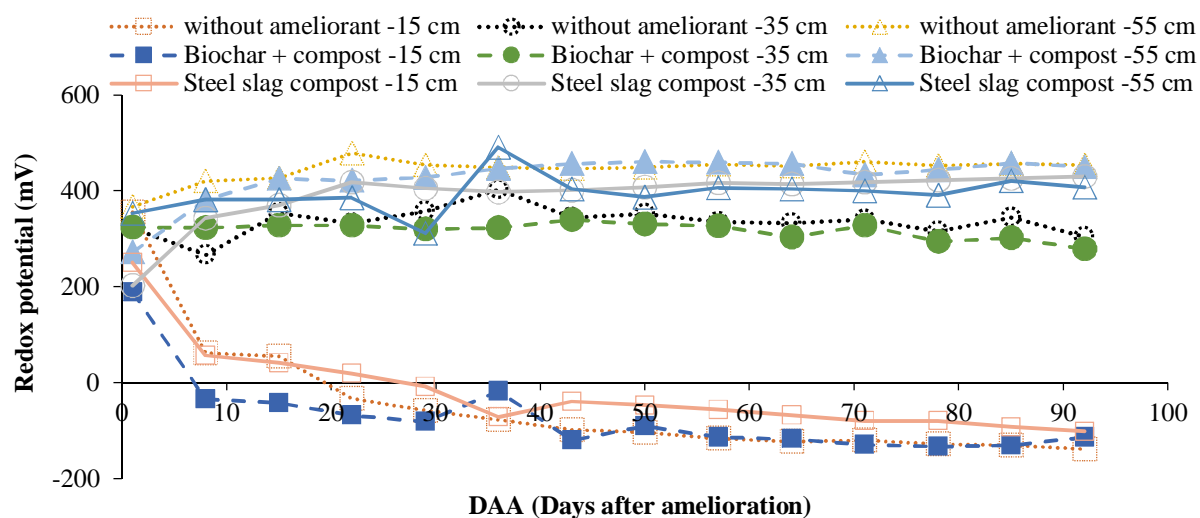
| Sites                        | Application of steel slag        | N <sub>2</sub> O                          |           |
|------------------------------|----------------------------------|---|-----------|
|                              |                                  | DS  | RS        |
|                              |                                  | g N ha <sup>-1</sup> season <sup>-1</sup> |           |
| Jakenan                      | Control                          | 38.75 a                                   | 19.31 bc  |
|                              | Steel slag 1 Mg ha <sup>-1</sup> | 30.57 a                                   | 13.84 c   |
|                              | Steel slag 2 Mg ha <sup>-1</sup> | -   | 12.78 c   |
| Wedarijaksa                  | Control                          | 45.99 a                                   | 46.03 a   |
|                              | Steel slag 1 Mg ha <sup>-1</sup> | 41.43 a                                   | 33.70 ab  |
|                              | Steel slag 2 Mg ha <sup>-1</sup> | -   | 28.37 abc |
| <b>ANOVA</b>                 |                                  |   |           |
| Sites                        |                                  | ns  | ***       |
| Application of steel slag    |                                  | ns  | *         |
| Sites*steel slag application |                                  | ns  | ns        |

#### 4.3.4. Redox potential (Eh) and pH

At the first measurement, the Eh from in all treatments had a positive value (Figure 4.11). The different treatments had distinct effects on the Eh starting from the second measurement until the end of the experiment. Highly reducing conditions developed at the 15 cm water depth in the three amelioration treatments. In the 15 cm water depth treatment, the first Eh value was positive and then gradually decreased. During the entire experimental period, the Eh values measured at the 15 cm water depth ranged from approximately +356 mV to -138 mV. At the lower water depths (35 and 55 cm), the Eh increased to more or less constant values ranging from +202

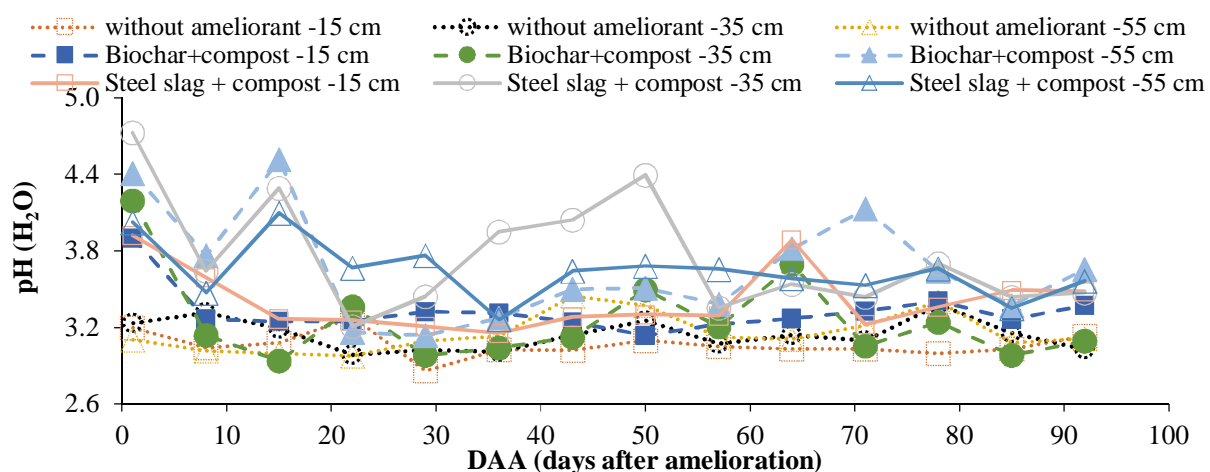
to +479 mV which were significantly higher than those measures at 15 cm water depths. The clear differences in Eh observed in this study likely occurred because of the lowering of water. There was a relationship between the Eh and water depth ( $r = 0.89$ ,  $n = 42$ ,  $P < 0.01$ ). On the other hand, under anaerobic conditions, oxygen in soil decreases and then the Eh gradually decreases as a result of the biochemical activity of numerous facultative and obligate anaerobes that use  $\text{NO}_3^-$ , Fe(III), Mn(IV) compounds and  $\text{SO}_4^{2-}$  as terminal electron acceptors. The production of  $\text{CO}_2$  decreases linearly with Eh (Włodarczyk *et al.* 2002).

The Eh was correlated with the three gases, i.e.,  $\text{CO}_2$ ,  $\text{CH}_4$ ,  $\text{N}_2\text{O}$  (Table 4.4). There was a positive correlation between the Eh and the  $\text{CO}_2$  fluxes in all treatments. The Eh and  $\text{CO}_2$  fluxes were positively and negatively correlated in without ameliorant and the biochar+compost treatment, respectively, at a water depth 55 cm. There were correlations between the Eh and  $\text{CH}_4$ ,  $\text{N}_2\text{O}$  fluxes in all treatments. In addition, there was a correlation between the Eh and the  $\text{N}_2\text{O}$  fluxes in without ameliorant at water depth of 55 cm.



**Figure 4.11.** The changes in redox potential from 3 different water depths and ameliorations during the experimental period.

At the first measurement, the soil pH in ameliorant treatments was higher than that measured in the treatment of without ameliorant for all water depths (Figure 4.12). Thereafter, the soil pH of without amelioration was consistently lower than that measured in the biochar+compost and steel slag compost treatments at all water depths. The soil pH values measured after amelioration increased slightly throughout the measurement period. The soil pH in without ameliorant ranged from approximately 2.98 to 3.40 for the water depths and that measures in the biochar+compost and steel slag compost treatments was approximately 2.94 to 4.41 and 3.21 to 4.73, respectively. Soil pH affects the dynamics of carbon. The intensity of soil respiration is closely related to the decomposition of soil organic carbon and soil (Silva *et al.* 2008). Moreover, the compost in this study contained lime. Lime is used in agriculture to increase soil pH (West and McBride 2005). Steel slag contains high iron content, which changed the pH to alkaline; therefore the increase in soil pH may have enhanced the activity of soil microorganisms, and accelerated the decomposition of organic matter.



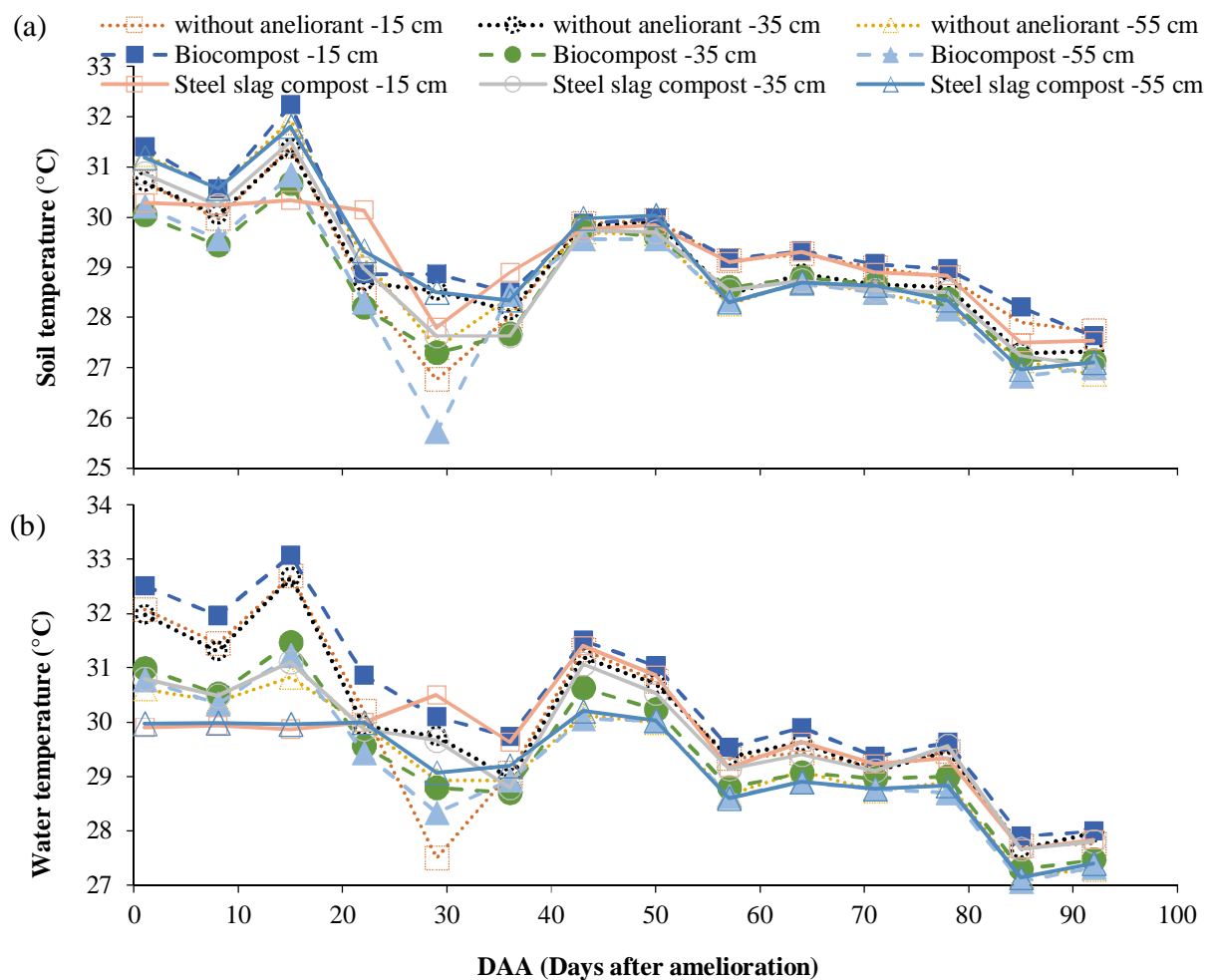
**Figure 4.12.** The changes in pH from 3 different water depths and ameliorations during the experimental period.

There were positive correlations between pH and the CO<sub>2</sub> and N<sub>2</sub>O fluxes with pH ( $P < 0.01$ ) (Table 4.4). Correlations were observed between pH and the CO<sub>2</sub> fluxes in the Biochar+compost treatment at 35 cm water depth and in without ameliorant at 55 cm water depth. Correlations were observed between pH and CH<sub>4</sub> fluxes in steel slag+compost treatment at 35 cm water depth and between pH and N<sub>2</sub>O fluxes in without ameliorant at 35 and 55 cm water depths.

#### 4.3.5. Soil and water temperature

There was a decreasing trend of soil and water temperature in all treatments over the experimental period (Figure 4.13a and 4.13b). The soil and water temperatures ranged from 25.7 - 32.2<sup>0</sup>C and 27.1 - 33.1<sup>0</sup>C, respectively. The average soil temperature was 29<sup>0</sup>C. At depths of 15, 35 and 55 cm, the average soil temperature was 29.2, 29.0 and 28.9<sup>0</sup>C and the average of water temperatures were 30.0, 29.8 and 29.2<sup>0</sup>C, respectively. The soil and water temperature pattern were similar across all treatments. The water and soil temperature decreased at 30 DAA. There was no significant difference between treatments. The soil temperature was strongly correlated with the CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O fluxes from all treatments (Table 4.4). According to Furukawa *et al.* 2005, soil temperature controls the biological reaction in the soil and then influences gas production. This study showed that lower water depths resulted lower in soil and water temperatures. An increase in soil temperature could lead to increased CO<sub>2</sub> and CH<sub>4</sub> fluxes from peatlands (Williams and Crawford 1984). In this study, there was a strong correlation between soil temperature and CO<sub>2</sub> emission ( $P < 0.01$ ) (Table 4.4). The soil and water temperatures were strongly correlated with CH<sub>4</sub> emission (Table 4.3). The correlation between temperature and CH<sub>4</sub> emission is more complicated than that for CO<sub>2</sub> emission due to the differing responses of CH<sub>4</sub> production and consumption processes in peat soil (Moore and Dalva 1993). Dunfield *et al.* (1993)

showed that CH<sub>4</sub> production and consumption reached an optimum at 25-30°C in temperate peat. There was a correlation between the soil temperature and the CH<sub>4</sub> fluxes in the steel slag+compost treatment at a depth of 55 cm. The water temperature was only correlated with the CH<sub>4</sub> fluxes (Table 4.4).



**Figure 4.13.** The changes in soil (a) and water (b) temperature from 3 different water depths and ameliorations during the experimental period.



**Table 4.4.** Correlation co-efficient (r) CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O fluxes with soil Eh, pH and temperature from each (n =14) and whole treatments

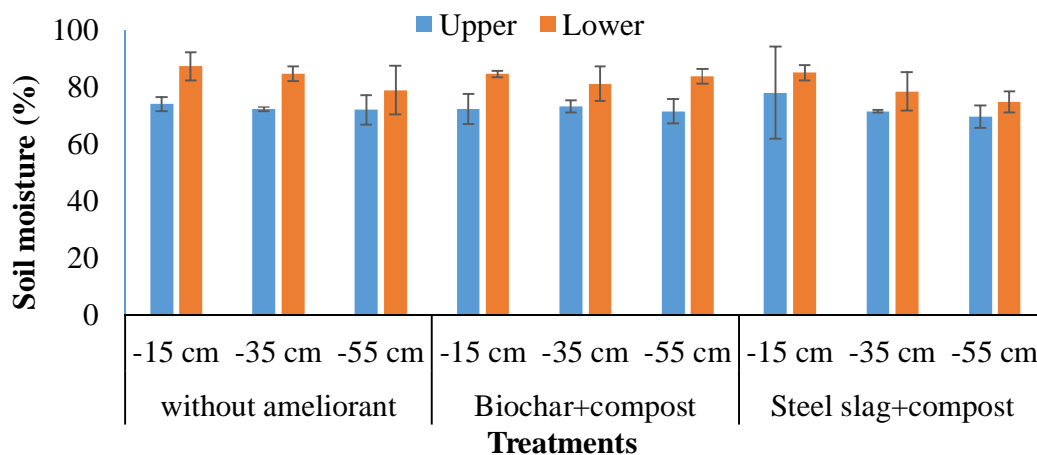
(n = 126)

| Parameters             | 15 cm              |                  |                     | 35 cm              |                  |                     | 55 cm              |                  |                     | Whole treatments |
|------------------------|--------------------|------------------|---------------------|--------------------|------------------|---------------------|--------------------|------------------|---------------------|------------------|
|                        | without ameliorant | Biochar+ compost | Steel slag+ compost | without ameliorant | Biochar+ compost | Steel slag+ compost | without ameliorant | Biochar+ compost | Steel slag+ compost |                  |
| <b>CO<sub>2</sub></b>  |                    |                  |                     |                    |                  |                     |                    |                  |                     |                  |
| Eh (mV)                | -0.285             | 0.280            | -0.013              | 0.143              | -0.042           | -0.108              | 0.828**            | -0.736**         | -0.168              | 0.47**           |
| pH                     | -0.290             | -0.081           | 0.255               | 0.142              | 0.538*           | 0.487               | 0.688**            | 0.359            | 0.287               | 0.66**           |
| Soil temperature (°C)  | 0.159              | 0.036            | -0.023              | 0.166              | 0.195            | 0.190               | 0.438              | 0.343            | 0.266               | 0.34**           |
| Water temperature (°C) | 0.049              | -0.009           | -0.049              | 0.202              | 0.084            | 0.039               | 0.147              | 0.185            | 0.167               | 0.12             |
| <b>CH<sub>4</sub></b>  |                    |                  |                     |                    |                  |                     |                    |                  |                     |                  |
| Eh (mV)                | 0.004              | 0.113            | 0.026               | -0.031             | 0.065            | -0.405              | 0.475              | 0.307            | -0.392              | -0.24*           |
| pH                     | 0.261              | -0.357           | -0.648**            | 0.177              | 0.592*           | 0.026               | 0.003              | -0.477           | 0.437               | 0.14             |
| Soil temperature (°C)  | -0.100             | -0.106           | 0.220               | 0.032              | 0.449            | -0.013              | 0.420              | -0.358           | 0.736**             | 0.46**           |
| Water temperature (°C) | 0.070              | -0.034           | 0.323               | -0.069             | 0.346            | -0.034              | 0.060              | -0.344           | 0.513               | 0.32**           |
| <b>N<sub>2</sub>O</b>  |                    |                  |                     |                    |                  |                     |                    |                  |                     |                  |
| Eh (mV)                | -0.343             | -0.175           | -0.019              | 0.413              | 0.282            | -0.201              | 0.760**            | -0.020           | 0.073               | -0.43**          |
| pH                     | -0.167             | 0.002            | 0.350               | 0.667**            | 0.186            | 0.431               | 0.702**            | 0.263            | -0.069              | 0.45**           |
| Soil temperature (°C)  | -0.195             | -0.039           | 0.093               | 0.083              | 0.314            | 0.099               | 0.212              | -0.023           | -0.086              | 0.28**           |
| Water temperature (°C) | -0.412             | -0.034           | 0.133               | -0.115             | 0.196            | 0.018               | -0.128             | -0.089           | -0.073              | 0.07             |

\* and \*\* denote significant at  $p < 0.05$  and  $0.01$ , respectively

#### 4.3.6. Soil moisture

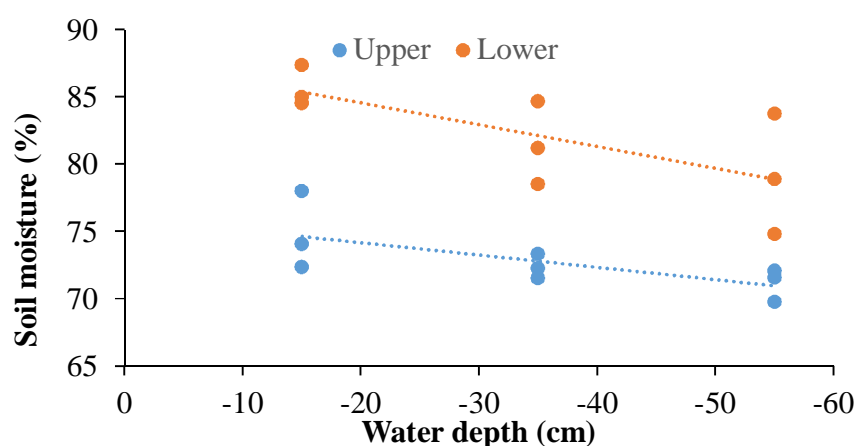
Soil samples to measure soil moisture were taken from 2 different depth, i.e., upper layer with the depth 0-10 cm from soil surface and lower layer with the depth 40-50 cm from soil surface. In all the treatments showed lower soil moisture in upper layer than in lower layer (Figure 4.14.). In upper layer, soil moisture were approximately around 70-74%, while in lower layer, soil moisture ranged approximately 75-87%. The lowest soil moisture in both of soil depth was found in the -55 cm water depth below soil surface, followed by -35 cm and -15 cm water depth below soil surface. Soil moisture effect the diffusion of soluble substrates at lower soil water content whilst at higher soil moistures diffusion of oxygen can become constrained; both of which are limiting to soil microbial respiration (Skopp *et al.* 1990).



**Figure 4.14.** The soil moisture from 2 different soil depth under 3 different water depths and ameliorations during the experimental period.

There were linear correlation between soil moisture and water depth in different of soil layer (Figure 4.15.). Deeper water depth resulted lower soil moisture. Drainage create aerobic

conditions and increase in the redox potentials are the condition that favour microbial activity and nitrogen mineralization (Ueda *et al.* 2000, Jali 2004). Lowering water depth increase peat soil aeration, then optimising microbial oxidation of organic matter to release of CO<sub>2</sub> to the atmosphere and reduce CH<sub>4</sub> emission. CH<sub>4</sub> is a characteristic product of organic matter breakdown under anaerobic peat, and gas production is highest when the water depth is near or at the peat surface and less decomposed litter becomes available for anaerobic decomposers (Conrad 1989).



**Figure 4.15.** Correlation between soil moisture from 2 different soil depth and water depth

#### 4.3.7. Total GHG emissions

There was no interaction effect between water depth and amelioration on the CO<sub>2</sub> emissions (Table 4.5). The effect of each treatment on CO<sub>2</sub> emissions was independent. The CO<sub>2</sub> emissions were significantly ( $P < 0.05$ ) and highly significantly ( $P < 0.01$ ) affected by water depth and amelioration, respectively. The annual emissions of CO<sub>2</sub> eq from all treatments ranged from 10.7-35.8 ton CO<sub>2</sub> eq ha<sup>-1</sup> year<sup>-1</sup>. There was no interaction effect between water depth and amelioration on the CH<sub>4</sub> emissions. Moreover, there was no significant water depth and amelioration effect on the CH<sub>4</sub> emissions. The annual CH<sub>4</sub> emissions from all treatments ranged

from 0.04-0.38 ton CO<sub>2</sub> eq ha<sup>-1</sup> year<sup>-1</sup>, respectively. There was no interaction effect between water depth and amelioration on N<sub>2</sub>O emissions. There was a significant ameliorations effect on the N<sub>2</sub>O fluxes, but no significant water depths effect. The annual N<sub>2</sub>O emissions from all treatments ranged from 1.01-6.15 ton CO<sub>2</sub> eq ha<sup>-1</sup> year<sup>-1</sup>, respectively. Amelioration significantly increased the total N<sub>2</sub>O emissions.

There was no interaction effect between water depth and amelioration on the total GHG emissions. In addition, there was no significant water depths effect on the total GHG emissions from the peat soil columns, but there was a significant amelioration effect. The total GHG emissions from the different water depths and soil ameliorants added to peat soil were ranged from approximately 13.6-40.2 ton CO<sub>2</sub> eq ha<sup>-1</sup> year<sup>-1</sup>. The total GHG emission from without ameliorant was approximately around 18.1 ton CO<sub>2</sub> eq ha<sup>-1</sup> year<sup>-1</sup> and those from Biochar+compost and steel slag+compost treatments were approximately around 36.4 and 24.8 ton CO<sub>2</sub> eq ha<sup>-1</sup> year<sup>-1</sup>, respectively. The total GHG emissions from biochar+compost treatment were almost double than those from without ameliorant. The biochar+compost and steel slag+compost application significantly stimulated total GHG emissions from peat soil by approximately 18.3 and 6.7 ton CO<sub>2</sub> eq ha<sup>-1</sup> year<sup>-1</sup>, respectively. The contribution of CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O to the total GHG emissions from the peat soil column were approximately 87; 0.5 and 12.6%, respectively. Total GHG emissions expressed as CO<sub>2</sub> eq facilitate access to the most acceptable technologies for reducing GHG emissions without having to separate each gas. In this study, ameliorations added to peat soil stimulated the total GHG emissions compared the treatment without ameliorants. Therefore, adding ameliorants to peat soil to enhance soil fertility should be more considered due to their effect on GHG emissions.

**Table 4.5.** CO<sub>2</sub>, CH<sub>4</sub>, N<sub>2</sub>O and total GHG emissions from different water depths and ameliorants at peat soil columns

| Ameliorants             | Treatments<br>Water<br>depth (cm) | CO <sub>2</sub> emission                   |   | CH <sub>4</sub> emission                   |   | N <sub>2</sub> O emission                  |   | Total GHG<br>emissions<br>ton CO <sub>2</sub> eq ha <sup>-1</sup><br>year <sup>-1</sup> |
|-------------------------|-----------------------------------|--|---|--|---|--|---|---|
|                         |                                   | mg C m <sup>-2</sup><br>hour <sup>-1</sup> | ton CO <sub>2</sub> eq ha <sup>-1</sup><br>year <sup>-1</sup> | mg C m <sup>-2</sup><br>hour <sup>-1</sup> | ton CO <sub>2</sub> eq<br>ha <sup>-1</sup> year <sup>-1</sup> | µg N m <sup>-2</sup><br>hour <sup>-1</sup> | ton CO <sub>2</sub> eq<br>ha <sup>-1</sup> year <sup>-1</sup> |   |
| without<br>ameliorant   | 15                                | 33 b                                       | 10.72 b   | 0.049 a                                    | 0.29 a  | 67 a                                       | 2.74 a  | 13.6 c  |
|                         | 35                                | 53 ab                                      | 17.06 ab  | 0.043 a                                    | 0.26 a  | 39 a                                       | 1.60 a  | 18.8 bc   |
|                         | 55                                | 64 ab                                      | 20.43 ab  | 0.029 a                                    | 0.17 a  | 25 a                                       | 1.01 a  | 21.5 bc   |
| Biochar +<br>compost    | 15                                | 76 ab                                      | 24.33 ab  | 0.012 a                                    | 0.07 a  | 151 a                                      | 6.15 a  | 30.5 ab   |
|                         | 35                                | 110 a                                      | 35.33 a   | 0.034 a                                    | 0.20 a  | 116 a                                      | 4.74 a  | 40.2 a  |
|                         | 55                                | 112 a                                      | 35.87 a   | 0.022 a                                    | 0.13 a  | 57 a                                       | 2.32 a  | 38.3 a  |
| Steel slag +<br>compost | 15                                | 45 ab                                      | 14.52 ab  | 0.063 a                                    | 0.38 a  | 100 a                                      | 4.08 a  | 18.8 bc   |
|                         | 35                                | 78 ab                                      | 25.08 ab  | 0.007 a                                    | 0.04 a  | 77 a                                       | 3.16 a  | 28.3 abc  |
|                         | 55                                | 76 ab                                      | 24.31 ab  | 0.059 a                                    | 0.35 a  | 58 a                                       | 2.35 a  | 26.8 abc  |
| <b>ANOVA</b>            |                                   |  |   |  |   |  |   |   |
|                         | Amelioration                      | **   | **  | ns   | ns  | *  | *   | **  |
|                         | Water depth                       | *  | *   | ns   | ns  | ns   | ns  | ns  |
|                         | Interaction                       | ns   | ns  | ns   | ns  | ns   | ns  | ns  |

$P < 0.05$ ,  $**P < 0.01$ , ns = not significant. Values in each column are means of three replicates. Different letters in the same column indicate significant differences between means at  $P = 0.05$  according to Tukey's HSD test.

#### **4.3.8. Changes in carbon stock and net carbon budget associated with water depths and ameliorations**

The carbon budget of agroecosystems is important in the global terrestrial C cycle (Pan *et al.* 2004). Increasing agricultural soil C stocks has been suggested as an important potential measure to sequester CO<sub>2</sub> from the atmosphere (Paustian *et al.* 1998). Soils are the second largest terrestrial carbon (C) reservoir (Jacinthe *et al.* 2002). The net carbon budget is typically estimated from soil organic carbon (SOC) measurements (Mosier *et al.* 2005; Robertson *et al.* 2000; Shang *et al.* 2010).

There was no difference among the treatments on C stock before and after the treatments. The highest C stock after the treatments was found in biochar+compost application, followed by steel slag+compost and without ameliorant and application approximately 1.26, 1.23 and 1.19 kg C column<sup>-1</sup>, respectively. Based on the water depth, the highest C stock after the treatments was found in -35 cm below soil surface and followed by -15 cm and -55 cm below soil surface approximately 1.39, 1.30 and 1.00 kg C column<sup>-1</sup>. C stock after the treatments was lower compare than before the treatments likely occur because there was peat decomposition resulted CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O emission (Table 4.5). Soil carbon stocks (per unit area) are estimated as the product of carbon concentration (% C), bulk density (g cm<sup>-3</sup>), and soil volume (m<sup>-3</sup>) (Warren *et al.* 2012). These properties cannot be measured directly from satellite or airborne sensors, and therefore rely on intensive field sampling for data acquisition. Bulk density and C concentration can vary spatially and throughout the vertical peat profile (Page *et al.* 2004). Therefore, in the future study multiple measurements of carbon concentration and bulk density from samples taken

at various depths in the soil profile are needed to accurately determine the soil carbon stocks (Donato *et al.* 2011; Kauffman *et al.* 2011; Murdiyarso *et al.* 2010).

In all the treatments, C stock was higher than total GHG emission (Table 4.6). The net CO<sub>2</sub>-C exchange between the atmosphere and terrestrial were approximately 0.76-1.23 kg CO<sub>2</sub>-C column<sup>-1</sup> 92 days<sup>-1</sup>. There was no significant difference on net carbon among the treatments. Net CO<sub>2</sub> exchange between the atmosphere and terrestrial systems represents the balance between C inputs by autotrophic fixation and outputs by heterotrophic oxidation of organic material. In this study, net carbon was determined from changes in topsoil organic C and total CO<sub>2</sub>-C fluxes.

**Table 4.6.** C stock, total GHG emission and net carbon under different water depths and ameliorations

| Treatments          |       | C stock kg (C column-1) |                  |           | Total GHG emission   | Net Carbon   |
|---------------------|-------|-------------------------|------------------|-----------|--|--|
|                     |       | Before treatments       | After treatments | Δ c stock | (kg CO <sub>2</sub> -C column <sup>-1</sup> 92 day <sup>-1</sup> ) | (kg CO <sub>2</sub> -C column <sup>-1</sup> 92 day <sup>-1</sup> ) |
| without ameliorant  | 15 cm | 1.75                    | 1.26             | -0.49     | 0.10   | 1.15   |
|                     | 35 cm | 1.80                    | 1.37             | -0.43     | 0.14   | 1.23   |
|                     | 55 cm | 1.76                    | 0.96             | -0.80     | 0.15   | 0.81   |
| Biochar+ compost    | 15 cm | 1.87                    | 1.35             | -0.52     | 0.23   | 1.12   |
|                     | 35 cm | 1.85                    | 1.41             | -0.45     | 0.29   | 1.11   |
|                     | 55 cm | 1.88                    | 1.03             | -0.85     | 0.27   | 0.76   |
| Steel slag+ compost | 15 cm | 1.80                    | 1.29             | -0.51     | 0.14   | 1.15   |
|                     | 35 cm | 1.84                    | 1.41             | -0.44     | 0.21   | 1.20   |
|                     | 55 cm | 1.82                    | 1.00             | -0.82     | 0.19   | 0.81   |

#### 4.4. Conclusions

This study quantified the effects of water depth and soil amelioration on greenhouse gas emissions from peat columns. Less CO<sub>2</sub> was emitted lower when the water depth was near the soil surface. Conversely, a deeper water depth resulted in a slight decrease in the CH<sub>4</sub> emissions. However, the highest N<sub>2</sub>O emissions were measured at intermediate water depths. The soil pH, Eh and temperature were associated with the three gases (CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O) emitted from the tropical peat soil. The biochar+compost and steel slag+compost treatments increased the CO<sub>2</sub> and N<sub>2</sub>O emissions from the peat soil columns. Long-term experiments should be developed to monitor changes that occur over time in response to amelioration at various water depths. It might be better to apply ameliorants at higher rates to reach a sustainable reduction in GHG emissions but it should consider the applicability to be used by the farmer. These experiments should be conducted in peat soil and peat soil-crop systems in peat soil to determine how GHG emissions from these types of treatments can be reduced.



## CHAPTER 5

### GENERAL DISCUSSION AND CONCLUSIONS

#### 5.1 General discussion

Addressing the global challenges of climate change, food security, and the increase of human population requires enhancing the adaptive capacity and mitigation potential of agricultural sector. The agricultural sector is one of the sectors that currently responsible for global GHG emissions and is also a key driver of deforestation due to food demand. Global warming continues to dominate the world's science and policy agenda on global change. One fundamental concern is the impact of this climate change on water supply. Rice production requires large amounts of water to produce more rice. On the other hand, in the future was expected that rice production will face water competition and water scarcity. To reduce water use in irrigated lowland rice, water-saving regimes can be introduced, such as alternate wetting and drying (AWD), intermittent irrigation, mid-season drainage. Reducing water from rice field can save the water as well as reduce environmental burden such as CH<sub>4</sub> emission, but it sometimes has side effects such as increasing N<sub>2</sub>O emissions. The trade-off relationship of N<sub>2</sub>O and CH<sub>4</sub> production in rice soils makes it a real challenge to reduce the production of one gas but not to increase the production of the other. A better understanding of this relationship is needed in order to be able to possibly mitigate the emission of these important greenhouse gases through changes in agricultural practices. The approaches to save the water and minimize the environmental burden should allow rice production to be maintained or increased because there is high demand on food crop as staple food.

There are important factor in soil that control CH<sub>4</sub> and N<sub>2</sub>O production from rice fields. The content of soil oxidants used as electron acceptors for organic matter degradation contributes

significantly to CH<sub>4</sub> and N<sub>2</sub>O production. The reduction of various oxidants in homogeneous soil suspensions occurs sequentially at corresponding soil redox potential values (Ponnamporuma 1972). Fluctuation of soil water content determines soil aerobic and anaerobic conditions, which can be characterized by Eh. In flooded rice soil, there are two aerobic/anaerobic interfaces—flooded soil surface layer maintained by O<sub>2</sub> dissolved in the standing water, and plant rhizosphere maintained by O<sub>2</sub> diffusing through the rice plant. These aerobic/anaerobic interfaces control many redox reactions including nitrification, denitrification, cycling of iron and manganese compounds, sulphate reduction and sulphide oxidation, and CH<sub>4</sub> formation and oxidation.

Simultaneous mitigation options are different for CH<sub>4</sub> and N<sub>2</sub>O emission since the production of these two gases take place under contrasting conditions, thus minimizing one gas should not increasing the emission of the other. The trade-off both of the emission should be well prepared for a balanced set of mitigation options, which optimize the emission trade-off in minimum cumulative radiative forcing of the two gases on GWP, thus having a lowest possible greenhouse effect. Reducing water from field not only could mitigate GHG emission but also could save the water. This study showed that the practice of AWD and site specific of AWD (S-AWD) produce no significant yield difference to CF and save the water. Water-saving techniques can reduce GHG emissions in a given area of rice land, but in most cases, the water saved will then be used to irrigate more rice land or new crops in future seasons. Subsequently, emission savings are offset by emissions created on newly irrigated land. It means that rice production will increase. In the community and as a social concern, the relationship among water users improved, especially within an irrigation unit, because water had become available not only upstream but also downstream.

Growing rice in continuously flooded fields has been taken for granted for centuries, but water crisis may change the way rice is produced in the future. Technologies that save water for rice and increase productivity of a post-rice crop will be more acceptable to farmers. Assuming that AWD or S-AWD have been successfully introduced by a significant number of farmers to guide the irrigation of their rice crop, there is still the issue of providing mechanisms for the continued spread of AWD or S-AWD to be accepted on a larger scale in Indonesia. One difficulty in communicating about water savings itself is that this term carries different meanings to different people. The meaning is often dependent on the scales of interest: frequency, timing and volume of application, field preparation to control percolation and seepage and to capture rain, fertilizer use, pest control and more. Spread out the technology to the farmer is need more effort since implementing AWD or S-AWD require a coordinated approach for water scheduling in irrigation areas. Irrigation in each regions is managed in many different ways. In some cases, farmers are involved in decision-making processes on water scheduling, farmers and the pump owner discuss the irrigation schedule and payment arrangement before the onset of the irrigation season. In other cases, farmers who have their own water pump, are not formally organized and often independently implements the entire irrigation schedule by themselves. Thus, farmers with a bigger holding seem to be able to run the risk of losing some yield by trying AWD, who own a pump, greater economic stability and are able to control irrigation as prescribed by AWD.

Bouman and Tuong (2001) summarized 31 field experiments on AWD and they found the yield reductions of 0–70% in AWD treatments compared with continuously flooded controls in 92% of the experiments. To maintain or even increase the rice production. Another approaches could be done such as soil amelioration. Soil amelioration is the way to improve the soil quality to support the live of the plant and to enhance the production. Steel slag could be used as an oxidizing

agent to suppress CH<sub>4</sub> emissions from rice fields. Electron acceptors such as NO<sub>3</sub><sup>-</sup>, Mn<sup>4+</sup>, Fe<sup>3+</sup> and SO<sub>4</sub><sup>2-</sup> can decrease CH<sub>4</sub> production because of inhibitory and competitive effects with different microorganisms for common electron donors (Jakobsen *et al.* 1981; Achtnich *et al.* 1995). Biochar, charcoal from biomass that has been pyrolysed in a zero or low oxygen environment, owing to its inherent properties, scientific consensus exists that application to soil at a specific site is expected to sustainably sequester carbon and concurrently improve soil functions. A mechanism of using soil amelioration used to decrease CH<sub>4</sub> emissions is the addition of electron acceptors, which influence the sequential soil Eh reactions. The electron acceptors are ordered according to their Eh, and the substrate is used at lower concentrations by electron acceptors with a higher Eh (Lovley and Phillips 1988). In this study, application of soil amelioration in mineral soil, i.e., steel slag application could slightly reduce CH<sub>4</sub> and N<sub>2</sub>O emission in mineral soil although no significance difference. However, steel slag application as well as biochar and compost application in peat soil have tendency to stimulate GHG emission. Application of ameliorant in peat soil enhance GHG emission most likely because the electron acceptor activity associated with ameliorants were not sufficient to accept all of the electrons released from the reduction process due to the high organic matter content (Lee *et al.* 2012). Thus, it might be better to apply ameliorants at higher rates to reach a sustainable reduction in GHG emissions but it should consider the applicability to be used by the farmer. There is an urgent need for further experimental research with regard to long-term effects of soil amelioration on soil functions, as well as on the behaviour and fate in different soil types and under different management practices.

Recently, the peat-land utilization in Indonesia for agriculture has received much attention although the peat mostly contains of poor to very poor in nutrients for plant growth and organic materials (Sabiham 1988). Initially, all these peat deposits were covered with pristine peat

swamp forest but, as a result of economic development, the peatlands have been subjected to intensive logging, drainage and conversion to plantation estates especially in Sumatra and Kalimantan (Rieley *et al.* 1996; Rieley and Page 2005). Mega Rice Project (MRP) initiated in 1995, disrupted the peatland ecosystem over an area of more than 1 million ha. MRP was constructed canals up to 30 m wide and length of approximately 4500 km. After the drainage, peatlands become susceptible to fire. Fires are most severe during El Niño periods, as in 1997/1998 when about 2.4–6.8 million ha of peatlands burnt in Indonesia (Page *et al.* 2002) after that El Niño reoccurred and burnt peatlands in Indonesia in 2002, 2006 and 2015. Furthermore, peatlands burnt once are more likely to burn again (Siegert *et al.* 2001; Cochrane 2003; Langner *et al.* 2007).

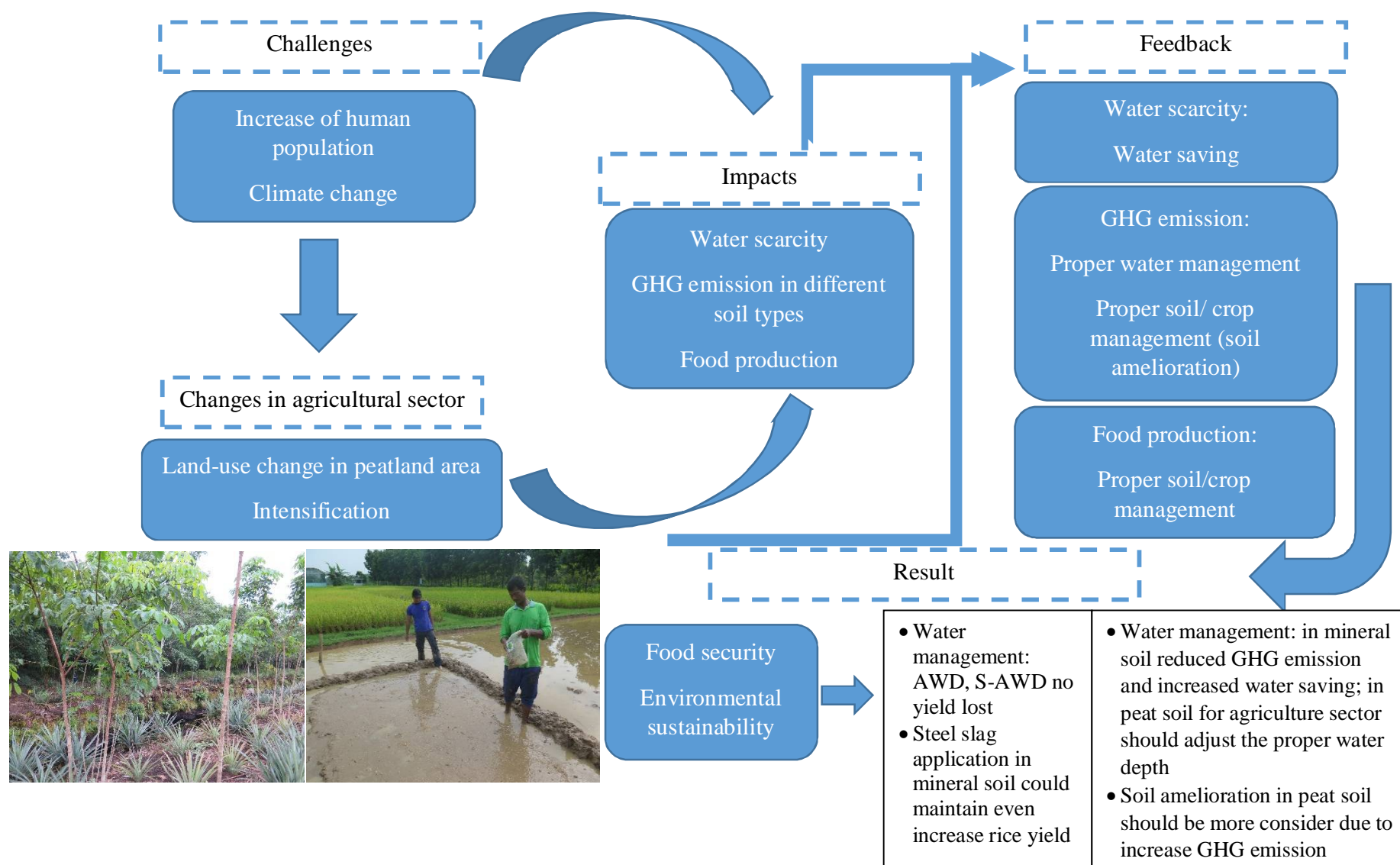
Expansion of agricultural land is widely recognized as one of the most significant human alterations to the global environment. On the other hand, there is increased competition for land, water, energy, and other inputs into food production. Consequently, the other way to meet increasing agricultural demands is looking toward the areas of arable land. The use of peat forest for agricultural activities has led to widespread declines in organic carbon (C) and hence in peat quality. The declines occur because, in such activities, the loss of organic-C is not offset by the gains of C through the deposition of biomass. The lowering of groundwater level by constructing drainage ditches in peatland areas was needed to convert the land to agriculture, but its impact on increasing greenhouse gas fluxes and soil decomposition resulted soil compaction was feared. Lowering of groundwater level by the drainage ditches in the peat lands contributed to greatly increased CO<sub>2</sub> fluxes, although CH<sub>4</sub> fluxes slightly decreased. However, the highest N<sub>2</sub>O emissions were measured at intermediate water depths. In this study, soil amelioration in peat soil stimulate GHG emissions from the peat soil. Long-term experiments should be developed to monitor changes that occur over time in response to amelioration at various water depths. It might

be better to apply ameliorants at higher rates to reach a sustainable reduction in GHG emissions but it should consider the applicability to be used by the farmer. These experiments should be conducted in peat soil and peat soil-crop systems in peat soil to determine how GHG emissions from these types of treatments can be reduced. Tropical forest peat land is the best land-use management in the peat lands to suppress carbon loss and greenhouse gas emission. Further long-term investigation is essential to do when lowering of groundwater level is needed to convert the land to agriculture to fulfil food demand.

## **5.2 General conclusion**

From the above study, it could be concluded that:

1. This study results clearly delineated that appropriate management of irrigation or soil amelioration is likely a feasible approach if the approaches could eliminate the trade-off between CH<sub>4</sub> and N<sub>2</sub>O production in rice soils and this condition makes more challenge on reducing the production of one gas but not to increase the production of the other.
2. Approaches that may contribute to reduce GHG emission from agricultural sector through the management of cultivated in mineral soil and peat soil should increase the yield then, it could be adopted, have a significant advantage for farmers. Further studies to verify the mitigation options should focus on feasibility for local farmers.



**Figure 5.1.** The overall conclusion from application of water managements and soil ameliorations in mineral soil and peat soil

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## ABBREVIATION

|  |                                |                                 |   |
|--|--------------------------------|---------------------------------|---|
| ABA  | : abscisic acid                | ECD                             | : electron capture detector                     |
| AEZs   | : agro-ecological zones        | Eh                              | : redox potential                               |
| Al <sub>2</sub> O <sub>3</sub>               | : aluminum oxide               | F                               | : fertilization                                 |
| ANOVA  | : analysis of variance         | FDF                             | : flooded-drained-flooded                       |
| AWD  | : alternate wetting and drying | Fe, Fe(II), Fe(III)             | : iron  |
| B  | : boron                        | Fe <sub>2</sub> O <sub>3</sub>  | : ferric oxide                                  |
| BD   | : bulk density                 | FID                             | : flame ionization detector                     |
| C  | : carbon                       | GC                              | : gas chromatography                            |
| Ca   | : calcium                      | Gg                              | : Giga-gram                                     |
| CF   | : continuous flooded           | GHG                             | : greenhouse gas                                |
| CH <sub>3</sub> CO <sub>2</sub> <sup>-</sup> | : acetate                      | GHGI                            | : greenhouse gas intensity                      |
| CH <sub>4</sub>                              | : methane                      | GWP                             | : global warming potentials                     |
| Cl   | : chlorine                     | H <sup>+</sup> , H <sub>2</sub> | : hydrogen                                      |
| CO <sub>2</sub> eq                           | : carbon dioxide equivalent    | H <sub>2</sub> O                | : water   |
| CO <sub>2</sub>                              | : carbon dioxide               | H <sub>2</sub> S                | : hydrogen sulphide                             |
| Cu   | : copper                       | I                               | : irrigation                                    |
| D  | : drainage                     | IAEA                            | : International Atomic<br>Energy Agency         |
| DAA  | : days after amelioration      | IAERI                           | : Indonesian Agricultural<br>Research Institute |
| DAT  | : days after transplanting     | ICALRRD                         | : Indonesian Center for<br>Agricultural Land    |
| DOI  | : days of incubation           |                                 |   |
| DS   | : dry season                   |                                 |   |
| EC   | : electrical conductivity      |                                 |   |

Resources Research and  
Development

Inventory of Indonesia

IPCC : Intergovernmental Panel  
on Climate Change

IRRI : International Rice  
Research Institute

K : potassium

K<sub>2</sub>HPO<sub>4</sub> : potassium hydrogen  
phosphate

K<sub>2</sub>O : potassium oxide

KCl : potassium chloride

kPa : kilopascal

LULUCF : land use and land use  
change and forestry

Mg : magnesium

Mn, MnO : manganese

MnO<sub>4</sub> : permanganate

Mo : molybdenum

MRP : Mega Rice Project

mV : millivolt

N, N<sub>2</sub> : nitrogen

N<sub>2</sub>O : nitrous oxide

NGHGI : National Greenhouse Gases

NH<sub>2</sub>OH : hydroxylamine

NH<sub>4</sub><sup>+</sup> : ammonium

NO<sub>2</sub><sup>-</sup> : nitrite

NO<sub>3</sub><sup>-</sup> : nitrate

O<sub>2</sub> : oxygen,

P : phosphorous

P<sub>2</sub>O<sub>5</sub> : super phosphate

Pg : Peta-gram

PVC : polyvinyl chloride

RS : rainy season

S : sulphur

SAS : statistical analysis system

S-AWD : site specific AWD

Si : silicon

SiO<sub>2</sub> : silicon dioxide

SNC : second national  
communication

SO<sub>4</sub><sup>2-</sup> : sulphate

SOM : soil organic matter

TCD : thermal conductivity  
detector

Tg : Tera gram

Tukey's HSD test: Tukey's honest significant difference test

US EPA : United State Environmental Protection Agency

USDA : United State Department of Agriculture

WPI : water productivity index

Zn : zinc