



**Chemistry of stratosphere and mesosphere
revealed by ISS/JEM/SMILES for Earth Diagnosis.**

Makoto SUZUKI¹, Naohiro Manago¹
(JAXA/ISAS, Chiba U./CeRES)

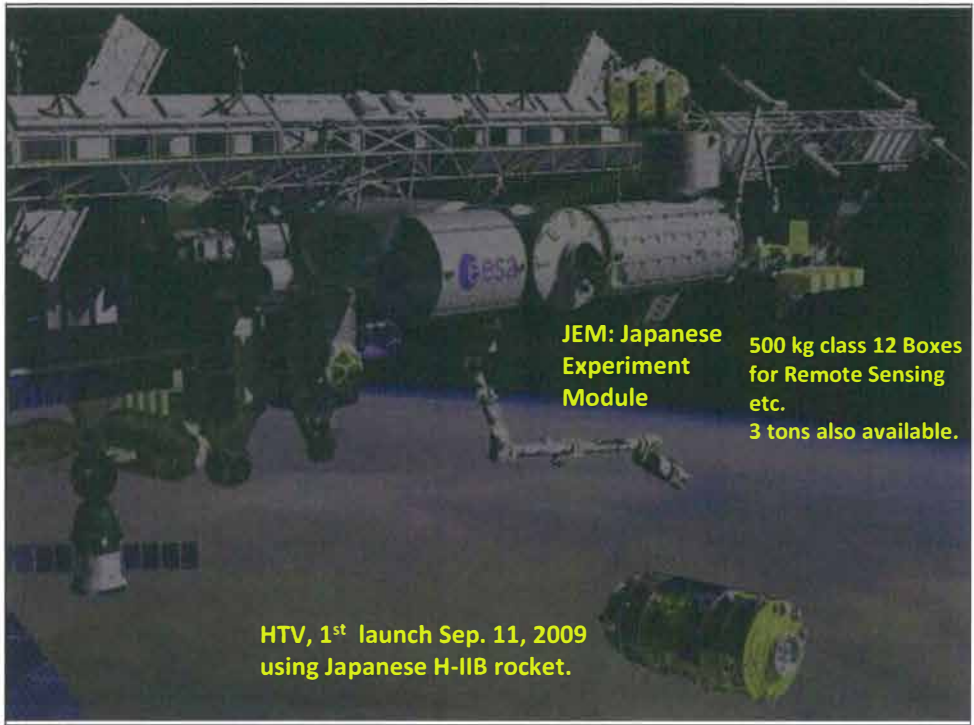
Naohiro Manago¹, Chihiro MITSUDA², Koji IMAI³, Hideharu AKIYOSHI⁴, Takuki SANO⁵, Yoko NAITO⁵,
D. Kinnison⁶, and Masato SHIOTANI⁷

¹Institute of Space and Astronautical Science, JAXA
²Fujitsu FIP, ³Tome R&D Inc.
⁴National Institute for Environmental Studies
⁵Department of Geophysics, Kyoto University, ⁶NCAR
⁷Research Institute for Sustainable Humanosphere, Kyoto University

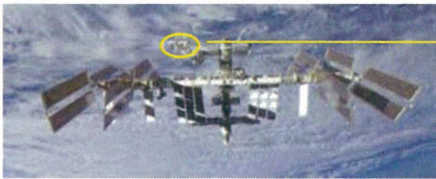


**Passive microwave remote sensing in
Japan**

- **AMSR, AMSR-E have shown excellent performance for remote sensing of SST, sea ice, etc.**
- **AQUA/AMSR-E had stopped operation last year, but AMSR-2 instrument and GCOM-W satellite is now waiting for launch (it was delayed from Feb. 2012, by the delay of another satellite provider).**
- **A sub-mm chemistry mission, SMILES had been conducted jointly by JAXA and NICT. It proposed around 1987 originally, and it finally onboard ISS on Sep. 18, 2009. It worked only 6 months, but it demonstrated breakthrough performance at 625, 650 GHz region, Tsys = 340K, using 4 K cooled SIS detector.**
 - **This is the my 3rd instruments that worked only 6 months.**
 - ADEOS-I/ILAS, ADEOS-II/ILAS-II, ISS/JEM/SMILES (this talk)
 - And I suffered another satellite failure, not entering Venus orbit, on Dec. 9, 2010, but it failed. Venus Climate Orbiter, Akatsuki, is now wandering around 0.6-0.7 AU to reach Venus again around 2015/16.



JEM/SMILES Payload



The SMILES was carried by the H-II/B with the H-II Transfer Vehicle (HTV) (Sep. 11); the HTV was attached to the ISS (Sep. 18); the SMILES was attached to the JEM (Sep. 25) (All dates in JST)

JEM/SMILES Mission

(JEM/SMILES: Superconducting Submillimeter-Wave Limb-Emission Sounder designed to be aboard the Japanese Experiment Module on ISS; Collaboration project of JAXA - Japan Aerospace Exploration Agency - and NICT - National Institute of Information and Communications Technology -)

1. Demonstration of superconductive mixer and 4-K mechanical cooler for the submillimeter limb-emission sounding in space.



[Mechanical Cooler] Two-stage Stirling and J-T;
20mW @4K, 200mW @20K, 1000mW @100K;
Power Consumption: <300 W; Mass: 90 kg



[SIS Mixer]
RF: 640 GHz, IF: 11-13 GHz; Junction: Nb/AlOx/Nb, ~7 kA/cm²;
Fabricated at Nobeyama RO

2. Observation on atmospheric minor constituents in the middle atmosphere

[Standard Products]

- 1 scan: O₃, HCl, ClO, CH₃CN, O₃ isotopes, HOCl, HNO₃
- Multi-scan: HO₂, BrO

[Research Products] UTH, Cirrus Clouds, volcanic SO₂, H₂O₂

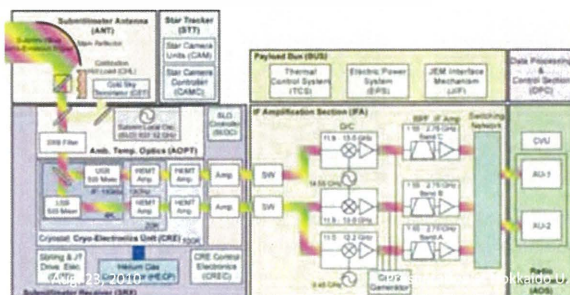
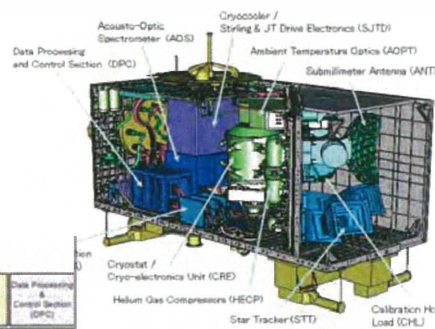
Aug. 23, 2010

Presentation at Hokkaido U.

5

SMILES Instrument

625, 650 GHz region, SSB system
4 K cooled, T_{sys} = 340 K
Two 1.2 GHz bandwidth spectrometers
0.8 MHz spectral resolution.
Comparable or even better than best laboratory instruments.



Scientific targets of SMILES

1. Inorganic Chlorine chemistry
 - ClO to HCl ratio
(O₃ trend in the US)
 - HOCl production
(O₃ trend in the LS)
 - Global ClO
(background ClO)
2. Bromine budget (very short-lived source gas issue)
3. HO_x budget
- etc.

Simulated SMILES observation performance

Error estimation for the mid-latitude case based on the single scan measurement

Sep. 27-29, 2010

AURA meeting, M. Suzuki et al

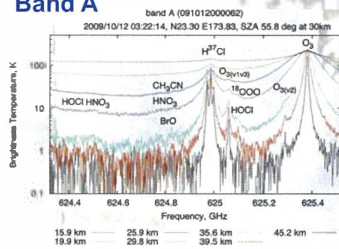
7

Overview of JEM/SMILES Instruments: SIS Device

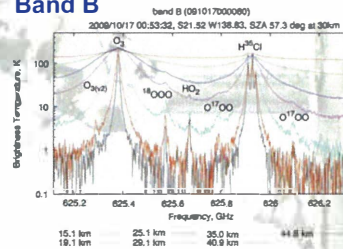
SIS Junction: Nb/AlO_x/Nb
Junction Area: ~1 x 1 mm²
Current Density: 6-7 kA/cm²
Fabricated at Nobeyama Radio Observatory, JAPAN

Two Bands among Band A, B, C can be observed.

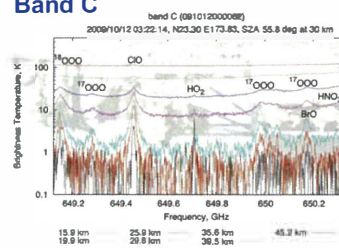
Band A



Band B



Band C



Frequency region has been selected by engineering interest, as high as possible, but 625-626 GHz region is the only frequency to measure HCl below 1 THz.

At 600 GHz troposphere is opaque in limb.

Tsys ~ 350 K, and Noise floor is ~0.4 K, given by

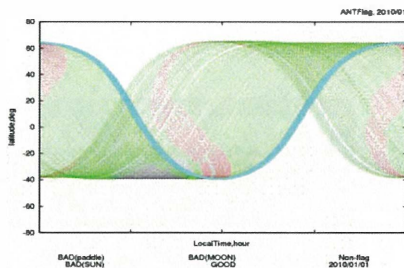
$$NE\Delta T_B = \frac{T_{sys} + T_{Atm}}{\sqrt{\Delta\nu \cdot \tau}}$$

Sep. 27-29, 2010

AURA meeting, M. Suzuki et al

ISS Orbit plane rotates in ~90 days, 45 days for diurnal coverage. It will be good platform for diurnal variation study and solar occultation, such as ISS/SAGE-III (2015 ?).

Actual Local Time Coverage of SMILES in January 2010. Blue: January 1st. Red: not observed.



- ISS orbit
- Low altitude: 350 km
- Moderate inclination: 51.2°



Sep. 27-29, 2010

AURA meeting, M. Suzuki et al

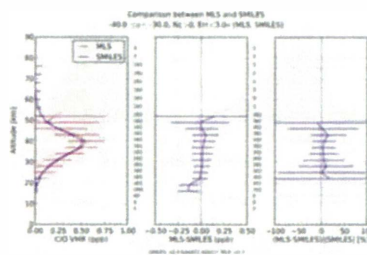
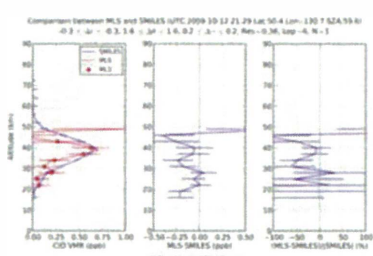
10

Comparison of SMILES CIO v2.0 with MLS 3.3

MLS coincidence, DSZA < 2°, SMILES ver.2 release candidate.

Coincidence event on Oct. 12, 2009
SMILES, **MLS** at 50.4°N 130.7°E (the
 first coincidence in current criteria).
 SZA= 59.6°

Coincidence Statistics 30S-40S, **161 SMILES**
 vs. **301 MLS profiles**, most coincidence
 cases in current criteria. < 10% agreement
 with MLS 3.3 between 22-48 km



2011 8.10

AOGS 2011 Suzuki et al

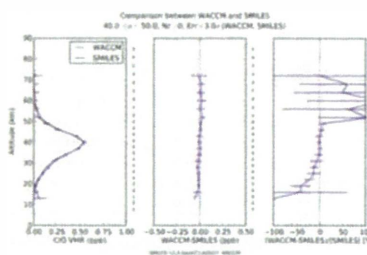
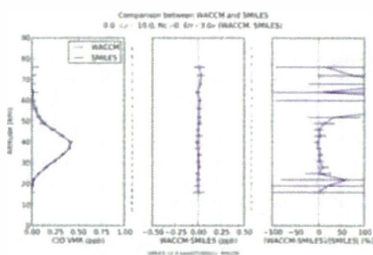
11

Daytime CIO comparison with NCAR WACCM

Other latitude slightly worse than these examples.

coincidences 0-10N, Oct. - Feb.,
SMILES vs. WACCM

coincidences 30-40N, Oct. - Feb.,
SMILES vs. WACCM

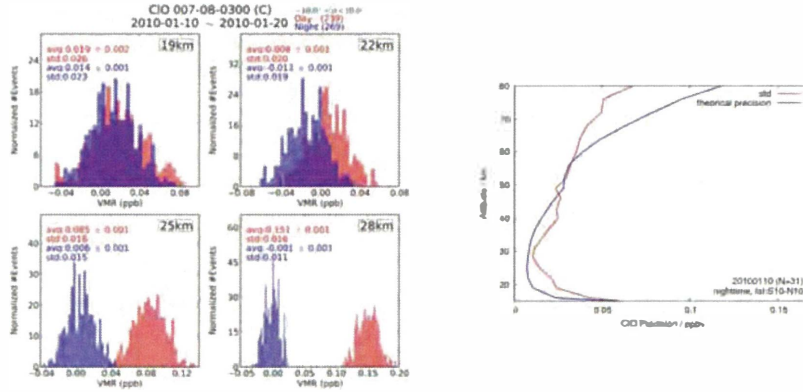


Detection limit of SMILES ClO, 15 ppt at 25 km in single shot, negative bias at 22 km in nighttime

Detection limit changes with pressure.

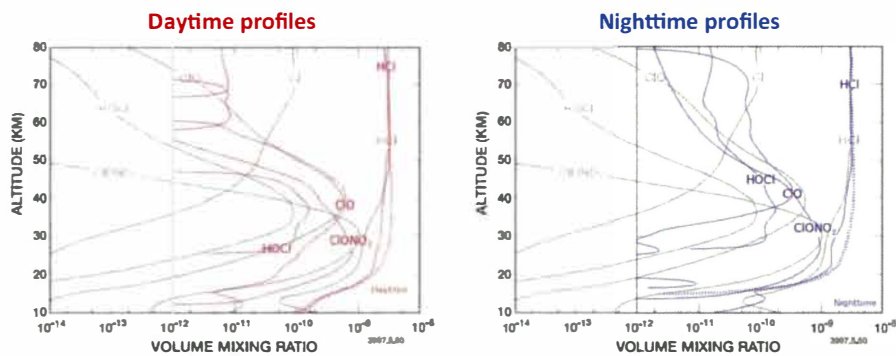
Daytime (noon) and Nighttime (midnight) difference at 10N-10S

Theoretical detection limit and standard variation of nighttime. Below 35 km, it should be detection limit.



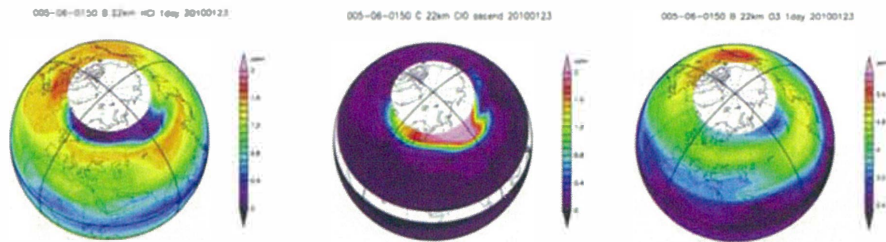
SMILES (+ MIPAS) can provide knowledge of chlorine partitioning at the background atmosphere based upon observations.

2010/10/12, local solar noon (53N-60N) and midnight (23S-33S),
ClONO₂: MIPAS IMK, day 51N-57N, night 50N-54N



Brasseur and Solomon, pp.373

Fig. 1 HCl (left), ClO (center), and O₃ (right) distribution on Jan. 23, 2010 at the 22 km altitude in the northern hemisphere.



HCl is about 1.6 ppbt at outside polar vortex and it is almost entirely converted to the ClO (1.6 to 2.0 ppbt). O₃ destruction has occurred as much as 20% (from 4 ppmv to 3.2 ppmv) after 3 weeks of heterogeneous chemical process.

Fig. 2 (a) shows trajectory of observation points of SMILES (large circles) from 15:23UT to 15:47 in Jan. 23, 2009, and CALIPSO observation points which passed north of Europe. Fig. 2(b) shows SMILES ClO vertical section.

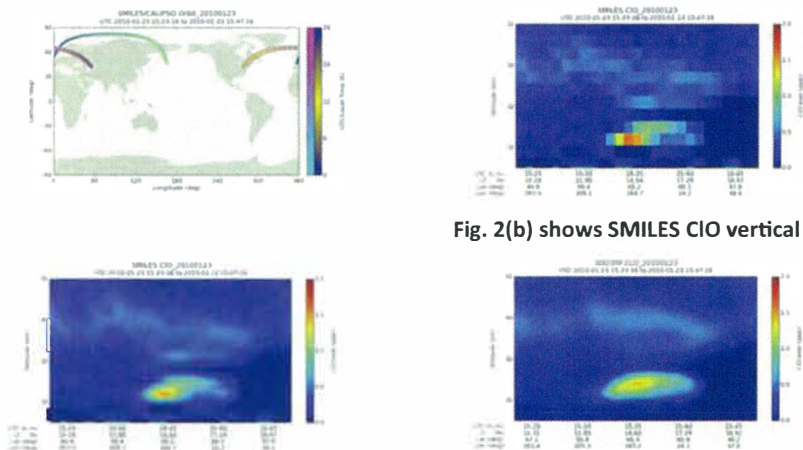
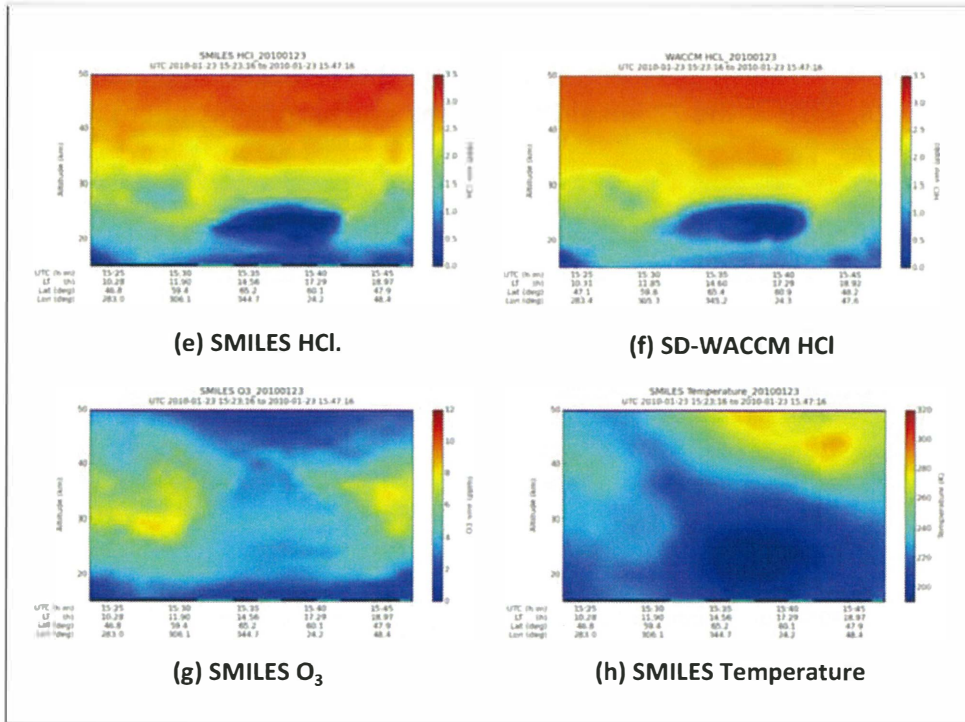


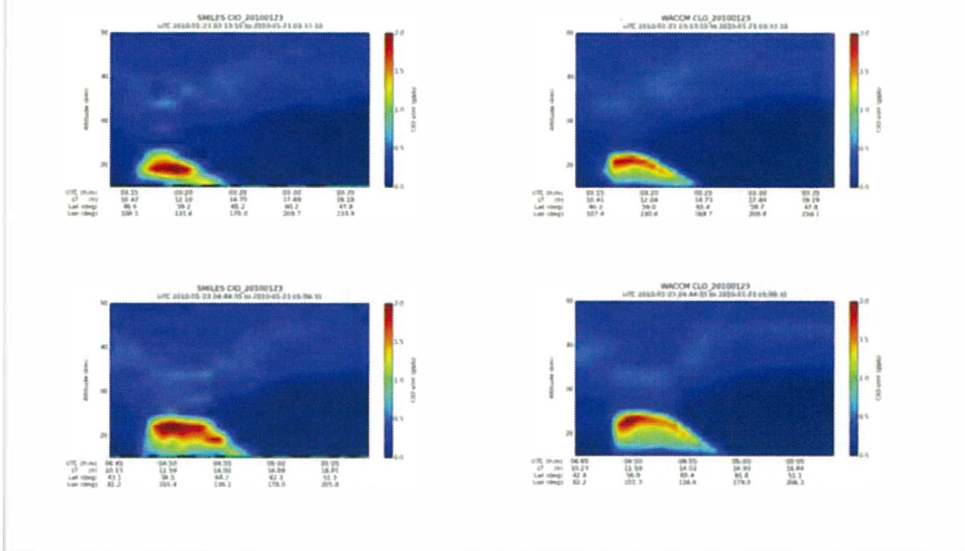
Fig. 2(b) shows SMILES ClO vertical section.

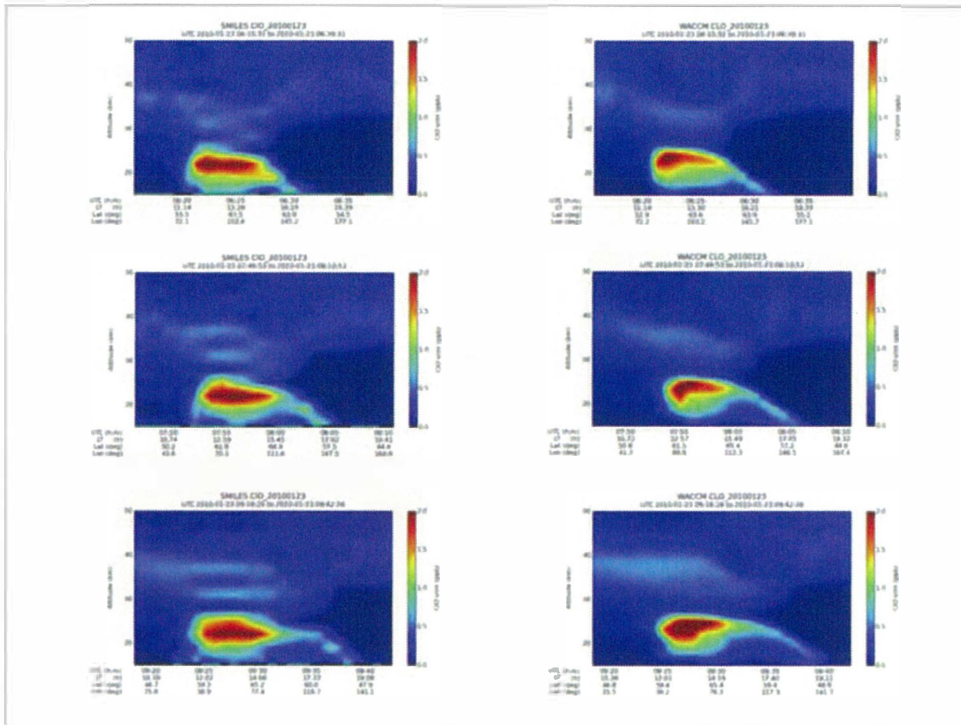
(c) SMILES ClO vertical section interpolated.

(d) SD-WACCM ClO vertical section.

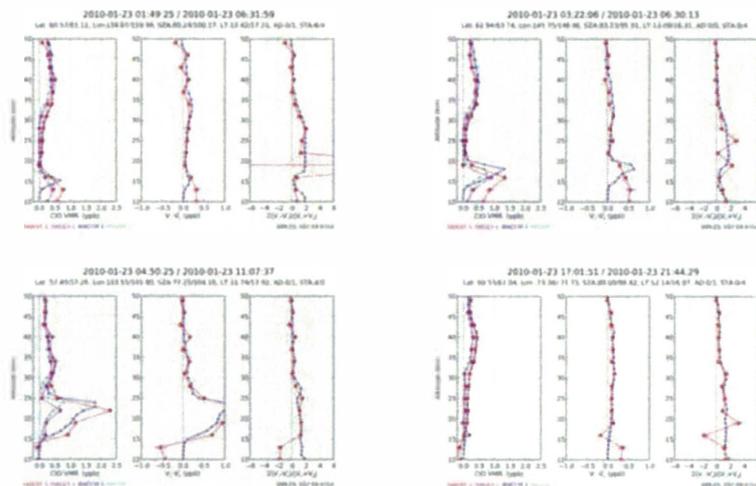


Checking ClO more thoroughly, Jan 23 at different orbits.
 (left) SMILES interpolated, (right) WCCAM interpolated.

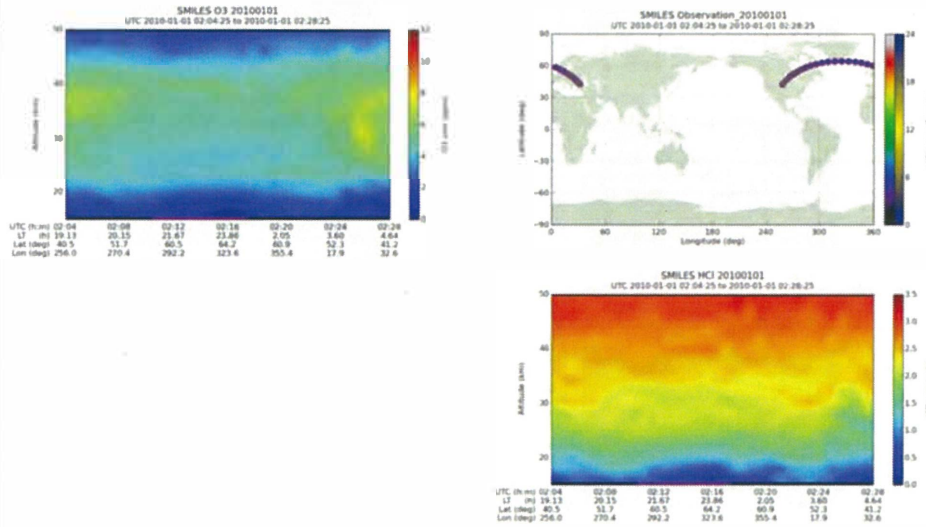




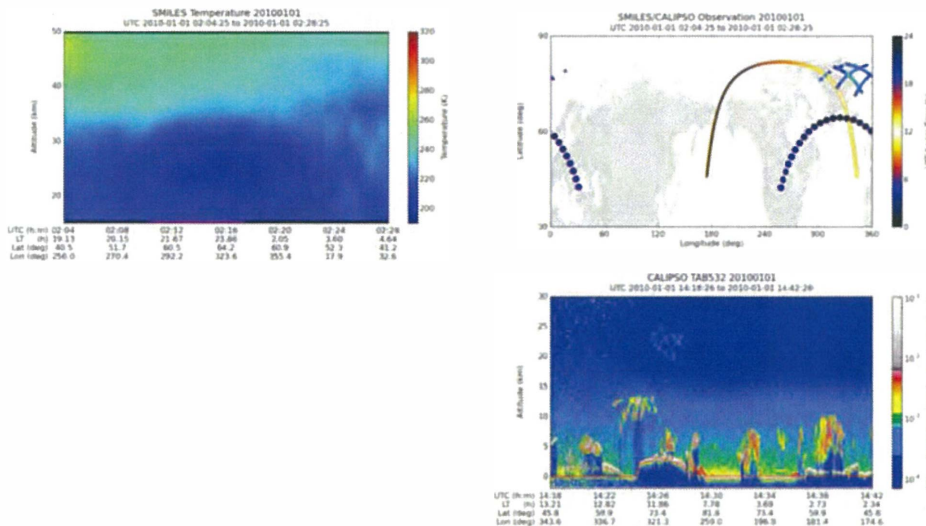
SMILES can measure same location at different local times. Jan. 23rd measurements showed, nighttime CLO decline generally agreed well with WACCM calculation. This suggest current knowledge of (CLO)₂ formation rate is acceptable.

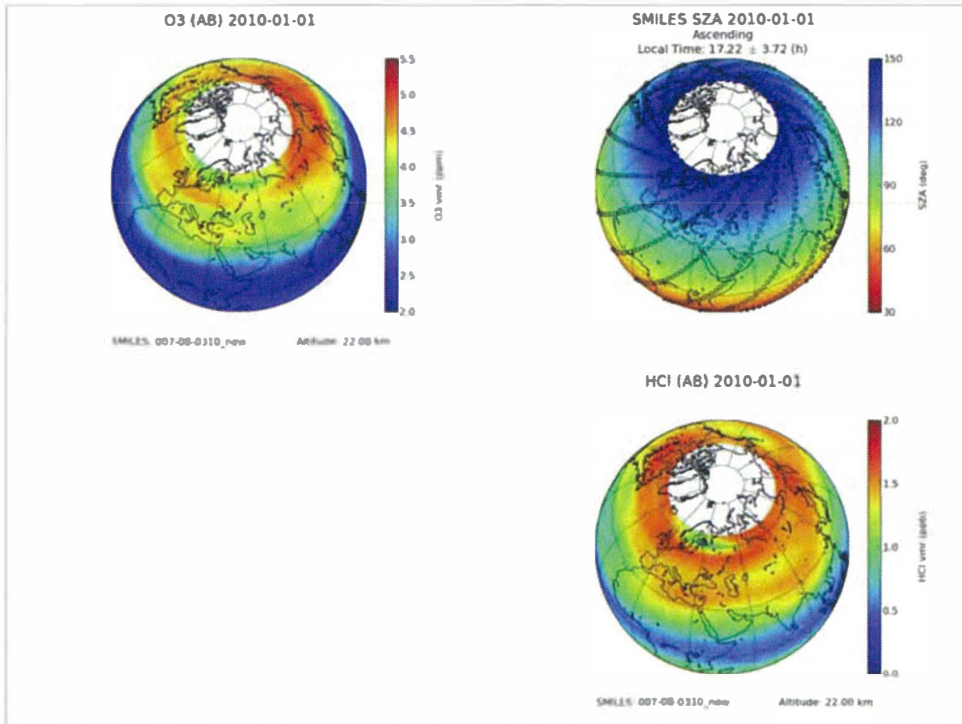


SMILES O₃(Left-Upper), Locations (Right-Upper) ClO (Left-Lower), HCl (Right-Lower)

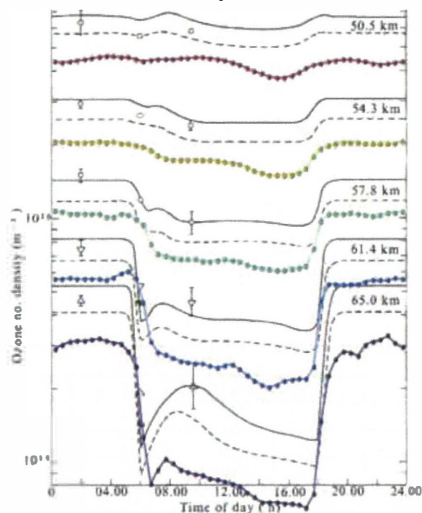


SMILES O₃(LeftUpper), ClO(LD), Locations(RU). CALIPSO (RD)





Diurnal variation of O₃ in the mesosphere observed by SMILES



- mesospheric O₃ shows strong diurnal variation, but it was not observed clearly from ground and space.
- Daytime O₃ and HO₂ etc have relations to H₂O, which is now under evaluation. (Clancy et al, JGR 1994)

$$O_3 \propto (k_5 + k_{19})^{1/3} (k_7 \cdot J_3 a)^{-1} (k_9 \cdot k_{10})^{-1/2} [H_2O]^{-1/2} \quad (1)$$

$$HO_2 \propto (k_5 + k_{19})^{-1/2} (k_7 \cdot J_3 a \cdot k_9)^{1/2} (k_{10})^{-1/2} [H_2O]^{1/2} \quad (2)$$

SMILES and a model calculation (Vaughan, *Nature*, 1984)

Kovalenko et al (2007) have questioned the reaction rate of $\text{ClO} + \text{HO}_2 \rightarrow \text{HOCl} + \text{O}_2$ (JPL206/2011), and it proposed to be near the value reported by Stimpfle (1979). von Clarmann et al (2009) also supported this claiming from MIPAS-B balloon measurements. This could affect Total Ozone Loss at the lower stratosphere significantly. We checked the reaction rate by using steady state relations around [HOCl] at the 30-45 km.

Zonal mean temperature = 249K.

k_1 (nominal)

$$k_1 = 2.7 \times 10^{-12} \cdot \exp(+220/T)$$

$$6.53 \times 10^{-12} \text{ cm}^3/\text{molecule/s (JPL2006)}$$

k_1 (upper limit)

$$9.66 \times 10^{-12} \text{ cm}^3/\text{molecule/s}$$

k_1 (lower limit)

$$4.41 \times 10^{-12} \text{ cm}^3/\text{molecule/s}$$

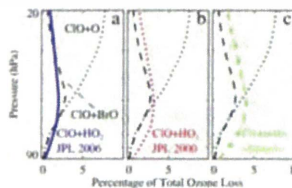


Figure 4. Calculated contribution of the three most important catalytic cycles involving ClO to 24-hour averaged ozone loss, for the 20 September 2005 flight, for model runs (a) k_1^{JPL2006} , (b) k_1^{Stimpfle} , and (c) $k_1^{\text{Kovalenko}}$.

Kovalenko et al (2007) [Balloon and Model] showed better agreement with k_1 by Stimpfle et al (1979). von Clarmann et al (2009) also reported same results using ENVISAT/MIPAS.

$$k_1 = 3.3 \times 10^{-11} \exp(-850/T) + 4.5 \times 10^{-12} (T/300)^{3.7}$$

$$3.34479 \times 10^{-12} \text{ cm}^3/\text{molecule/s (Stimpfle)}$$

$\text{ClO} + \text{HO}_2 \rightarrow \text{HOCl} + \text{O}_2 \quad (k_1)$

At the middle atmosphere 40-50 km, chlorine species are distributed among HCl, HOCl, and ClO. The chemical balance among these species is controlled by following chemical reactions;



The daytime chemical equilibrium [ClO]/[HOCl] can be simplified at 40-50km, where [OH] is small, as following,

$$\frac{[\text{ClO}]}{[\text{HOCl}]} = \frac{j_3}{k_1[\text{HO}_2]} \quad (5)$$

- SMILES ver. 2.0a, [ClO], [HO2], [HOCl]
- GEOS-5 meteorological data
- j_3 : HOCl photolysis is calculated, from cross-section JPL2006, multiple scattering calculation using MODTRANS (DISORT, 16 streams).

SMILES観測(ver. 2.0a)に基づく k_1 の計算は30, 35, 40 kmにおいてJPL2006を支持している。

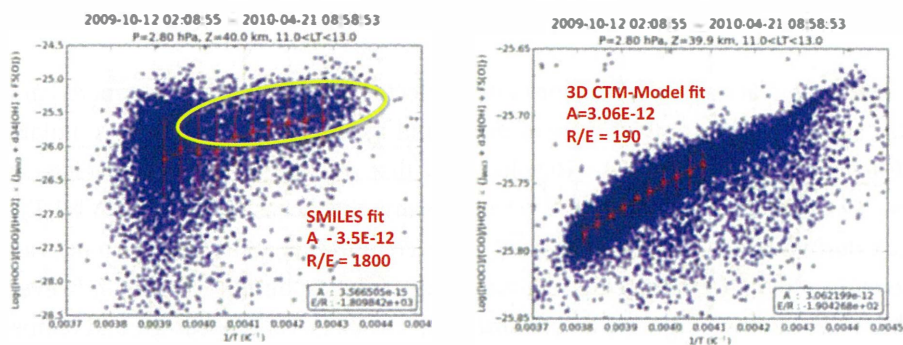
Oct. 12, 2009, 57.1±1.8°N, SZA=64.8±1.6°

H (km)	T (K)	k_1 (cm ³ /molecule/s)			
		SMILES	WACCM	JPL2006	Stimpfle
30	223.0	5.91E-12	6.87E-12	7.24E-12	2.23E-12
35	228.3	9.75E-12	6.10E-12	7.08E-12	2.44E-12
40	239.8	7.25E-12	5.38E-12	6.76E-12	2.92E-12
45	249.2	3.72E-12	3.70E-12	6.53E-12	3.36E-12

Stratospheric ClO, HO₂, and HOCl measurements by SMILES can estimate $k(d_{34})$: $\text{ClO} + \text{HO}_2 \rightarrow \text{HOCl} + \text{O}_2$

JPL2011 value: $A=2.7\text{E-}12$, $R/E = 220$

$$d_{34} = [\text{HOCl}]/[\text{ClO}]/[\text{HO}_2] \times \{ \text{JHOCl} + d_{34}[\text{OH}] + f_{35}[\text{O}] \}$$



SMILES data fit (left), was not successful but agree with JPL2011 at 250 K.

If we look at only the yellow circle region $T < 250$ K ($1/T > 0.004$), it may agree with JPL2011.

Summary

- SMILES observed chemistry of 2009-10 Arctic winter with higher sensitivity (~ 0.015 ppb precision for ClO, better than Aura/MLS ~ 0.1 ppb), it should make description of chemistry much easy and clear.
- Spatial and temporal features agreed quite well between SMILES and SD-WACCM, in general.
- $(\text{ClO})_2$ formation was tentatively checked through ClO decay during nighttime by comparison with SD-WACCM results, and it looks current knowledge of chemical kinetics is acceptable.
- Chlorine partitioning inside polar vortex can be studied with ClONO_2 and other data (from other sensor or model calculations).
- SMILES L2 data (currently ver. 2.1) will be kept updating (to ver. 2.2, 2.3, ...), and ver. 2.1 will be open to general public soon.